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# Phase evolution and dielectric properties of fluorite-type $Bi_3(Nb_{0.9}M_{0.1})O_{7+\delta}$ ceramics (M=Ti, Zr, Sn, W, $\delta=\pm0.05$ )



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#### ABSTRACT

Phase structure and dielectric behavior of  $Bi_3(Nb_{0.9}M_{0.1})O_{7+\delta}$  ceramics (M = Ti, Zr, Sn, W,  $\delta = \pm 0.05$ ) were discussed. Cubic fluorite solid solutions  $Bi_3(Nb_{0.9}M_{0.1})O_{7+\delta}$  (M = Zr, Sn, W,  $\delta = \pm 0.05$ ) were obtained through a conventional solid state reaction method. An analysis of the microwave dielectric properties showed that substitution of  $Nb^{5+}$  by  $W^{6+}$  was favorable to improving the temperature coefficient of resonant frequency (TCF value). A temperature stable dielectric material  $Bi_3(Nb_{0.9}W_{0.1})O_{7.05}$  was introduced. The  $Bi_3(Nb_{0.9}W_{0.1})O_{7.05}$  ceramics sintered at 930 °C for 2 h exhibited a high permittivity of 92.4, a Qf value of 260 GHz and a TCF value of -34 ppm/°C.

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#### 1. Introduction

Modern communication systems have moved to the microwave (MW) frequency region, where advanced dielectric ceramics are frequently used in resonators and filters. Miniaturization requires high relative permittivity  $(\varepsilon_r)$  materials (since the size of resonators is inversely proportional to  $\varepsilon_r^{1/2}$ ) with a near zero temperature coefficient of resonance frequency (TCF value). Furthermore, ceramics with low dielectric loss (high Qf value) are needed for the high selectivity and optimized bandwidth of the filters [1]. More recently, an important breakthrough in the telecommunications came with the introduction of low-temperature cofired ceramic (LTCC) technology, which has enabled miniaturization, the integration of passive functions and a reduction in costs, and has led to the production, for example, of the well-known Bluetooth module. LTCC modules are produced by co-firing ceramic layers with a threedimensional Ag-microstrip circuitry. To avoid melting of the Ag-microstrips, the firing temperature must be lower than 950 °C, which is extremely low for a ceramic material and represents the major problem with this technology [2]. Microwave dielectric ceramics with high permittivity ( $\varepsilon_r > 80$ ) has been widely studied [3–8], while most of them need high sintering temperatures. In the Bi<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> system, BiNbO<sub>4</sub> and Bi<sub>3</sub>NbO<sub>7</sub> possess low sintering temperatures and good microwave dielectric properties, and have attracted much attention in recent years [9,10]. BiNbO<sub>4</sub> ceramics have good properties for microwave application and LTCC, but the reaction between BiNbO4 and the silver electrode limited its further application [11,12]. Valant et al. [12–14] studied the crystalline structure and microwave dielectric properties of Bi<sub>3</sub>NbO<sub>7</sub> ceramics, and they found that the cubic incommensurate phase Bi<sub>3</sub>NbO<sub>7</sub> exhibited a high permittivity of 100, a lower Qf value of 300 GHz and a high negative TCF value of 200 ppm/°C. The cubic incommensurate phase Bi<sub>3</sub>NbO<sub>7</sub> underwent an order-disorder phase transition into a commensurate tetragonal phase at temperatures bellow 900 °C, and the tetragonal phase exhibited a lower permittivity of 91, a higher Qf value of 730 GHz and a positive TCF value of +100 ppm/°C. Meanwhile, both the cubic incommensurate phase and the commensurate tetragonal phase Bi<sub>3</sub>NbO<sub>7</sub> ceramics did not react with Ag and they would be suitable for LTCC modules and microwave applications if their TCF value could be modified to near zero.

The solid solutions of  $Bi_3(Nb_{1-x}Ta_x)O_7$  [15,16] were sensitive to the sintering conditions and the phase structure and microwave dielectric properties can be tailored by changing sintering conditions. The TCF value of  $Bi_3(Nb_{0.4}Ta_{0.6})O_7$  with tetragonal phase was near to zero but the permittivity decreased greatly to 72 [15]. Our previous work [17] showed that the cubic incommensurate phase  $Bi_3NbO_7$  ceramics transmitted into commensurate tetragonal phase



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by means of substitution of Nb<sup>5+</sup> by V<sup>5+</sup>, and the TCF value was tailored to near zero while the dielectric permittivity decreased from 91 to 80. In the present study, we examine substitution of Nb<sup>5+</sup> by tetravalent cations such as Ti<sup>4+</sup>, Zr<sup>4+</sup> and Sn<sup>4+</sup>, and/or hexavalent cations such as W<sup>6+</sup> in Bi<sub>3</sub>NbO<sub>7</sub>. The sintering behavior, phase structure, microstructure, and dielectric properties of Bi<sub>3</sub>(Nb<sub>0.9</sub>M<sub>0.1</sub>)O<sub>7+ $\delta$ </sub> (M = Ti, Zr, Sn, W,  $\delta = \pm 0.05$ ) ceramics were studied.

