



Editor's choice paper

Photo-epoxidation of cyclohexene, cyclooctene and 1-octene with molecular oxygen catalyzed by dichloro dioxo-(4,4'-dicarboxylato-2,2'-bipyridine) molybdenum^(VI) grafted on mesoporous TiO₂



Henry Martínez^{a,*}, María F. Cáceres^b, Fernando Martínez^a, Edgar A. Páez-Mozo^a, Sabine Valange^{c,*}, Nelson J. Castellanos^d, Daniel Molina^e, Joël Barrault^c, Henri Arzoumanian^f

^a Centro de Investigaciones en Catálisis-CICAT, Universidad Industrial de Santander, Escuela de Química, Km 2 vía El Refugio, Piedecuesta, Santander, Colombia

^b Laboratório de Química Atmosférica (LQA), Universidade Católica do Rio de Janeiro, Rua Marquês de São Vicente, 225, Gávea-Rio de Janeiro, RJ, Brazil

^c Institut de Chimie des Milieux et Matériaux de Poitiers (IC2MP), Université de Poitiers, CNRS, ENSIP, B1, 1 rue M. Doré, F-86073 Poitiers, France

^d Estado Sólido y Catálisis Ambiental (ESCA), Universidad Nacional de Colombia, Carrera 30 No. 45-03, Bogotá, Colombia

^e Laboratorio de Resonancia Magnética Nuclear, Universidad Industrial de Santander, Km 2 vía El Refugio, Piedecuesta, Colombia

^f Chirosciences UMR CNRS 7313 Institut des Sciences Moléculaires de Marseille (iSm2), Aix-Marseille Université, Campus St Jérôme, 13397 Marseille, France

ARTICLE INFO

Article history:

Received 12 November 2015

Received in revised form 31 May 2016

Accepted 4 July 2016

Available online 6 July 2016

Keywords:

Dioxomolybdenum complex

Mesoporous TiO₂

Photocatalysis

Selective oxidation

Epoxidation

ABSTRACT

The photo-assisted oxygen transfer to cyclohexene, cyclooctene and 1-octene was investigated using molecular O₂ as a primary oxidant with a dichloro-dioxo-(4,4'-dicarboxylato-2,2'-bipyridine) molybdenum^(VI) complex grafted on the surface of mesoporous TiO₂ supports exhibiting different textural properties. Such mesoporous titania samples were prepared following a hydrothermal treatment (HT) and through an original procedure under supercritical CO₂ (SC) conditions. Non porous TiO₂ Degussa P-25 was also used for comparison purposes. The porosity of the titania supports, as well as the amount and dispersion of the covalently anchored dioxo Mo^(VI) complex were shown to strongly influence the resulting photo-catalytic properties. A particular attention was also devoted to the stability of the grafted catalysts by performing long-term epoxidation experiments. The new hybrid Mo^(VI)-derivatized mesoporous SC TiO₂ catalyst proved to be the most efficient in the photo-assisted oxygen transfer to various alkenes. Additionally, the high stability of such catalyst was ascertained by recycling studies.

© 2016 Elsevier B.V. All rights reserved.

1. Introduction

Traditional oxidation reactions based on stoichiometric oxidants tend to produce waste side products generating a negative environmental impact [1]. In this respect, the use of environmentally friendly oxidants such as molecular oxygen is a more attractive and challenging option in order to generate oxygenates without formation of by-products [2–4].

Epoxidation of alkenes is a suitable route prime of importance for the preparation of relevant industrial intermediates [5]. Epoxides are versatile and reactive compounds for the synthesis of fine

chemicals, pharmaceuticals, agrochemical products, epoxy paints and dyes [6,7]. However, the production of epoxide compounds is still a big issue because several side reactions can take place, such as oxidation in the allylic positions, ring opening of the epoxides or complete oxidation to CO₂ [8]. Furthermore, the epoxidation of terminal olefins such as 1-octene is difficult and therefore requires long time reaction [9]. Traditionally, epoxides are prepared by partial oxidation of alkenes using “exotic” oxidants like iodosylbenzene, hypochlorite or organic hydroperoxides [10]. Hydrogen peroxide or *t*-butyl hydroperoxide (TBHP) are known as efficient oxidants [11] with the assistance of homogeneous transition metal-based catalysts [12–14]. The epoxidation of ethylene and propylene was recently reported with molecular oxygen or air as a primary oxidant with a heterogeneous silver catalyst [15,16]. However, the reaction conditions are rather severe and risky. Thus, the development of catalytic systems that use molecular oxygen or air in mild

* Corresponding authors.

