



# Soil microbial anaplerotic CO<sub>2</sub> fixation in temperate soils

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## ABSTRACT

In soils both  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values of soil organic matter (SOM) tend to increase with soil depth. This change in soil C and N isotopic composition is at least partially linked to soil microbial activity that has been suggested to drive fractionation during decomposition, preferential microbial decomposition of components of soil organic matter and mixing of diverse sources of C. Soil microbes are capable of fixing N<sub>2</sub> and CO<sub>2</sub> from the soil atmosphere. Apart from surface photosynthetic CO<sub>2</sub> fixation and chemoautotrophic fixation, dark anaplerotic (i.e. non-photosynthetic) fixation of CO<sub>2</sub> is especially important for provision of C-skeletons for amino acid synthesis. We hypothesized that these N<sub>2</sub> and CO<sub>2</sub> fixing processes may contribute to determining SOM  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values. Soils from 10 sites across South Africa differing in soil properties were incubated in the dark for 3 d under continuous exposure to <sup>13</sup>CO<sub>2</sub>- and <sup>15</sup>N<sub>2</sub>-enriched atmospheres with varying soil moisture (10, 50 and 100% of field capacity) and temperature (4, 25, 40 °C). We did not detect significant N<sub>2</sub> fixation in any treatment. Significant soil microbial anaplerotic CO<sub>2</sub> fixation, however, occurred in all soils. Highest rates of anaplerotic CO<sub>2</sub> fixation occurred in soils at 50% field capacity and 25 °C, suggesting a link with microbial biotic activity. Soils with low C and N concentrations and low C:N ratios exhibited the highest rates of CO<sub>2</sub> fixation, indicating a possible link between anaplerotic CO<sub>2</sub> fixation rates and soil nutrient status. The higher rates of CO<sub>2</sub> fixation in soils with low nutrients may indicate that soil microbes rely increasingly on anaplerotic fixation as SOM-N declines, forcing greater reliance on de novo amino acid synthesis, and thus anaplerotic CO<sub>2</sub> fixation. Diffusion of bulk atmospheric CO<sub>2</sub> ( $\delta^{13}\text{C}$  ca. −10‰) into the soil atmosphere ( $\delta^{13}\text{C} \ll -10\text{‰}$ ) drives soil atmospheric CO<sub>2</sub>  $\delta^{13}\text{C}$  values up towards those of the bulk atmosphere. Anaplerotic CO<sub>2</sub> fixation in this CO<sub>2</sub> may contribute to determining soil  $\delta^{13}\text{C}$  values.

## 1. Introduction

Both  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values of soil organic matter (SOM) tend to increase with depth in soil in tropical, temperate and boreal systems (Hobbie and Ouimet, 2009). While a number of processes have been suggested to play a part in this trend, soil microorganisms play a key role in the fractionation of C and N compounds which enter the soil environment (Wynn et al., 2006; Hobbie and Ouimet, 2009). Indeed, many of the biogeochemical processes in soils are microbially mediated and a wide range of autotrophic and heterotrophic microorganisms are involved in below-ground soil processes that include mineralization, oxidation and assimilation of C and N into forms available to the plants (Curiel Yuste et al., 2007). Microbial mineralization proceeds through decomposition of more complex molecules into simpler forms. During decomposition, organic N is converted into forms that are readily available to plants (e.g. NO<sub>3</sub><sup>−</sup>) while C is lost as CO<sub>2</sub> during soil respiration (Knoepp and Swank, 1998; Schimel and Bennett, 2004).

The processes of C and N mineralization are highly spatially and

temporally heterogeneous at the global scale, and are regulated by a number of factors, including climate and soil characteristics (Knoepp and Swank, 1998; Chen et al., 2013). Microbial mineralization involves biochemical reactions that are sensitive to activation energies, substrate supply and moisture. This mineralization leads to isotopic fractionation with retention of heavier isotopes, i.e. <sup>13</sup>C and <sup>15</sup>N relative to <sup>12</sup>C and <sup>14</sup>N, respectively, with the limit that no fractionation can occur under substrate-limiting conditions where reactants are entirely converted into products (Cerling et al., 1991). This fractionation of SOM is most evident from <sup>13</sup>C enrichment through soil depth profiles and has been attributed to four explanations: 1) The influence of historical changes in atmospheric CO<sub>2</sub>  $\delta^{13}\text{C}$  values driven by fossil fuel use, 2) Microbial respiratory fractionation during decomposition, 3) Preferential microbial decomposition of less recalcitrant fractions that have lower  $\delta^{13}\text{C}$  values and 4) incorporation of a proportion of atmospheric CO<sub>2</sub> into microbial biomass (Ehleringer et al., 2000). Indeed, it is likely that some combination of these processes contribute to changes in SOM  $\delta^{13}\text{C}$  with depth (Boström et al., 2007). The role of microbial respiratory

