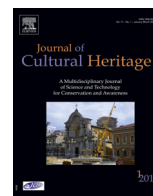




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Short note

## Multi-technique characterization of madder lakes: A comparison between non- and micro-destructive methods

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### ABSTRACT

The chemical characterization of paint material is paramount for the understanding of painting techniques, provenance studies and for assessing conservation strategies. In particular, the chemical characterization of both the organic and inorganic fraction of lakes is fundamental to assess the technologies used in their production. In this short note, we present a pilot study by comparing several micro-destructive and non-destructive methods for the comprehensive characterization of the organic and inorganic fraction of reference madder lakes. In the final procedure, the chromophores-containing molecules were separated using a sample preparation procedure based on acid hydrolysis and solvent extraction, and analysed by high-pressure liquid chromatography with UV-Vis detector (HPLC-UV/Vis). Laser induced breakdown spectroscopy (LIBS) and X-Ray Fluorescence (XRF) were used for the study of the elemental composition. Multispectral Imaging was also applied in order to evaluate its potentialities to distinguish amongst different red lakes. The final multi-technique method allowed for the characterization of both organic and inorganic fraction from the same lake micro sample.

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### 1. Introduction

A lake is an artificial pigment constituted by one or more organic natural dyes fixed by absorption or complexation on an insoluble inorganic material [1]. Organic dyes are indeed soluble in water or in other binding media and they need to be transformed into lakes, which are insoluble in binders and thus suitable to be used for painting purposes. Potash alum ( $KAl[SO_4]_2 \cdot 12H_2O$ ), ammonium alum ( $Al_2(NH_4)_2(SO_4)_4 \cdot 12H_2O$ ), iron sulphate ( $FeSO_4$ ), and tin(II) chloride ( $SnCl_2$ ) are among the salts commonly used as mordants in lake pigments making [2].

Before the introduction of synthetic dyes, the most important lakes were those obtained from anthraquinoid dyestuffs, such as madder lake, kermes lake, cochineal lake and Indian lac [3]. The characterization of lakes is of utmost importance to inform conservation practice, being lakes amongst the most fugitive materials in paintings. The source of the dye, substrate, metal cation and dye/cation ratio are all crucial parameters affecting the light-

fastness and chemical features of the lakes [1,4]. Usually only tiny amounts of examined paints can be sampled in contrast to other objects, for instance textiles, for which sampling procedures are usually more flexible.

Several methods have been developed to analyze anthraquinoid dyes in lakes painting matrices in the last twenty years, exploiting different analytical techniques and testing different sample pretreatment procedures. Most of them are based on high-pressure liquid chromatography coupled with Diode-Array, fluorescence or mass spectrometric detectors (HPLC-DAD-FD-MS) [5,6], but also 3D fluorescence and surface enhanced raman spectroscopy (SERS) are employed [7,8]. On the other hand, literature references on the characterization of inorganic mordants are scarce. Most of the studies are focused on the determination of the coordination, structures and photo-physical properties of anthraquinoid lakes [9]. Only few studies report the characterization of the mordant in textiles artworks by XRF or inductively coupled plasma-mass spectrometry [10,11], or used multianalytical approaches for the study of both organic and inorganic composition in paints [12–16].

In this paper, a multi-analytical approach has been developed to characterize three reference madder lakes containing different metal cations. Multispectral Imaging provided us a first

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discrimination. Then, a previously developed extraction procedure based on acid hydrolysis was applied to separate the inorganic fraction from the organic one. The organic phase containing the pigments was analysed by HPLC-UV/Vis. The inorganic mordant was characterized both by LIBS and XRF.

## 2. Materials and methods

### 2.1. Materials

Water (Carlo Erba, Italy), hydrochloric acid (HCl, 37%, Merck, Germany), methanol (MeOH; LC/MS grade Sigma-Aldrich USA), formic acid (LC/MS grade sigma-Aldrich USA) and n-heptane (Carlo Erba, Italy) were used for sample pre-treatment and HPLC-DAD analyses. Whatman RC Mini-Uniprep (0.20  $\mu\text{m}$  pore diameter, Sigma Aldrich) were chosen for filtration. Ultrapure water prepared with an Elga Purelab-UV system (Veolia Environment, France) was used throughout.

Reference madder lake powders N.37203 (Natural Red 9, madder red pigment precipitated with tin salt), 372051 (Natural Red 8) and 372057 (Natural Red 9, madder red pigment precipitated with copper salt) were supplied by Kremer Pigmente GmbH & Co (Germany). Linseed oil (Maimeri Company, Italy), rabbit glue (CTS, Italy) and gypsum ( $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ , CTS, Italy) were used as binder and fillers for the preparation of paint model systems. A barium sulfate ( $\text{BaSO}_4$ ) disk was used as reference white in all reflectance and imaging measurements.

