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Chitosan: A macromolecule as green Corrosion Inhibitor for mild

steel in sulfamic acid useful for sugar industry

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Abstract

The present investigation aims at investigation of low cost nontoxic carbohydrate biopolymer

chitosan as corrosion inhibitor alone and in combination with KI for mild steel in 1 M sulfamic

acid medium using gravimetric, electrochemical and surface analysis techniques. It is found that

chitosan alone exhibits inhibition efficiency of 73.8% at 200 ppm concentration. However, in

combination with KI (5ppm), it gave more than 90% inhibition efficiency. The significant increase

in the inhibition performance of chitosan has been explained by the synergistic mechanism. The

results of Potentiodynamic polarization study shows that chitosan and its blend with KI decreases

both anodic and cathodic reactions occurring at mild steel surface in 1 M sulfamic acid medium

by blocking active sites of the metal and acts as mixed type inhibitor. EIS study reveals that the

polarization resistance increases with increase in the concentration of inhibitors which increases

charge transfer resistance across the metal/solution interface. The adsorption of chitosan followed

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