



An efficient, one-pot, three-component procedure for the synthesis of chiral spirooxindolopyrrolizidines via catalytic highly enantioselective 1,3-dipolar cycloaddition



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ABSTRACT

The catalytic, highly regio-, diastereo-, and enantioselective synthesis of a small library of chiral spirooxindolopyrrolizidines via a three-component 1,3-dipolar cycloaddition reaction of azomethine ylides, derived from isatin, with electron-deficient dipolarophiles, 3-(2-alkenyl)-1,3-oxazolidin-2-ones, is described. A chiral copper(II) complex of cyclohexane-1,2-bis(arylmethyleneamine) catalyzed this process at room temperature. The reaction mechanism is discussed on the basis of the assignment of the absolute configuration of the cycloadducts.

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Catalytic asymmetric multicomponent reactions (CAMCR) are efficient processes in terms of chirality economy and environmental impact. In addition, this strategy is a powerful tool for the rapid introduction and expansion of molecular diversity.¹ It is therefore desirable to utilize and develop this method for the synthesis of important heterocycles such as chiral spirooxindolopyrrolizidines and spirooxindoloproline, for example, horsfiline,² elacomine³ and rhynchophylline, which exhibit significant biological activities⁴ (Fig. 1). Asymmetric multicomponent 1,3-dipolar cycloadditions of azomethine ylides with alkenes represent a useful strategy for stereoselective synthesis and the development of compounds having similar structures.⁵

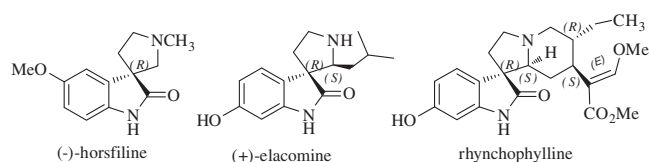


Figure 1. Spiropyrrolizidine oxindole alkaloids.

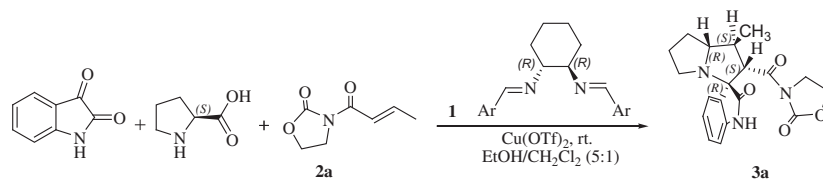
We previously reported the synthesis of enantiomerically pure novel spirooxindolopyrrolizidines by applying an optically active cinnamoyl oxazolidinone as a chiral auxiliary and the obtained diastereoselectivities were very high.⁶ However, this protocol required the use of at least one equivalent of an enantiopure auxiliary which represents a major drawback. To resolve this problem, and in continuation of our previous work on the synthesis of spirooxindoles,⁷ herein, we have utilized a copper complex of cyclohexane-1,2-bis(arylmethyleneamine) **1** as a catalyst to synthesize a small library of this important class of spirooxindoles.⁸ We report a highly regio-, diastereo-, and enantioselective 1,3-dipolar cycloaddition reaction of azomethine ylides, derived from isatin, with an electron-deficient dipolarophile by using bidentate bis(imine)-Cu(II) complex **1**, that can be readily assembled from commercially available *trans* 1,2-cyclohexanediamine and a variety of suitable aldehyde precursors,⁹ under optimized reaction conditions. Based on our previous experience and a literature survey,¹⁰ initially, the effects of substituents on the bis(imine) ligands **1** were examined using 10 mol % of Cu(OTf)₂ as the catalyst in the three-component reaction of isatin, (*S*)-proline, and dipolarophile **2a** at room temperature. The results are summarized in Table 1.

