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# ACCEPTED MANUSCRIPT

## STABILIZATION OF LOW VALENT 14 GROUP METAL COMPLEXES BY 9,10-DIAMIDOPHENANTHRENE LIGAND

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#### **Abstract**

Stable germylenes and stannylenes bearing 9,10-diamidophenanthrene ligands were synthesized. Complexes were prepared by two methods: a) the reaction of dianionic lithium salts of phenanthrene-9,10-diimine with low valent 14 group metal halides (GeCl<sub>2</sub>·dioxane, SnCl<sub>2</sub>·dioxane); b) the treatment of phenanthrene-9,10-diamine with M[N(SiMe<sub>3</sub>)<sub>2</sub>]<sub>2</sub> (M = Ge, Sn). The analogous Pb(II) complex was found to be unstable. The structures of the metallenes were determined by single crystal X-ray diffraction analysis. The germylene 1 as well as the stannylene 2 react with different radicals giving unstable paramagnetic derivatives that were identified by EPR spectroscopy.

Keywords: germylene, stannylene, 9,10-diamidophenanthrene ligand, EPR, X-ray diffraction.

It is known that N-heterocyclic carbenes being sigma-donor nucleophilic reagents are able to form complexes with many elements and stabilize metals in a low valence state [1]. Metal complexes of N-heterocyclic carbenes known as a second-generation Grubbs catalysts are widely used in homogeneous catalysis [2]. In such metal derivatives carbenes behave as spectator ligands affecting the processes through a combination of steric and electronic effects without interacting with the substrate. Considerable interest in the chemistry of heavier carbene analogues [3] has been developed following the isolation of the first stable N-heterocyclic carbene in 1991 by Arduengo[4]. At present a plethora of stable silylenes[5], germylenes[6],

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