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been made by means of circular dichroism spectroscopy.

Synthesis, crystal structures and spectral characterization of chiral 4-R-1,2,4-triazoles

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ABSTRACT

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1. Introduction

Derivatives of 1,2,4-triazole have become a matter of many synthetic developments in regards to their medical and material applications. Biological activity of these heterocyclic systems is brightly represented by the antifungal medications, such as fluconazole, itraconazole, voriconazole, posaconazole etc. [1–3]. Some triazoles were confirmed to have a promising antitubercular activity [4]. Certain pesticides, weedicides, fungicides based upon 1,2,4-triazole fragment were proposed. In material chemistry, a special attention towards 4-substituted 1,2,4-triazoles is related to their coordination chemistry. 1D iron(II) polymeric complexes with 4R-1,2,4-triazoles are known for their spin crossover behaviour [5], which is one of the most spectacular phenomenon of molecular bistability [6].

Elaboration of homochiral pharmaceutically important molecules is frequently a crucial step towards their medical applications. On the other hand, the synthesis of chiral ligands constitutes an important challenge to the synthetic chemistry in regards of their

* Corresponding author. E-mail address: illia.guralskyi@univ.kiev.ua (I.A. Gural'skiy). use for the stereoselective catalysis [7] and for the construction of

advanced materials [8]. To the best of our knowledge, no chiral 4-

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substituted 1,2,4-triazoles have been reported to date.

1.2.4-triazoles attract attention as actively used medications and ligands for constructing coordination

architectures. In this paper we describe four optically active 4-substituted 1,2,4-triazoles that have been

prepared by Bayer's synthesis from the corresponding aliphatic chiral amines. This approach tends to be

universal towards different triazoles and permits to conserve a homochirality of substrates. Novel

asymmetric molecules have been characterized by spectroscopic techniques and their structures have

been retrieved from the single crystal X-ray analysis. Chiro-optical studies of these heterocycles have

2. Experimental

2.1. Synthesis

Freshly prepared monoformylhydrazide was refluxed for 3.5 h with a 20% excess of triethyl orthoformate in water-free methanol. After cooling to 40 °C, the n-primary amine (1 equiv.) was added and the mixture refluxed for 6–7 h. In order to get rid of excess of the amine, the resulting mixture was heated under vacuum at 220 °C. The solid triazoles **1–3** were recrystallized from i-PrOH.

(2S)-2-(4H-1,2,4-triazol-4-yl)propan-1-ol (1). Yield: 2.02 g (53.5%); white solid; mp 98 °C. ¹H NMR (400 MHz, DMSO-*d*₆): $\delta = 1.45$ (d, 3H), 3.58 (m, 2H), 4.37 (m, 1H), 5.02 (s, 1H), 8.38 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): $\delta = 142.12$, 64.75, 53.88, 17.14. Anal. Calcd for C₅H₉N₃O: C, 47.23; H, 7.13; N, 33.05. Found: C, 47.32; H, 7.02; N, 33.03. *m*/*z* = 128.1 [M+H]⁺. IR (KBr): 3164, 3100, 2854, 1527, 1462, 1372, 1198, 985, 650 cm⁻¹.

(2S)-3-methyl-2-(4H-1,2,4-triazol-4-yl)butan-1-ol (2). Yield: 2.52 g (54.4%); white solid; mp 106 °C.







Table 1	
Crystal data and structure refinement for 1-4.	

Identification code	1	2	3	4
Empirical formula	C ₅ H ₉ N ₃ O	C ₇ H ₁₃ N ₃ O	C ₈ H ₁₅ N ₃	C ₁₀ H ₁₇ N ₃
Formula weight	127.15	155.20	153.23	179.26
Crystal system	Orthorhombic	Orthorhombic	Orthorhombic	Triclinic
Space group	$P2_{1}2_{1}2_{1}$	P2 ₁ 2 ₁ 2 ₁	$P2_12_12_1$	P1
a/Å	4.9153 (2)	6.6618 (6)	7.8453 (4)	11.423 (3)
b/Å	6.8128 (4)	7.5779 (6)	10.5984 (6)	12.5029 (17)
c/Å	19.1627(10)	16.6766 (16)	11.2101 (5)	13.6295 (17)
α/°	90.00	90.00	90.00	114.301 (13)
β/°	90.00	90.00	90.00	101.892 (17)
γ/°	90.00	90.00	90.00	100.953 (16)
Volume/Å ³	641.70 (6)	841.88 (13)	932.09 (8)	1651.6 (6)
Z	4	4	4	6
$\rho_{calc}g/cm^3$	1.316	1.225	1.092	1.081
μ/mm^{-1}	0.096	0.085	0.069	0.067
F (000)	272.0	336.0	336.0	588.0
Crystal size/mm ³	0.5 imes 0.5 imes 0.3	$0.6\times0.3\times0.3$	$0.4\times0.3\times0.2$	$0.2 \times 0.05 \times 0.05$
2Θ range for data collection/ $^{\circ}$	6.34 to 55.12	6.58 to 55	7.28 to 60	5.922 to 51.998
Reflections collected	2596	4366	6210	17793
Independent reflections	$R_{int} = 0.0131$, $R_{sigma} = 0.0260$	$R_{int} = 0.0227$, $R_{sigma} = 0.0519$	$R_{int} = 0.0267$, $R_{sigma} = 0.0449$	$R_{int} = 0.0803$, $R_{sigma} = 0.2861$
Data/restraints/parameters	1478/0/118	2639/0/103	2713/0/104	11925/288/806
Goodness-of-fit on F ²	1.070	1.040	0.948	0.747
Final R indexes $[I \ge 2\sigma (I)]$	$R_1 = 0.0370$, $wR_2 = 0.0813$	$R_1 = 0.0493$, $wR_2 = 0.0878$	$R_1 = 0.0514$, $wR_2 = 0.0958$	$R_1 = 0.0649$, $wR_2 = 0.0666$
Final R indexes [all data]	$R_1 = 0.0499$, $wR_2 = 0.0897$	$R_1 = 0.0939$, $wR_2 = 0.1086$	$R_1 = 0.1115$, $wR_2 = 0.1105$	$R_1 = 0.3527$, $wR_2 = 0.1139$
Largest diff. peak/hole/e Å ⁻³	0.10/-0.16	0.12/-0.17	0.10/-0.11	0.14/-0.11