#### 2. Experimental procedure

Proportionate amounts of reagent-grade Bi<sub>2</sub>O<sub>3</sub> (99%, Shu-Du Powders Co. Ltd, China), Nb<sub>2</sub>O<sub>5</sub> (>99%, Guo-Yao Co, Ltd, Shanghai, China), TiO<sub>2</sub> (>99%, Linghua Co. Ltd, Zhaoqing, China), ZrO<sub>2</sub> (>99.5%, Xinxing Co.Ltd, Yixing, China), SnO<sub>2</sub> (>99.5%, Xinxing Co.Ltd, Yixing, China), and WO<sub>3</sub> (99%, Tianjin, China) were weighed according to the stoichiometric formulation of Bi<sub>3</sub>(Nb<sub>0.9</sub>M<sub>0.1</sub>)O<sub>7+ $\delta$ </sub> (M = Ti, Zr, Sn, W,  $\delta = \pm 0.05$ ). Powders were mixed and milled for 4 h using a planetary mill (Nanjing Machine Factory, Nanjing, China) with a running speed of 300 r/min. The mixed oxides were then dried and calcined at 750 °C for 4 h. After being crushed and re-milled for 5 h using the ZrO<sub>2</sub> milling media and deionized water, powders were pressed into cylinders (10 mm in diameter and 5 mm in height) in a steel die with 5 wt.% PVA binder addition under a uniaxial pressure of 20 MPa. Samples were sintered in the temperature range from 850 to 950 °C for 2 h.

The crystalline structures of samples were investigated using Xray diffraction with Cu K $\alpha$  radiation (Rigaku D/MAX-2400 X-ray diffractometer, Tokyo, Japan). Microstructures of sintered ceramic were observed on the as-fired surface with scanning electron microscopy (SEM) (JSM-6460, JEOL, Tokyo, Japan). The dielectric properties of ceramic as a function of frequency (1 K–10 MHz) and temperature (25–300 °C) were measured by an LCR analyzer (4284, Agilent, Santa Clara, CA) with a self-made high temperature system. Dielectric behaviors at microwave frequency were measured with the TE<sub>01 $\delta$ </sub> shielded cavity method with a network analyzer (8720 ES, Agilent, Palo Alto, CA) and a temperature chamber (Delta 9023, Delta Design, Poway, CA). Temperature coefficient of resonant frequency (TCF value) and temperature coefficient of capacity (TCC value) were calculated with the following formulas:

$$TCF = \frac{f_{85} - f_{25}}{f_{25} \times (85 - 25)},\tag{1}$$

where  $f_{85}$  and  $f_{25}$  are the TE<sub>01 $\delta$ </sub> resonant frequencies at 85 and 25 °C, respectively;

$$TCC = \frac{\varepsilon_{85} - \varepsilon_{25}}{\varepsilon_{25} \times (85 - 25)},\tag{2}$$

where  $\epsilon_{85}$  and  $\epsilon_{25}$  are the permittivity at 100 kHz at 85 and 25 °C, respectively.

#### 3. Results and discussions

According to the literature [18,19], there are three types of superstructures in the Nb-doped  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> solid solution system. The structure of cubic Bi<sub>3</sub>NbO<sub>7</sub> is the so-called type II superstructure, which is an incommensurate superstructure extends from 6.25 to 26.0 mol% of Nb<sub>2</sub>O<sub>5</sub> in the Bi<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> system. Fig. 1 presents the XRD patterns of Bi<sub>3</sub>(Nb<sub>0.9</sub>M<sub>0.1</sub>)O<sub>7+ $\delta$ </sub> (M = Ti, Zr, Sn, W,  $\delta = \pm 0.05$ ) ceramics sintered at 930 °C for 2 h. As shown in Fig. 1 (a-c), all peaks can be indexed on a cubic cell with space group *Fm*-3*m*, except the superlattice ordering diffraction peak at  $2\theta = 33.7$ , which was consistent with the XRD pattern of parent Bi<sub>3</sub>NbO<sub>7</sub> in the



Fig. 1. XRD patterns of  $Bi_3(Nb_{0.9}M_{0.1})O_{7+\delta}$  ceramics  $(M=Ti, Zr, Sn, W, \delta=\pm0.05)$  sintered at 930 °C for 2 h: (a)  $Bi_3(Nb_{0.9}Ti_{0.1})O_{6.95}$ , (b)  $Bi_3(Nb_{0.9}Zr_{0.1})O_{6.95}$ , (c)  $Bi_3(Nb_{0.9}Sn_{0.1})O_{6.95}$ , (d)  $Bi_3(Nb_{0.9}W_{0.1})O_{7.05}$ .

literature [14,16,17]. Single phase with incommensurate cubic fluorite-type structure was obtained in the  $Bi_3(Nb_{0.9}Zr_{0.1})O_{6.95}$ ,  $Bi_3(Nb_{0.9}Sn_{0.1})O_{6.95}$ , and  $Bi_3(Nb_{0.9}W_{0.1})O_{7.05}$  ceramics, while in the  $Bi_3(Nb_{0.9}Ti_{0.1})O_{6.95}$  ceramics, a small amount of  $Bi_2Ti_2O_7$  phase appeared (as shown in Fig. 1 (d)).

