Accepted Manuscript

Title: Pressurized hot water extraction of bioactives

Author: Merichel Plaza, Charlotta Turner

 PII:
 S0165-9936(15)00136-3

 DOI:
 http://dx.doi.org/doi:10.1016/j.trac.2015.02.022

 Reference:
 TRAC 14449

To appear in: Trends in Analytical Chemistry



Please cite this article as: Merichel Plaza, Charlotta Turner, Pressurized hot water extraction of bioactives, *Trends in Analytical Chemistry* (2015), http://dx.doi.org/doi:10.1016/j.trac.2015.02.022.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Pressurized hot water extraction of bioactives

Merichel Plaza, Charlotta Turner

Lund University, Department of Chemistry, Center for Analysis and Synthesis (CAS), P.O. Box 124, SE-221 00 Lund, Sweden

HIGHLIGHTS

- Review of pressurized hot water extraction of bioactive compounds, 2009-14
- Chemical and physical properties of pressurized hot liquid water
- Equipment, method optimization, applications, coupling and future prospects
- Challenges with degradation and other chemical reactions during extraction

ABSTRACT

The purpose of this review is to give the reader a thorough background to the fundamentals and applications of pressurized hot water extraction (PHWE) for the analysis of bioactive compounds. We summarize the field in the period 2009–14, and include fundamentals of water as a solvent: equipment; method optimization; applications; coupling; and, future prospects. We highlight that solvent properties of water are tunable by changing the temperature, particularly self-ionization, dielectric constant, viscosity, diffusivity, density and surface tension. Furthermore, important aspects to consider are the risk of degradation of the analytes and other potential reactions, such as hydrolysis, caramelization and Maillard reactions that may lead to erroneous results. For the extraction of bioactive compounds, we report PHWE methods based on using water of 80–175°C and short extraction times. In conclusion, PHWE provides advantages over conventional extraction methods, such as being "greener", faster and more efficient.

Keywords: Anthocyanin Bioactive compound Diterpene Extraction Polyphenol Pressurized hot water extraction Pressurized liquid extraction Solubility Subcritical water Superheated water

Abbreviations: CED, Cohesion energy density; DPPH, 2,2-diphenyl-1-picrylhydrazyl; FC, Folin-Ciocalteu; HSP, Hansen solubility parameters; MRP, Maillard reaction products; PHWE, Pressurized hot water extraction; PLE, Pressurized liquid extraction; USWE, Ultrasound-enhanced subcritical water extraction

D, Diffusion coefficient; K_w , Dissociation constant; H_v , Heat of vaporization; P, Pressure; π^* , Polarisability; ε_r , Relative static permittivity (dielectric constant); $C_{p,m}$, Specific heat capacity (isobaric, molar); T, Temperature; η , Viscosity, dynamic

Corresponding author. Tel.: +46 46 222 8125; Fax; +46 46 222 8209. *E-mail address:* <u>Charlotta.Turner@chem.lu.se</u> (C. Turner) Download English Version:

https://daneshyari.com/en/article/7689226

Download Persian Version:

https://daneshyari.com/article/7689226

Daneshyari.com