Scheme 1. Synthetic scheme for the preparative preparation of chiral 4-substituted 1,2,4-triazoles by Bayer's synthesis.

¹H NMR (400 MHz, DMSO-*d*₆): δ = 0.71 (d, 3H), 0.99 (d, 3H), 2.14 (m, 1H), 3.67 (1H), 3.77 (1H), 3.89 (1H), 4.93 (s, 1H), 8.33 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ = 141.16, 64.34, 61.16, 29.14, 18.83. Anal. Calcd for C₇H₁₃N₃O: C, 54.17; H, 8.44; N, 27.08. Found: C, 53.56; H, 8.61; N, 26.39. *m*/*z* = 156.2 [M+H]⁺. IR (KBr): 3196, 3087, 2971, 1527, 1198, 1024, 662 cm⁻¹.

4-[(2S)-3,3-dimethylbutan-2-yl]-4H-1,2,4-triazole (3). Yield: 2.80 g (61.7%); white solid; mp 135 °C. ¹H NMR (400 MHz, DMSO*d*₆): δ = 0.89 (s, 9H), 1.46 (d, 3H), 4.17 (m, 1H), 8.34 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ = 142.75, 60,53, 34,36, 26.08, 15.48. Anal. Calcd for C₈H₁₅N₃: C, 62.71; H, 9.87; N, 27.42. Found: C, 62.85; H, 9.73; N, 27.31. *m/z* = 154.2 [M+H]⁺. IR (KBr): 3100, 2971, 1701, 1514, 1455, 1368, 1179, 1062, 656 cm⁻¹.

4-[(1S)-1-cyclohexylethyl]-4H-1,2,4-triazole (4). Yield: 4.02 g (75.4%); yellowish oil; bp 220 °C (1 atm.). ¹H NMR (400 MHz, DMSO-*d*₆): δ = 1.1 (m, 11H), 1.45 (d, 3H), 4.13 (m, 1H), 8.35 (s, 2H). ¹³C NMR (400 MHz, DMSO-*d*₆): δ = 141.09, 57.39, 43.52, 28.93,



Scheme 2. Homochiral substituents in 1,2,4-triazolic ring of compounds 1-4.

28.52, 25.14, 17.32. Anal. Calcd for $C_{10}H_{17}N_3$: C, 67.00; H, 9.56; N, 23.44. Found: C, 66.88; H, 9.46; N, 23.31. $m/z = 180.1 \text{ [M+H]}^+$. IR (KBr): 3106, 2932, 2848, 1669, 1520, 1449, 1179, 663 cm⁻¹.

2.2. X-ray diffraction

Crystals of **1–3** suitable for the X-ray diffraction analysis were obtained by recrystallization from i-PrOH. Crystals of **4** were obtained by slow freezing of oil to $0 \degree C$.

X-Ray diffraction studies of **1**–**4** were performed on an "Xcalibur 3" diffractometer (graphite-monochromated MoK α radiation ($\lambda = 0.71073$), CCD detector, ω -scans) at room temperature (295–303 K). Structures were solved by direct method and refined against F^2 within anisotropic approximation for all non-hydrogen atoms using OLEX2 program package [9] with SHELXS and SHELXL modules [10]. In **2**–**4**, all H atoms were placed in idealized positions and constrained to ride on their parent atoms with U_{iso} = nU_{eq} (n = 1.5 for OH and CH₃ groups and n = 1.2 for all other H atoms). OH and CH₃ groups were refined as rotating groups except structure **4**. In **1**, all H atoms were located from difference electronic map and refined isotropically.

In **4**, there are six molecules per independent part of the unit cell, and two of them carry structurally disordered cyclohexane rings (relative refined occupancy factors 0.5/0.5 for "D" molecule and 0.6/0.4 for "E" molecule). Similar bonds in disordered groups were restrained to be of similar lengths within an effective standard deviation of 0.01 Å. Also, the high thermal disordering of all six molecules requires rather strict restraints, namely: (a) all C–C bonds in cyclohexane rings were restrained to have fixed values of 1.54 Å to within 0.01 Å; (b) all N–N bonds were restrained to have approximately same values to within 0.01 Å; (c) anisotropic thermal parameters of all C atoms in cyclohexane rings were restrained

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