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Nickel Bis(dithiolene) Complexes for Electrocatalytic Hydrogen Evolution: A Computational Study

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Abstract

The substituent effect of dithiolene ligand on electrocatalytic activity of Ni bis(dithiolene) towards hydrogen evolution reaction (HER) are thoroughly explored with computational methods. A series of differently substituted dithiolene ligands are compared and evaluated. These computational results indicate that adding electron-withdrawing groups on the ligand or enlarging the conjugate π system of the ligand can greatly reduce the reduction potentials, while tuning the electronic structure of the non-innocent framework can optimize the geometry of dihydrogen species to achieve lower energy barrier for H₂ elimination. This study may benefit the deep understanding of the electrocatalyzed HER mechanism and give inspirations on the design of efficient nickel bis(dithiolene) electrocatalysts with low overpotential and improved efficiency.

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