

Influences of subcritical and supercritical CO₂ treatment on the pore structure characteristics of marine and terrestrial shales

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ABSTRACT

To better understand and implement the CO₂ sequestration project, it is of significance to investigate the interaction of shales with CO₂ and its potential effects on the pore morphology. In this study, two marine shale samples and two terrestrial shale samples were prepared and treated with subcritical CO₂ (30 °C and 5 MPa) and supercritical CO₂ (80 °C and 20 MPa) in a geochemical reactor. Various methods, including low-pressure carbon dioxide adsorption (LP-CO₂A), low-pressure nitrogen adsorption (LP-N₂A), high-pressure mercury intrusion porosimetry (HP-MIP) and fractal theory, were used to gain insights into the changes in the shale pore structure after 14 days of CO₂ saturation. According to the results, the phase states of CO₂ obviously affected the variations of pore structure parameters during the physical and chemical reactions in shales. Interactions of supercritical CO₂ with shales created a more obvious effect on the pore structure compared to those of subcritical CO₂, which was attributed to the greater dissolution and expansion effect as well as the extraction mechanism associated with supercritical CO₂. After exposing the shale samples to subcritical CO₂, the pore size distributions (PSDs) of the treated shale samples were lower than those of the raw samples at all diameter scales, indicating that the number of pores decreased due to the reactions. Furthermore, it was found that after supercritical CO₂ treatment, the micropore and mesopore structure parameters of the marine shale samples obviously decreased with an increase in macropore structure parameters, leading to the reduction in fractal dimensions in smaller pores, while the terrestrial shale samples appeared to represent a contrary trend. These findings will provide experimental evidence for further assessment of the mechanisms for CO₂ geological sequestration with enhanced shale gas recovery.

1. Introduction

With the growing demand for clean energy, shale gas, as a typical unconventional gas resource, has attracted increasing attention worldwide. It is found that the utilization of shale gas can effectively reduce the concern for air quality problems and pressure on global climate changes, since it releases a small amount of greenhouse gas and almost no sulfur dioxide (SO₂) when producing the same amount of electricity power compared to other fossil fuels [1]. However, as one of the key technologies for shale gas development, hydraulic fracturing consumes huge amount of water resource, resulting in a “competition for water” problem which may influence the development of local agriculture and industry. Recently, supercritical CO₂ (ScCO₂), as a non-aqueous fracturing and drilling fluid, exhibits several advantages over water, including promoting the desorption of methane, enhancing fracture propagation and reducing flow-blocking mechanisms [2], which can be

applied to the shale gas development, especially for areas lacking water. Therefore, the shale gas reservoirs have been considered as a promising geological locations for the CO₂ sequestration, which can enhance shale gas recovery while also achieving CO₂ storage (CS-ESGR) [3].

In contrast to conventional reservoirs, organic-rich shales are typically impermeable sedimentary rock characterized by complex petrophysical and geochemical systems. Based on the pore classification reported by the International Union of Pure and Applied Chemistry (IUPAC), the pores can be defined as micropores (pore diameter smaller than 2 nm), mesopores (pore diameter in the range of 2–50 nm) and macropores (pore diameter larger than 50 nm) [4]. In general, the heterogeneous pores are associated with organic matter and mineral compositions, and vary widely in size, types and morphology. Thus, it is still a challenging task to accurately characterize the multi-scale nanopore system in shale reservoir. In recent years, various techniques, such as field emission scanning electron microscopy (FE-SEM) imaging

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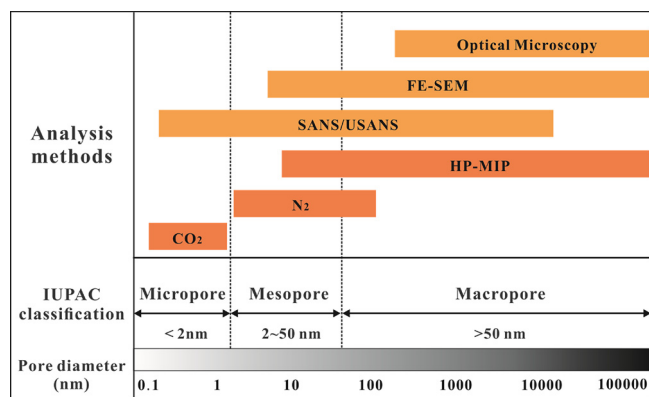
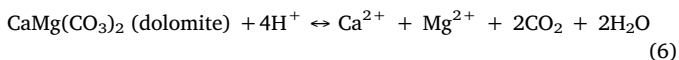
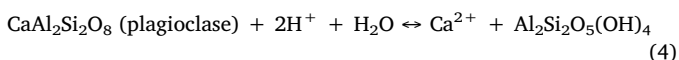
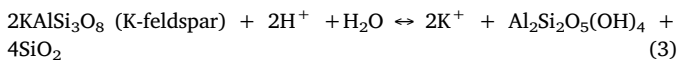
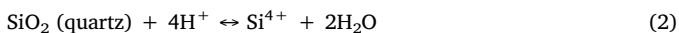
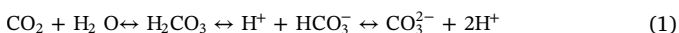


Fig. 1. Comparison of the different methods used to investigate pore size in shales.

[5], high-pressure mercury intrusion porosimetry (HP-MIP) [6], low-pressure gases (CO₂ and N₂) adsorption (LP-GA) [7], and small-angle and ultra-small-angle neutron scattering (SANS/USANS) [8], are applied to acquire qualitative and quantitative information of pore morphology in shale reservoirs as illustrated in Fig. 1. Among these techniques, the combination of LP-GA and HP-MIP is considered to be a simple and effective method to investigate the pore structure in shale.