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fractionation has received much attention with the suggestion that this leads to the fractionation of both C and N as lighter isotopes are progressively used up as mineralization progresses, leaving SOM  $^{13}\text{C}$ - and  $^{15}\text{N}$ -enriched (Price and Sowers, 2004). Kinetic fractionation of soil C by decomposition is, however, controversial as a number of studies have found contradictory evidence for the role of respiration in fractionation. For example, several studies have demonstrated no kinetic fractionation during decomposition (Lin and Ehleringer, 1997; Ekblad and Högberg, 2000; Ekblad and Nordgren, 2002; Boström et al., 2007; Breecker et al., 2015; Hall et al., 2017), while others have demonstrated a significant fractionation of respired  $\text{CO}_2$  during decomposition (Fernandez et al., 2003; Šantrůčková et al., 2000a, 2000b). Amongst microbes, however, some are able to fix  $\text{CO}_2$  photosynthetically, chemolithotrophically and anaplerotically and some also fix  $\text{N}_2$ . These processes incorporate C and N into biomass, possibly resulting in microbes exhibiting  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values, respectively, closer to those of the source  $\text{CO}_2$  and  $\text{N}_2$ .

Soil is generally a net source of  $\text{CO}_2$ , although some microorganisms photosynthetically fix  $\text{CO}_2$  in surface soils, and this can be an important process in determining the  $\delta^{13}\text{C}$  value of soils (Macdonald et al., 2015). Microbes are commonly considered to have a negligible positive contribution to soil C, however, because the C input from plants into the SOM pool is commonly large, but is approximately balanced by the mineralization of the same SOM (Chadwick et al., 1994). There are a large number of soil organisms from the archaeal and bacterial domains that are capable of dark anaplerotic  $\text{CO}_2$  fixation in a wide range of aerobic to anaerobic conditions (Saini et al., 2011). Anaplerotic  $\text{CO}_2$  fixation refers to the process of replenishing TCA cycle intermediates (Atomi, 2002) that is apparently ubiquitous in living organisms. Anaplerotic reactions are important in all heterotrophic organisms as they play the role of regenerating the intermediates of the citric acid cycle (TCA cycle) which have been withdrawn for use in the biosynthesis of amino acids and other compounds (Feisthauer et al., 2008). This links anaplerotic reactions to microbial biomass and growth (Miltner et al., 2005) with up to 10% of the total cell C derived from anaplerotic  $\text{CO}_2$  fixation (Perez and Matin, 1982). Anaplerotic reactions typically use the enzyme phosphoenolpyruvate carboxylase (PEPc) to incorporate dissolved inorganic carbon in the form of bicarbonate into the TCA cycle where organic acids are formed and used for biosynthesis (Taybi et al., 2004). PEPc, however, also functions in plant roots where it incorporates a significant amount of soil derived  $\text{CO}_2$  into organic and amino acids (Cramer et al., 1993).

Since anaplerotic  $\text{CO}_2$  fixation is common in microbes and in plant roots, it is surprising that few studies have addressed its relevance for soil microorganisms (Šantrůčková et al., 2018). The potential role of anaplerotic  $\text{CO}_2$  fixation in influencing  $\delta^{13}\text{C}$  values of SOM has, however, been suggested (Ehleringer et al., 2000; Boström et al., 2007), and there is empirical support for the role of anaplerotic  $\text{CO}_2$  fixation in SOM  $^{13}\text{C}$  enrichment with depth in arctic soils (Šantrůčková et al., 2018). The influence of anaplerotic  $\text{CO}_2$  fixation on SOM  $\delta^{13}\text{C}$  is likely to vary with depth because the  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values of  $\text{CO}_2$  and  $\text{N}_2$  in the soil are not constant, being influenced by the degree of mixing between soil and bulk atmospheric reservoirs of these gases (Susfalk et al., 2002; Clough et al., 2003) and by diffusion of  $\text{CO}_2$  through the soil profile (Cerling et al., 1991). For example, shallow soils have a larger component of atmospheric gases and a steep gradient of soil  $\text{CO}_2$   $\delta^{13}\text{C}$  values (Cerling et al., 1991), whereas diffusional limitations result in dilution of atmospheric  $\text{CO}_2$  deeper in the soil (Susfalk et al., 2002). Both waterlogging (Greenway et al., 2006) and soil texture (Wynn et al., 2005) play a role in determining the extent of mixing of soil gases. While the mixture of gases and their consequent isotopic characters may be irrelevant for respiration occurring during decomposition, it is relevant for processes, such as  $\text{N}_2$  fixation and anaplerotic  $\text{CO}_2$  fixation, that incorporate these gases into biological structures.

