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Biodiesel production from sewage sludge by using alkali catalyst catalyze

Xuemin Wu^a, Fenfen Zhu^{a*}, Juanjuan Qi^a, Luyao Zhao^a

^aSchool of Environment & Natural Resources, Renmin University of China, Beijing, China, 100872

Abstract

Sewage treatment produces sewage sludge, which is in large amount and continue to increase. The sludge contains considerable lipid and they can react with methanol to produce biodiesel. Normally acid catalyst was used in the transesterification reaction with sludge as a raw material, because sewage sludge contains a lot of free fatty acid which is an inhibition factor for transesterification reaction by alkali catalyst. However, alkali catalyst has other advantages such as easy to separate and fast reaction speed. As a result, in this research in-situ transesterification by alkali catalyst with sewage sludge as raw material was discussed. KOH/activated carbon as solid base catalysts were prepared by using wet impregnation method, which were used as catalysts in sewage sludge in-situ transesterification reaction. The influence of the catalyst mass fraction (15%,20%,25%,35%,45%), and dosage of catalyst (0.1g,0.2g,0.3g,0.35g,0.4g,0.5g,0.6g,0.7g), on the in situ-transesterification has been examined. The results show that: 25% (mass fraction, the same below) KOH/activated carbon was used as catalyst in the reaction process. When the mass ratio of alcohol and sludge was 10:1, reaction temperature was 60°C, catalytic amount (3wt%) and reaction time was 8h,the yield of FAME was 6.8%.Those were the optimum conditions. The surface features of catalyst were analysed by BET.

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1. Introduction

Fatty acid alkyl esters, also known as biodiesel, is a renewable fuel that is commonly produced by the reaction of refined pre-extracted vegetable oil and a simple aliphatic alcohol in the presence of a alkaline or acid catalyst^[1,2]. It provides a similar energy density to petrodiesel and can be used in most diesel engines in pure form (B100) or may be blended with petroleum diesel at any concentration^[3-7]. However, because 70–85% of the overall biodiesel production cost is associated with the raw materials, sewage sludge is more competitive when the lipid concentration in it is more than 10% ^[4,8,9]. Sewage sludge is abundant and can get continuous yield. The higher value of the food products, more competitive the sewage sludge as the raw material for biodiesel. Therefore, the worldwide research in

the biodiesel industry has currently focused on how to convert low cost feedstock especially renewable wastes such as sewage sludge into biodiesel products via low cost technology i.e. cheap catalyst, simple processes, economic and environmentally friendly, etc. ^[10-12]. Sewage sludge normally has a large amount of free fatty acid (FFA), which is an obstacle for biodiesel production when alkaline catalyst is used. Because FFA will react with alkaline catalysts as a kind of acid which lead to yield of water and finally lead to soap production, so it will need much more consumption of catalyst and it also lead to lower yield of biodiesel. However, comparing with acid catalyst, alkaline catalyst is easier to be separated from the final product and it reacts much faster. Thus, developing new efficient alkaline catalyst for producing biodiesel by sewage sludge is still of interest.

Yongtao An et al, as to alkaline catalyst, supported solid alkaline catalysts are more widely used and faster for biodiesel production ^[13]. They can be filtered easily from the liquid products and be reused, and it also can be designed to give higher activity, selectivity and longer catalyst lifetime ^[14-16]. Therefore, research of heterogenous catalysts for biodiesel production is now taken more and more seriously. Shaotong Jiang et al. provided result on FAMEs yields from cottonseed oil by transesterification (95.3%) is the highest of all methods (several extraction strategies, different dosage of Na₃PO₄/MgO, and transesterification) tried ^[17].

The main objective of this paper is to determine the optimal parameters of in-situ transesterification of sewage sludge for maximizing the yield of biodiesel by alkaline catalyst, which are the catalyst mass fraction, dosage of catalyst.

2. Materials and Methods

2.1 Chemicals

Methanol (analytically pure), activated carbon (analytically pure), Anhydrous sodium sulfate(analytically pure), Sodium chloride (analytically pure), and n-hexane (analytically pure) were purchased from Xilong Chemical Co., Ltd. Potassium Hydroxid(analytically pure) was purchased from Sinopharm chemical reagent Co. Ltd. All chemicals used in this process were at analytical pure level and used without further purification.

2.2 Sample preparation

Sewage sludge sample was collected from a municipal wastewater treatment plant in Beijing, China. The moisture content of the sludge sample is 83.3%, and the VSS of the sludge sample is 65.3%, and the average lipids content of the sludge sample is 9.4%. Sewage sludge was dried at 105 °C for 24 h and then was milled into fine powder by vibration mill. The sludge samples were stored in sealed plastic vials at 4 °C.

2.3 Catalysts preparation

In this research, alkali catalyst which was KOH/activated carbon prepared by using wet impregnation method was used. KOH/activated carbon preparation process: First, we had prepared KOH aqueous solution the mass fraction of which were 15%, 20%, 25%, 30%,35%,40%,45%. Then respectively got 30 ml KOH aqueous solution mixed with 6g activated carbon, impregnated for 30h, and then dried for 30h at 120 °C.Finally, removed them and set aside.

2.4 Experimental design

Two factors(catalyst mass fraction, dosage of catalyst), and seven kinds catalysts of different mass fractions, and eight levels of catalyst dosage were shown in Table 1, the mass ratio of alcohol and sludge was 10:1, reaction temperature was 60° C, and reaction time was 8h, KOH/activated carbon was respectively taken as catalysts. The results were shown in Figure 1. The methanol to sludge mass ratio was 10:1 and the reaction temperature was 60° C. The reaction time and mechanical stirring speed were maintained at 8 h and 300 rpm, respectively.

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