Shale reservoir has the capacity to store an extensive amount of CO₂ due to its complex compositions with huge surface area. With the injection of CO₂ into shale formations, complicated geochemical reactions occur between them, resulting in uncertainties for the CS-ESGR projects. Some researchers have performed numerous CO₂-brine-rock chemical reaction experiments to explore the mechanisms of geological CO₂ storage. These researchers reported that the dissolution of CO₂ in formation water created the carbonic acid, leading to the dissolution and precipitation of the minerals as listed in Eqs. (1)–(6) [9–12]. Generally, the kinetics rate of these geochemical reactions were related to the temperature and pressure. When the temperatures and pressures are above the critical point of CO₂ (T_C of 31.08 °C and P_C of 7.38 MPa), the CO₂ will be compressed to a supercritical state (ScCO₂) [13], which has distinct characteristics in terms of density, viscosity and diffusion ability compared to subcritical CO₂ (SubCO₂). According to the findings of Yin et al. [14] and Perera et al. [15], ScCO₂ saturation caused a more significant modification to the mechanical properties of shales and coals compared to SubCO₂ saturation due to higher dissolution capacity and greater chemical potential of ScCO₂. Zhou et al. [16] found that the CO₂ adsorption isotherms exhibited an enhanced propensity when the CO₂ phase varied from gaseous to supercritical state. Moreover, it was also suggested that the amount of swelling caused by CO₂ adsorption clearly depended on the phase state of CO₂ [17]. ScCO₂ adsorption-induced expansion was approximately two times larger than that caused by SubCO₂ adsorption.



Recently, in-situ field implementations have been conducted to evaluate the feasibility of the CS-ESGR project. In 2016, the Virginia

Center for Coal and Energy Research (VCCER) conducted a successful ‘huff-and-puff’ test. Approximately 510 tons of carbon dioxide were successfully injected into the Chattanooga shale formation, which is located in north-central Tennessee in the United States. The results showed that there was an obvious increase in gas flow rate as well as the gas quality after a long-term CO₂ soaking phase, confirming the injectivity and sequestration potential of the CO₂ in shale gas reservoir [18]. Meanwhile, in 2017, the Shanxi Yanchang Petroleum Company (China) performed non-aqueous fracturing and transformation tests on four shale gas wells in the terrestrial Yanchang formation, using CO₂ as the fracturing fluids. They found that the average daily production of the wells reached its highest level after the injection of CO₂, and the retention rate of carbon dioxide in the shale formation was approximately 60%–70%, indicating that a significant amount of CO₂ could be sequestered [19]. These successful practices will be summarized and then applied in both terrestrial and marine shale reservoirs in China in the future. However, some fundamental mechanisms behind the CS-ESGR projects remain unclear. Since the sorption and transport behaviors of the fluids are controlled by the heterogeneous porous medium, knowledge of the variations in multi-scale pore structure after CO₂ treatment is critical for implementation of the CS-ESGR projects.

In recent years, numerous studies have investigated the influences of CO₂ exposure on the pore morphology of coals under various geological conditions. [20–24]. They suggested that these variations in pore structure depended on both coal ranks and phase states of the injected CO₂. However, to the best of our knowledge, the studies of the effect of SubCO₂ and ScCO₂ treatment on pore structure modifications of shale formations, especially comprehensive studies on the multi-scale pore structure, are relatively rare. Wu et al. [25] investigated the impacts of CO₂ exposure on the permeability of Utica shale samples and found that the CO₂ adsorption would induce volumetric expansion, leading to the reduction in porosity and permeability. Lahann et al. [26] combined the LP-CO₂A and LP-N₂A methods to determine the influence of CO₂ on micro- and mesopore structure parameters of the Indiana shale samples. The results showed that there was a slight increase in surface area as a result of temperature and CO₂. Yin et al. [27] documented that the ScCO₂ treatment had a significant influence on the micropores of the Longmaxi shales. After treatment, the number of micropores and the specific surface area showed a significant decrease. According to Pan et al. [28], the Longmaxi and Chang-7 shale samples exhibited different variation patterns in the micro- and mesopore structure parameters after long-term ScCO₂ saturation. Jiang et al. [29] also studied the influences of ScCO₂ exposure on the microstructure of Longmaxi shales by the HP-MIP technique. They reported that the porosity and surface area of the shale samples increased with the increasing of immersion pressure and time.

It can be seen from the above literature reviews that the characterization of variations in shale pore structure during CO₂ exposure is complex and has not been adequately studied. Furthermore, due to the pressure decline during fracturing engineering, the ScCO₂ may transform into SubCO₂ at the beginning of the injectivity [14]. Considering the distinct characteristics between SubCO₂ and ScCO₂, it is also essential to comprehensively evaluate how well CO₂ can influence the microstructure of shales under different phase states. This paper, therefore, focused on a comparison of the potential variations in nanopore structure system of shales with the exposure of CO₂ at sub- and supercritical state, aiming to provide new perspective to better implement the CS-ESGR project. Furthermore, two different types of sedimentary shales, including marine and terrestrial shales, were selected as samples to provide basic knowledge for the evaluation of the CO₂ storage in shale formations with different sedimentary backgrounds. Several methods, such as low-pressure carbon dioxide adsorption (LP-CO₂A), low-pressure nitrogen adsorption (LP-N₂A) and high-pressure mercury intrusion porosimetry (HP-MIP), were used to characterize the changes in micro-, meso- and macropore structure system of shales before and after CO₂ treatment.

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