The ligands **1b** and **1c** bearing the electron-withdrawing and relatively bulky Cl-containing substituents at the 2- and/or 6-positions of the benzene ring resulted in considerably higher yields and enantioselectivities in comparison with the other ligands.¹¹ The

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Table 1
Asymmetric synthesis of new chiral spirooxindolopyrrolidine **3a** with chiral ligands **1a–f**



Entry ^a	Ligand	Ar	Temp (°C)	Time (h)	Yield ^b (%)	ee ^c (%)
1	1a		25	24	84	55
2	1b		25	22	93	95
3	1c		25	20	89	63
4	1d		25	29	79	racemic
5	1e		25	29	73	racemic
6	1f		25	32	83	racemic
7	1b		0	35	35	93
8	1b		−40	48	<10	n.d.

^a The reaction of isatin (0.20 mmol), (*S*)-proline (0.21 mmol) with **2a** (0.20 mmol) was carried out in EtOH/CH₂Cl₂ (3 mL, 5:1) at rt in the presence of 10 mol % of the catalyst [Cu(OTf)₂-**1** = 1.0:1.1].

^b Isolated yield.

^c Determined by chiral HPLC analysis.

highest enantioselectivity (95%) and yield (93%) were achieved by employing ligand **1b**. The yields and enantiomeric ratios of the products showed the temperature dependence of this process. A decrease in the reaction temperature from 25 °C to −40 °C significantly decreased the yield and enantioselectivity (entries 2, 7 and 8). We next tested the effect of different Cu salts in this process using **1b** as the ligand (Table 2).

In all cases, Cu(OTf)₂ proved to be the best copper source, while other Cu salts led to a decrease in the ee, and longer reaction times (entries 2 and 3 vs 1). The use of Zn(OTf)₂ instead of Cu(OTf)₂ gave a poorer result in terms of the enantioselectivity (entry 4). The effects of catalyst loading were also investigated and the best results were obtained when 10 mol % of the catalyst was used. The ligand-to-metal ratio of 1.1:1 using 20 mol % of ligand was investigated under similar conditions and the isolated yield and enantioselectivity remained the same at 96% and 90%, respectively. Lowering the catalyst loading to less than 10 mol % led to a decrease in the yield, reactivity and enantioselectivity. It should be noted that the addition of additives such as 3 or 4 Å molecular sieves did not give any noticeable changes in the results of the reaction, and even led to lower yields.

Table 2
Effect of the Lewis acid

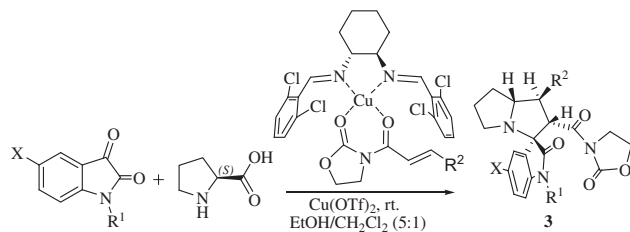
Entry	Lewis acid (10 mol %)	Time (h)	Yield (%)	ee (%)
1	Cu(OTf) ₂	22	93	95
2	Cu(OAc) ₂	23	92	66
3	CuCl ₂	28	76	racemic
4	Zn(OTf) ₂	12	>99	racemic
5	Cu(OTf) ₂ ^a	22	96	90

^a 20 mol % of catalyst was used.

Using the optimized reaction conditions, we next examined the scope and generality of this reaction with various types of azomethine ylides and two derivatives of 3-(2-alkenyl)-1,3-oxazolidin-2-ones (**2**), and synthesized a small library of new chiral spirooxindolopyrrolidines **3a–j** (Table 3).

The structures of the cycloadducts were assigned from their elemental and spectroscopic analyses including IR, ¹H NMR, ¹³C NMR, and mass spectral data.

Table 3
Asymmetric synthesis of new chiral spirooxindolopyrrolidines **3a–j**



Entry	X	R ¹	R ²	Cycloadduct	Yield (%)	ee (%)
1	H	H	Me	3a	93	95
2	H	H	Ph	3b	95	93
3	H	Me	Me	3c	93	89
4	H	Et	Ph	3d	92	87
5	H	Bn	Me	3e	92	91
6	Br	H	Me	3f	99	89
7	Br	Me	Me	3g	92	87
8	Br	Et	Me	3h	94	90
9	Br	Me	Ph	3i	91	89
10	NO ₂	H	Me	3j	88	92

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