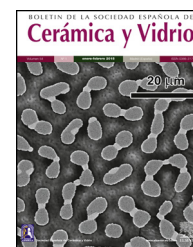




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## Q1 Review

# Preparation and characterization of clay bonded high strength silica refractory by utilizing agriculture waste

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### ABSTRACT

Clay bonded silica refractory was prepared by utilizing agriculture waste called rice husk ash (RHA) and refractory grog. Various samples were prepared with different compositions based upon partial replacement of quartz by RHA. Rectangular samples were prepared by following semi dry process prior to pressing in a uniaxial hydraulic press and sintering at a temperature of 1200 °C in air atmosphere. Various physical, mechanical and thermal characterizations were done like X-ray diffraction (XRD), scanning electron microscope (SEM), apparent porosity (AP), bulk density (BD), cold crushing strength (CCS), refractoriness and thermal conductivity measurement. The sample utilizing 30% of RHA was considered most optimum composition which produced cold crushing strength of 38 MPa and thermal conductivity of 2.08 W/mK at 800 °C with a considerable good refractoriness. Enhancement in the mechanical as well as thermal properties may be considered as attributed to the amorphous silica which has reacted more easily and efficiently with other material surrounding giving rise to the densification and produced stable crystalline phase to the refractory material. These promising characteristics suggests that the RHA may lead to be used as a potential material for the preparation of clay bonded high strength silica refractories.

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### Preparación y caracterización de sílice refractaria de alta resistencia unida a arcilla utilizando residuos agrícolas

#### RESUMEN

Se preparó sílice refractaria unida a arcilla con residuos agrícolas conocidos como ceniza de cascarilla de arroz (*rice husk ash* [RHA]) y grog refractario. Se prepararon varias muestras con diferentes composiciones basadas en la sustitución parcial de cuarzo por RHA. Las muestras rectangulares se prepararon siguiendo un proceso semiseco antes de prensarlas en una prensa hidráulica uniaxial y sinterizarlas a una temperatura de 1.200 °C en atmósfera de aire. Se realizaron diversas caracterizaciones físicas, mecánicas y térmicas, como la

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difracción de rayos X, el microscopio electrónico de barrido, la porosidad aparente, la densidad aparente, la resistencia al aplastamiento en frío, la refractariedad y la conductividad térmica. La muestra que utiliza el 30% de RHA se consideró la composición más óptima que produjo una resistencia al aplastamiento en frío de 38 MPa y una conductividad térmica de 2,08 W/m-K a 800 °C con una refractariedad considerablemente buena. La mejora tanto de las propiedades mecánicas como de las térmicas puede atribuirse a la sílice amorfa que reaccionó más fácil y eficazmente a otro material circundante, lo que produjo la densificación y una fase cristalina estable al material refractario. Estas características prometedoras sugieren que la RHA puede justificar que se use como material potencial en la preparación de sílices refractarias de alta resistencia unidas a arcilla.

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## Introduction

39<sup>Q4</sup> Silica refractories are generally known as refractory materi-  
40 als consisting more than 93% SiO<sub>2</sub> [1,2]. Silicon dioxide (SiO<sub>2</sub>)  
41 is most commonly found in nature as quartz which is also  
42 considered to be the most conventional source of silica to be  
43 used for refractory manufacturing [3]. Silica refractories are  
44 typically produced in the form of shaped products (bricks)  
45 as well as in bulk powder (castables). Silica refractory bricks  
46 possess excellent thermal shock resistance, particularly in cer-  
47 tain temperature range. Regardless the reduction in their use  
48 in the recent decades; silica refractories are still successful,  
49 beneficial or exceptional material for some of the special appli-  
50 cations such as glass tank furnace roof constructions, electro  
51 steel kilns, coke ovens, blast furnace air heaters and some  
52 construction elements in ceramic tunnel kilns and furnaces  
53 employed in the fabrication of non-ferrous metals [4–6].