E-mail addresses: henry.martinez1@correo.uis.edu.co (H. Martínez), sabine.valange@univ-poitiers.fr (S. Valange).

conditions remains a challenging target [17]. In previous studies [18–21], we have reported on the selective production of organic oxygenated compounds by Oxygen Atom Transfer (OAT) reactions, using O_2 as the primary oxidant, with a dioxoMo(VI) complex grafted on a commercial TiO_2 (MoO_2L/TiO_2) and UV–vis irradiation.

Here, the photo-assisted oxygen transfer to cyclohexene, cyclooctene and 1-octene was investigated on new grounds with a dichloro-dioxo-(4,4'-dicarboxylato-2,2'-bipyridine)- $Mo^{(VI)}$ complex (labelled as $Mo^{(VI)}Cl_2O_2Bipy$) grafted on the surface of various mesoporous titania powders. Two different TiO_2 synthesis routes were explored with the aim of highlighting the influence of the porosity on the dispersion of the $Mo^{(VI)}Cl_2O_2Bipy$ complex, and hence, on the resulting photo-catalytic properties. High surface area mesoporous TiO_2 supports with anatase walls were prepared following a CTAB-assisted hydrothermal process, as well as by a sol-gel route under supercritical carbon dioxide ($SC\ CO_2$). As a non-toxic and environmentally benign green solvent, $SC\ CO_2$ exhibits many unique features and great versatility in materials processing and synthesis. Such characteristics allow for a better control of the nanoarchitecture and the textural properties of the resulting materials [22].

Several complementary techniques, including XRD, nitrogen sorption isotherm, Raman, ATR-FTIR, UV–vis and ^{13}C CP-MAS NMR spectroscopies were used to characterize the parent mesoporous TiO_2 supports and the $Mo^{(VI)}Cl_2O_2Bipy$ grafted counterparts. Finally, the stability of these hybrid catalysts was evaluated after long-term photo-epoxidation experiments (56 h).

2. Experimental

2.1. Materials

All chemical reagents were purchased and used without further purification unless otherwise specified. All solvents were thoroughly degassed prior to use. Acetonitrile was distilled and kept under argon. Cetyltrimethylammonium bromide (CTAB) and HCl (37%) were purchased from Acros, while ethanol (absolute for analysis) and NaOH were obtained from Aldrich. Titanium oxide (Degussa-P25) with a specific area of $50\ m^2\ g^{-1}$ and a crystalline structure composed of 80% anatase and 20% rutile was used for comparison. All supports were dehydrated at $60\ ^\circ C$ under 5×10^{-4} mbar for 48 h prior to the anchoring procedure.

2.2. Catalysts synthesis

2.2.1. CTAB-assisted hydrothermal route

The hydrothermal preparation route of the mesoporous TiO_2 was based on the following procedure. An aqueous solution of titanium sulphate (1.1402 g) was first prepared and added to a CTAB ethanolic solution under stirring. The molar ratio of $Ti(SO_4)_2:CTAB$ is 1:0.12. The resulting mixture was then aged at room temperature for 12 h before being transferred to an autoclave at $100\ ^\circ C$ for 72 h. After the hydrothermal treatment, the solid was recovered by centrifugation, washed with water and ethanol, dried at $120\ ^\circ C$ and then calcined at $400\ ^\circ C$ in air for 6 h with a heating rate of $2\ ^\circ C\ min^{-1}$ [23]. The TiO_2 sample was labelled as TiO_2 (HT-400).

2.2.2. $SC\ CO_2$ -assisted sol-gel route

This procedure consists in the elaboration of mesoporous TiO_2 by a sol-gel route under supercritical CO_2 through hydrolysis-condensation of titanium tetraisopropoxide at $150\ ^\circ C$ under 30 MPa. Briefly, titanium tetraisopropoxide was introduced into the reactor prior to the injection of CO_2 . As soon as the desired reaction was reached, an excess of water was added under supercritical conditions. The alcohol formed through hydrolysis as well as the

excess of precursor were separated from the reaction medium during the CO_2 expansion. The TiO_2 sample synthesized under $SC\ CO_2$ was referred to as TiO_2 (SC-150).