### 2.2. Paint model systems

Three paint model systems were prepared on wooden panels and were composed of three layers: the preparation layer, the priming layer and the painting layer. The preparation layer, consisting in a solution of rabbit glue in water (1:16) saturated with gypsum, was applied on the panel in a six-fold coating. The priming layer, constituted by a solution of rabbit glue in water (1:32), was applied on top of the preparation layer. The ratios between glue and water are based on protocols found in the literature [17]. Three painting layers, consisting in a mixture of red lake and linseed oil, were applied on top of the priming layer. The ratio between the oil and the lake was evaluated time to time to achieve tonal uniformity.

### 2.3. Multispectral analysis

The multiband and fluorescence images were acquired using an 8.3 Megapixels Chroma C4 scientific camera (DTA scientific instruments s.r.l., Italy). Multi-band images were acquired using six interferential filters (centred at 400, 450, 550, 600, 750 and 1.050 nm, with band-pass  $\pm 25$  nm) in front of the CCD sensor. All the images were white-balanced using a reference barium sulfate white and then combined to form single high-resolution images of the whole sample. The classical four bands (RGB+IR) were used to build RGB, false colours and UV-visible fluorescence images. The UV-visible fluorescence image was obtained illuminating the painting with UV LEDs emitting at 395 nm.

### 2.4. Sample preparation and HPLC analysis

A micro-sample ( $<0.1$  mg) taken from the surface of the lake preparation were suspended in 200  $\mu\text{L}$  of 0.1% EDTA/DMF solution (1:1), and sonicated for 1 h at 60 °C. The extract in EDTA/DMF was filtrated through a RC Mini-Uniprep filter and then injected in the HPLC-DAD system. Chromatographic separation was achieved using an analytical reverse-phase Poroshell 120 SB-C18 (2.1 mm  $\times$  50 mm, 2.7  $\mu\text{m}$ , Agilent Technologies).

The HPLC system consists of an Infinity 1260 HPLC (Agilent Technologies), equipped with a gradient pump, a vacuum membrane degasser, an injection valve, and a variable wavelength detector. The working conditions were: column temperature 40 °C, injection volume 2  $\mu\text{L}$ , flow rate 0.8 mL/min. The two eluents were 0.1% phosphoric acid (A) and acetonitrile (B). The elution program was 15% B for 1.2 min, up to 50% B in 5.8 min, up to 70% B in 2.2 min, up to 90% B in 0.3 min, hold for 1 min. UV spectra were acquired at 250–435–460 nm every 0.025 min with a resolution of 2 nm.

### 2.5. XRF analysis

All experiments were performed using the Elio portable XRF Analyzer (XGLab, Italy), equipped with a 50 keV/200  $\mu\text{A}$  X-ray tube (Rh electrode, 1 mm collimated beam on the sample) and a large-area Energy Dispersive Si-Drift detector (130 eV FWHM at Mn  $K_{\alpha}$ ).

### 2.6. LIBS analysis

All experiments were performed with a Modì double pulse LIBS instrument (Marwan Technology, Italy). LIBS spectra were acquired in collinear double-pulse mode (70 mJ per pulse at 1064 nm in 15 ns FWHM pulses); the focusing lens had a focal length of 10 cm; the signal was collected with an optical fiber, placed at an angle of 45 degrees with respect to the beam direction. The tip of the optical fiber was placed about 2 cm from the surface to collect the signal from the whole laser-induced plasma. The signal was acquired with a two-channels spectrometer (AvaSpec-2048-2, Avantes, Netherlands) covering the spectral range from 180 to 900 nm; the spectral resolution of the spectrometer is 0.1 nm between 180 and 450 nm and 0.3 nm from 450 to 900 nm. LIBS spectra were acquired applying a delay of 1  $\mu\text{s}$  between the two pulses, 300 ns after the second laser pulse. The integration time was 2.5 ms [18].

## 3. Results and discussion

### 3.1. Multispectral analysis

Spectral characterization of the lakes was performed by combining triplets of single-band images, as reported in Table 1S. The obtained images for the three lakes are reported in Fig. 1. The comparison shows that the false color images, both in visible light and in fluorescence, only marginally improve the differences between lakes. The RGB images highlight differences in the lake casting on the wooden panel, but not on the color of the lakes themselves. As can be noted from Fig. 1, the near infrared images do not show noticeable differences between the three samples. On the other hand, fluorescence and false color images allow us to distinguish sample 37203 from the others.

The reflectance values of the three lakes were calculated for all the spectral bands by averaging raw reflectance data of the images. Values were also normalized by using barium sulphate disk as white reference. The obtained reflectance spectra are reported in Fig. 1S (left). The clearest marker for identifying each spectrum is the spectral behavior in the 500–700 nm region, where the spectral slope makes it possible to distinguish between the three different pigments. The 37203 lake shows a slightly different behavior from the other two lakes, as already observed in Fig. 1, at 600 nm.

### 3.2. XRF analysis

XRF spectra (Fig. 2) were acquired for the paint model systems as-is. High sulfur and calcium signals, deriving from the gypsum in the preparation layer, were observed in all three spectra. Chlorine and iron were also observed in all spectra and their presence is due to the lake process [2,19]. A small signal around 3.4 KeV, ascribable

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