There are a number of factors which may influence the rates of microbial soil anaplerotic  $\text{CO}_2$  fixation. For example, the addition of

readily degradable organic C (e.g. manure) increases soil heterotrophic  $\text{CO}_2$  fixation (Wu et al., 2015). Since microbial abundance in soil is strongly linked to both moisture and temperature (Fierer et al., 2009), these factors are also likely to strongly influence the anaplerotic  $\text{CO}_2$  fixation rate. For example, Tiwari et al. (1987) demonstrated that both waterlogged and air-dried soils evolve significantly less  $\text{CO}_2$  (through respiration) than soils at field capacity, as microbial activity decreases. Similarly, extreme temperatures decrease soil respiration rates and the highest rates are found to occur in soils between 15 and 30 °C, although the respiration rate of some soils has been shown to continue increasing up to 40 °C (Lloyd and Taylor, 1994). This would mean that extreme soil moisture and temperature conditions would likely suppress anaplerotic  $\text{CO}_2$  fixation, as microbes are far less active.

We hypothesized that  $\text{N}_2$  fixation and anaplerotic  $\text{CO}_2$  fixation are common in soils, not only in the arctic (e.g. Šantrůčková et al., 2018), and that this would result in changes in soil  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$ . We also hypothesized that these fixation rates would vary with environmental circumstances that modify the prevalence of soil microbes. Furthermore, soils in which C and N are less available may exhibit higher rates of  $\text{N}_2$  fixation and anaplerotic  $\text{CO}_2$  fixation compared to soils which are nutrient rich. We determined the effect of soil microbial  $^{15}\text{N}_2$ - and anaplerotic  $^{13}\text{CO}_2$ -fixation on soil  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  values under varying soil conditions in order to ascertain whether these are incorporated into soils. To achieve this, we collected 10 soils from different environments and exposed them to atmospheres enriched with  $^{13}\text{CO}_2$  and  $^{15}\text{N}_2$  and monitored the incorporation of these isotopes into soil organic matter with varying soil moisture and temperature.

## 2. Materials and methods

In order to determine the effects of soil conditions on both  $\text{CO}_2$  and  $\text{N}_2$  fixation,  $^{13}\text{CO}_2$  and  $^{15}\text{N}_2$  were supplied to soils sampled from a variety of edaphic circumstances. The effects of microbial activity, soil moisture and temperature on fixation rates of  $\text{CO}_2$  and  $\text{N}_2$  were then determined from changes in  $\delta^{13}\text{C}$  and  $\delta^{15}\text{N}$  of the soil to provide estimates of the responses of anaplerotic  $\text{CO}_2$  fixation and  $\text{N}_2$  fixation, respectively. We used sterilized soils and unenriched soils to provide controls for microbial fixation of  $\text{CO}_2$  and  $\text{N}_2$ .

### 2.1. Field sampling

Soil (A-horizon, 0–15 cm) was collected using a soil auger (8.25 cm diameter) at 10 locations across South Africa (Table 1). Surface litter and organic matter accumulation was scraped off the collection site before cores were collected. Soils were processed within 24 h or undisturbed cores were kept at 4 °C for up to 7 d prior to processing.

### 2.2. Soil variability

Most of the soils were collected from within the Cape Floristic Region, but in diverse settings representing both open-canopy nutrient poor Fynbos vegetation on shale-derived soils (Table 1; soils 2, 3) and closed canopy Afrotemperate forests both on relatively nutrient-rich granitic (soil 4) and nutrient-poor sandstone-derived soils (soils 5, 6, 10). Within this region we also collected soil from a shale-derived Renosterveld site (soil 1). Further afield, samples were taken from shale-derived Karoo soils (soils 7, 8) as well as grassland sites (soil 9). These samples exhibited strong variability in both physical and chemical properties. For example, while clay was ubiquitously low in these samples, silt content ranged between ca. 25% and 70% (Fig. A.1).

### 2.3. Field capacity

Approximately 100 ml of oven dried soil (40 °C for 24 h) were weighed into funnels with a wad of glass wool in the neck. Funnels with soil were placed into graduated cylinders and 100 ml of water was

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