Accepted Manuscript

Radium isotope response to aquifer storage and recovery in a sandstone aquifer

David S. Vinson, James R. Lundy, Gary S. Dwyer, Avner Vengosh

PII: S0883-2927(18)30017-9

DOI: 10.1016/j.apgeochem.2018.01.006

Reference: AG 4028

To appear in: Applied Geochemistry

Received Date: 21 June 2017

Revised Date: 1 December 2017

Accepted Date: 10 January 2018

Please cite this article as: Vinson, D.S., Lundy, J.R., Dwyer, G.S., Vengosh, A., Radium isotope response to aquifer storage and recovery in a sandstone aquifer, *Applied Geochemistry* (2018), doi: 10.1016/j.apgeochem.2018.01.006.

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.



ACCEPTED MANUSCRIPT

1 2	Radium isotope response to aquifer storage and recovery in a sandstone aquifer
3	David S. Vinson ^{1,*} , James R. Lundy ² , Gary S. Dwyer ³ , and Avner Vengosh ³
4 5	¹ -University of North Carolina at Charlotte, Department of Geography and Earth Sciences, 9201 University City Blvd, 324 McEniry, Charlotte, NC 28223 USA
6 7	² -Minnesota Department of Health, Environmental Health Division, 625 North Robert Street, St. Paul, MN 55164 USA
8	³ -Duke University, Nicholas School of the Environment, Box 90328, Durham, NC 27708 USA
9	*-corresponding author: dsvinson@uncc.edu, +1 704 687 5977
10	
11	Abstract
12	Radium isotopes and water-rock interaction were evaluated in an aquifer storage and recovery
13	(ASR) pilot study conducted in 2010-2011 in Minnesota (USA) in order to identify mechanisms
14	governing Ra activities when low-Ra water is recharged into a sandstone aquifer. Groundwater in the
15	aquifer selected for the study, the Cambrian Mt. Simon Sandstone, contains naturally occurring radium
16	that in many areas exceeds United States drinking water standards (185 mBq/L or 5 pCi/L combined
17	226 Ra+ 228 Ra), highlighting the need to identify the rates and mechanisms by which stored water acquires
18	Ra isotopes. Major element concentrations of recovered water largely resembled recharged water, while
19	Ra activities exceeded the Ra activities of the recharged water. ²²⁴ Ra reached ~100 mBq/L during the first
20	8 hours of recovery (overall range 15.5-133 mBq/L). ²²⁶ Ra and ²²⁸ Ra also increased during the 47-day
21	recovery period (23.7-82.5 mBq/L and 33.7-85.5 mBq/L, respectively). Ra isotope ratios indicate the
22	relative contribution of alpha recoil vs. chemical processes (e.g. adsorption/desorption). During recovery,
23	the ²²⁴ Ra/ ²²⁸ Ra and ²²⁸ Ra/ ²²⁶ Ra ratios declined, approaching their expected limiting values near unity.
24	Collectively, the rates of Ra activity change with time, trends in Ra isotope ratios, barium concentrations,
25	and manganese concentrations suggest that Ra was governed by chemical processes and alpha recoil, in
26	which the half-lives of each Ra isotope determined the primary mechanism(s) controlling radionuclide
27	mobilization from the aquifer solids. Radium-mobilizing processes during storage may include: (1)

Download English Version:

https://daneshyari.com/en/article/8863149

Download Persian Version:

https://daneshyari.com/article/8863149

Daneshyari.com