54 As reported by a study, rice production around the globe  
55 reached 661 million tons in 2008 [7]. Around 20% of the rice  
56 paddy produces rice husk (RH) during its processing in rice  
57 mills and refinement. RH occupies large areas, has a very low  
58 nutritious significance and takes very long time to decompose  
59 due to which it is not appropriate for composting or manure.  
60 Thus it generates disposal problem [8]. To reduce the RH effi-  
61 ciently, one effective method used these days is to use it in  
62 the rice mills to generate steam for the parboiling process or  
63 as fuel in industries. RH generally contains 70–80% organic  
64 matter such as cellulose, lignin etc. and rest 20–30% miner-  
65 alogical components such as silica, alkalis and trace elements  
66 [9–12]. Thus when this RH is burnt, ~25% of its weight is con-  
67 verted into ash during the firing process and is called rice husk  
68 ash (RHA). This burnt RHA has different chemical, mineralogical  
69 and morphological characteristics depending upon the rice  
70 variety, soil chemistry, climatic conditions, and geographic  
71 localization of the culture as well as the process acquired dur-  
72 ing burning conditions of the husk [13]. If the combustion  
73 is incomplete, large amount of unburnt carbon is found in  
74 the ash while complete combustion results in gray to whitish  
75 ash with minimal amount of carbon contamination. This RHA  
76 contains around 85–90% of amorphous silica. However, the  
77 amorphous silica content depends upon the burning temper-  
78 ature and holding time. Optimum properties can be achieved  
79 when husk is burnt at temperatures around 500–700 °C with  
80 shorter holding time [14].

In order to utilize this type of waste as silica precursor  
in ceramics, various studies were accomplished; particu-  
larly, in white ware [15], synthesis of colorants [16,17], glaze  
preparation [18,19] and other ceramics [20,21]. Utilization  
of RHA in the synthesis of borosilicate [22], cordierite [23],  
carbosil [24], alumina-silicate [25], and mullite [26,27] are  
some of the works worth mentioning. Furthermore, this  
amorphous silica has also been employed in the prepara-  
tion of various valuable materials such as solar grade silicon  
[28], magnesium-alumina-silica [29], lithium-aluminum-silica  
[30], and silicon carbide [31]. Particularly; RHA derived silica  
refractories have also developed earlier [32,33] but the major  
objectives of these works were focused upon the application in  
insulation purpose. On the other hand, effect of incorporation  
of clay and refractory grogs in a RHA based silica refractory  
has also not been investigated yet. In this present study, we  
have prepared clay bonded high strength silica refractory with  
utilizing agriculture waste-RHA by gradually replacing quartz  
and studied the effects of this partial replacement with various  
characterization techniques.

## Materials and experimental

### Materials and materials characterization

In this work, RHA was collected from rice mills where rice husk  
is often used as a fuel. Typically in these mills, temperature  
rises above 500 °C due to which this RHA contains a very neg-  
ligible amount of volatile matter as well as carbon content.  
However, to obtain highly pure amorphous silica, this RHA was  
further heat treated at 600 °C for 2 h to eliminate the possibil-  
ity of any carbon contamination. The chemical analysis was  
carried out by using the X-ray fluorescence (XRF) spectrometer  
according to ASTM C114-00 [34] shown in Table 1. According  
to XRF results, RHA contains more than 90% of SiO<sub>2</sub>, which  
is almost similar to the amount of SiO<sub>2</sub> present in quartz.  
So, in this work complete and partial replacement of quartz  
by RHA was investigated in the preparation of silica refrac-  
tory. The quartz was purchased by Lobachemie Pvt. Ltd., India  
having analytical grade quality and used without any further  
treatment.

Microstructural analysis of RHA heat treated at 600 °C was  
carried out using a SEM analysis for which results are shown  
in Fig. 1. It can be observed that the RHA particles possess an

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