The ionic radii of  $Ti^{4+}$ ,  $Zr^{4+}$ ,  $Sn^{4+}$  and  $W^{5+}$  were 0.605, 0.72, 0.69 and 0.60 Å, respectively, and the ionic radius of  $\mathrm{Nb}^{5+}$  was 0.64 Å, as suggested by Shannon [20]. There was no clear evidence suggesting that the solid solubility of  $Bi_3NbO_7$  substituted by M (M = Ti, Zr, Sn and W) is related with ionic radius. As described above, the cubic Bi<sub>3</sub>NbO<sub>7</sub> is a kind of Nb-doped  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> solid solution with defect fluorite  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> as the basic structure. According to the phase diagrams [21], the structure of  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> can be stabilized at room temperature by doping Bi<sub>2</sub>O<sub>3</sub> with a large number of metallic cations, such as trivalent rare earth, pentavalent  $V^{5+}$ ,  $Nb^{5+}$  and  $Ta^{5+}$ , or hexavalent  $W^{6+}$  and  $Mo^{6+}$ , except tetravalent  $Zr^{4+}$ ,  $Ti^{4+}$  and  $Sn^{4+}$ . I. Abrahams' work [22] showed that W<sup>6+</sup> has a very favorable solution energy for Nb<sup>5+</sup> substitution in Bi<sub>3</sub>NbO<sub>7</sub> through computer simulation techniques and  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> solid solutions can also be obtained in the  $Bi_3Nb_{1-x}W_xO_{7+x/2}$  system when  $x \le 0.2$ . Same result was obtained in the  $Bi_3Nb_{1-x}Zr_xO_{7-x/2}$  system in the composition range 0.0 < x < 0.40 [23], which means that the solution energy of  $Zr^{4+}$  can be lowered by combining with Nb<sup>5+</sup> substitution. In this work, δ-Bi<sub>2</sub>O<sub>3</sub> solid solutions with incommensurate cubic fluoritetype structure were also obtained in the Bi<sub>3</sub>(Nb<sub>0.9</sub>Sn<sub>0.1</sub>)O<sub>6.95</sub> ceramics, as well as in the  $Bi_3(Nb_{0.9}Zr_{0.1})O_{6.95}$  and  $Bi_3(Nb_{0.9}W_{0.1})O_{7.05}$ . It means that the solution energy of Sn<sup>4+</sup> can also be lowered by combining with Nb<sup>5+</sup> substitution. In the Bi<sub>3</sub>(Nb<sub>0.9</sub>Ti<sub>0.1</sub>)O<sub>6.95</sub> ceramics, the appearance of the secondary phase of Bi<sub>2</sub>Ti<sub>2</sub>O<sub>7</sub> with face-centered cubic structure (space group Fd-3m) suggests that the solution energy of  $Ti^{4+}$  is still too high to form a  $\delta$ -Bi<sub>2</sub>O<sub>3</sub> solid solution.

SEM images of as-fired surface of Bi<sub>3</sub>(Nb<sub>0.9</sub>M<sub>0.1</sub>)O<sub>7+ $\delta$ </sub> ceramics (M = Ti, Zr, Sn, W,  $\delta = \pm 0.05$ ) in Fig. 2 shows that all the ceramics can be well densified at sintering temperatures no less than 870 °C. The grain shapes were similar with each other for Zr, Sn, or W substituted Bi<sub>3</sub>NbO<sub>7</sub> ceramics because of the same crystalline phase. The grain size lay in 1–4 and 0.5–3 um for the Bi<sub>3</sub>(Nb<sub>0.9</sub>M<sub>0.1</sub>) O<sub>6.95</sub> ceramics (M = Zr, Sn) and the Bi<sub>3</sub>(Nb<sub>0.9</sub>W<sub>0.1</sub>)O<sub>7.05</sub> ceramics, respectively. In the Bi<sub>3</sub>(Nb<sub>0.9</sub>Ti<sub>0.1</sub>)O<sub>6.95</sub> ceramic, two kinds of grains were observed. Combining with the XRD results in Fig. 1, it can be declared that the principal part of the Bi<sub>3</sub>(Nb<sub>0.9</sub>Ti<sub>0.1</sub>)O<sub>6.95</sub> ceramic was fluorite-type Bi<sub>3</sub>NbO<sub>7</sub> phase which presented as bigger grains, and the smaller grains embedded in the larger ones were corresponding to Bi<sub>2</sub>Ti<sub>2</sub>O<sub>7</sub> phase.

Fig. 3 (a) shows the dielectric frequency spectrum at room

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