2.2.3. $Mo^{(VI)}Cl_2O_2Bipy$ grafting procedure

The covalent grafting of the dichloro-dioxo-(4,4'-dicarboxylato-2,2'-bipyridine)- $Mo^{(VI)}$ complex on the surface of both dehydrated titania supports was carried out according to the following procedure. Hexamethyldisilazane was added to a benzene suspension of TiO_2 , stirred 24 h at room temperature, filtered, washed thoroughly with benzene and dried under vacuum. A benzene suspension of freshly prepared trimethylsilylated TiO_2 was then added to a benzene suspension of 2,2'-bipyridyl-4,4'-dicarboxylic acid, stirred 72 h at room temperature, filtered and thoroughly washed with benzene. Finally, a THF suspension of MoO_2Cl_2 was added to a benzene suspension of the previously obtained solid, stirred 4 h at room temperature and the excess of solvent evaporated. The solid product was then washed with acetonitrile and dried at room temperature [20]. The dioxo $Mo^{(VI)}$ grafted titania samples were designated as follows: $Mo^{(VI)}Cl_2O_2Bipy/TiO_2$ (HT-400), $Mo^{(VI)}Cl_2O_2Bipy/TiO_2$ (SC-150) and $Mo^{(VI)}Cl_2O_2Bipy/TiO_2$ (P-25), where the label in brackets stands for the titania support prepared under hydrothermal treatment, supercritical conditions and the reference Degussa TiO_2 , respectively.

2.3. Catalysts characterization

The TiO_2 supports and the grafted catalysts were characterized by powder X-ray diffraction (XRD) using a Bruker AXS D8 Advance DaVinci geometry with monochromatized $Cu\ K\alpha$ radiation ($\lambda = 1.5418\ \text{\AA}$) at 40 kV and 30 mA. The diffraction patterns were recorded in the 2θ value range of $20\text{--}70^\circ$ (with a step size of 0.01° and a step time of 0.4 s) for wide-angle analysis and at $2\theta = 2\text{--}10^\circ$ for low-angle analysis. Nitrogen adsorption/desorption isotherms at 77 K were measured on a Micromeritics Tristar 3000 apparatus at $-196\ ^\circ C$. Before analysis, the samples were degassed under vacuum at $150\ ^\circ C$ for 12 h. The specific surface area was determined from the linear part ($0\text{--}0.23\ P/P_0$) of the BET plot. The total pore volume was measured from the isotherms at $P/P_0 = 0.95$ and the mean pore diameter was determined by the BJH method applied to the adsorption branch. Raman analysis of both supports and catalysts was carried out on a LabRAM HR800UV HORIBA Jobin Yvon spectrometer with a BXFM microscope, equipped with a CCD detector and an internal HeNe laser (632.81 nm) and an external Ar laser (514.532 nm). All grafted $Mo^{(VI)}Cl_2O_2Bipy/TiO_2$ catalysts were characterized by attenuated total reflectance (ATR FT-IR) spectroscopy on a Perkin Elmer spectrometer and ^{13}C CP-MAS NMR spectroscopy on a Bruker Avance-400 MHz NMR spectrometer equipped with a 4 mm magic angle spinning (MAS) probe head for solids. The CP technique was applied during Magic Angle Spinning (MAS) of the rotor at 10 kHz. A ramped 1H-pulse starting at 100% power and decreasing until 50% was used during contact time (2 ms) in order to circumvent Hartmann–Hahn mismatches. In order to obtain a good signal-to-noise ratio in ^{13}C CPMAS experiment, 2048 scans were accumulated using a delay of 2 s. The ^{13}C chemical shifts were referenced to tetramethylsilane and calibrated with the adamantane signal set at 38.48 ppm [24]. Diffuse reflectance spectra (200–700 nm) were measured on a Cary 5000 Varian spectrophotometer. The amount of grafted complex was determined by thermo gravimetric analyses (TGA) under N_2 atmosphere between 30 and $900\ ^\circ C$ with a $10\ ^\circ C\ min^{-1}$ heating ramp using a SDT-Q600 apparatus. Molybdenum elemental analysis was carried out with a Thermo S4 atomic absorption spectrophotometer (AAS), after acid digestion of the samples.

دانلود مقاله



<http://daneshyari.com/article/64546>



- ✓ امکان دانلود نسخه تمام متن مقالات انگلیسی
- ✓ امکان دانلود نسخه ترجمه شده مقالات
- ✓ پذیرش سفارش ترجمه تخصصی
- ✓ امکان جستجو در آرشیو جامعی از صدها موضوع و هزاران مقاله
- ✓ امکان پرداخت اینترنتی با کلیه کارت های عضو شتاب
- ✓ دانلود فوری مقاله پس از پرداخت آنلاین
- ✓ پشتیبانی کامل خرید با بهره مندی از سیستم هوشمند رهگیری سفارشات