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Insights into the plasticity of Ag₃Sn from density functional theory

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Abstract

The intermetallic Ag₃Sn is a strengthening phase in both silver amalgams as well as Sn–Ag lead-free solders, but is also a relatively ductile compound. Experimental reports have suggested the compound can both readily slip via dislocations as well as deform via twinning. To better understand the deformation mechanisms in Ag₃Sn, we investigate the energetics of slip via generalized stacking fault (GSF) curves as well as the structure and energetics of the twins using both density functional theory and analytical modeling. Our results suggest that [100](001) slip, which is equivalent to $\langle a \rangle_{\text{hcp}}$ slip on a prism plane in the hcp structure, is favorable due to a very low antiphase boundary energy. Slip on the (010) plane, which is equivalent to the basal plane in hcp metals, can occur in either the [100] or [001] directions via the formation of four superpartials. Our results further suggest that slip in the $\langle 102 \rangle$ direction could occur easily on either the {201} or {211} planes with the former being more likely since it is similar to prism planes in the hcp crystal. The overall values of the GSF curves are quite low, similar to high stacking fault energy metals, resulting in a rather ductile intermetallic. Simulations of the twins in Ag₃Sn demonstrate that they are structurally similar to twins in hcp metals with similar energy magnitudes. However, the experimentally observed activation of the {011} and {211} is likely a result of the small shear required to form the twins, and the suppression of other twin types is likely a result of chemical ordering. Finally, our results point out that twinning is likely a necessary deformation mode to account for a general state of plastic deformation.

Keywords: Ag₃Sn; twinning; dislocations; nanostructure; plastic deformation; intermetallics; density functional theory;

1. Introduction

Ag₃Sn is an intermetallic that has been of interest for a number of years since it is one of the main constituents of silver amalgams used in dentistry [1–3]. Ag₃Sn appears as a major

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