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### **Bioavailability of pollutants and chemotaxis** Tino Krell<sup>1</sup>, Jesús Lacal<sup>1</sup>, Jose Antonio Reyes-Darias<sup>1</sup>, Celia Jimenez-Sanchez<sup>2</sup>, Rungroch Sungthong<sup>2</sup> and Jose Julio Ortega-Calvo<sup>2</sup>

The exposure of bacteria to pollutants induces frequently chemoattraction or chemorepellent reactions. Recent research suggests that the capacity to degrade a toxic compound has co-evolved in some bacteria with the capacity to chemotactically react to it. There is an increasing amount of data which show that chemoattraction to biodegradable pollutants increases their bioavailability which translates into an enhancement of the biodegradation rate. Pollutant chemoreceptors so far identified are encoded on degradation or resistance plasmids. Genetic engineering of bacteria, such as the transfer of chemoreceptor genes, offers thus the possibility to optimize biodegradation processes.

#### Addresses

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### Introduction

Biodegradation of anthropogenic organic chemicals (AOCs) in natural and engineered environments is often not as efficient as expected due to a limited bioavailability, which represents the accessibility of a chemical for biotransformation and toxicity. As a result of sorption to soils and sediments, pollutants often only exhibit weak chemical activity gradients that promote their uptake and transformation by cells. Thus, the biodegradation rates may reflect the dependencies of restricted phase exchanges, and the pollutants, together with their environmental risks, may persist for longer periods of time (Figure 1). Research over the last decade has shown that the chemotactic movement of bacteria increases the bioavailability, which in turn was found to have a beneficial role in bioremediation. In this review, we consider chemotaxis as the diverse tactic reactions to pollutants of bacteria exhibiting flagellar motility. The focus will be on AOCs, but due to their environmental relevance, the recent advances in the field of chemotaxis to inorganic pollutants, such as metals and nanomaterials, will also be reviewed.

# Chemotaxis towards and away from pollutants

Chemotaxis has been extensively studied in enterobacteria that show chemotaxis to a limited number of compounds like amino acids, organic acids and sugars [1]. Many free-living bacteria have an increased number of chemoreceptors, which were shown to mediate chemotaxis to a wider range of compounds as compared to enterobacteria [2,3]. Interestingly, many of these compounds are chemicals of environmental concern. Chemoattraction was observed for example towards biphenyl, benzoic acid and chlorobenzoic acids [4<sup>••</sup>], toluene and its derivatives [5.6<sup>•</sup>], naphthalene and its derivatives [6<sup>•</sup>.7]. nitroaromatics [8], chloroaromatics [9], chloronitroaromatics [10<sup>•</sup>], aminoaromatics [11], explosives [12], aliphatic hydrocarbons [13] and herbicides [14<sup>•</sup>] in species like Rhizobium sp., Bradyrhizobium sp., Pseudomonas sp., Azospirillum sp., Ralstonia sp., Burkholderia sp. or Flavimonasoryzihabitans. In a significant number of cases the physiological relevance of chemoattraction to pollutants lies in the fact that these compounds serve as carbon and energy sources. This may be exemplified by the chemotaxis towards toluene or naphthalene by the Pseudomonas putida strains DOT-T1E and G7, respectively. Both strains possess specific degradation routes for these chemoattractants [15,16]. In some cases chemoattraction was observed towards pollutants that are not metabolized by the bacterium and the physiological relevance of this behavior is little understood.

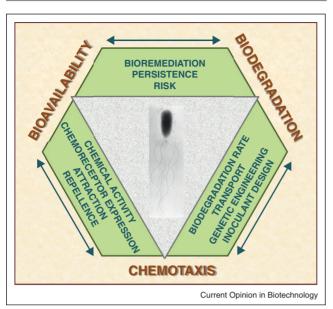
Given the toxic potential of most pollutants, it is conceivable that bacteria have also evolved chemorepellent responses. Bacterial repellence has been reported, for example, to hydrogen peroxide, hypochlorite and Nchlorotaurine [17], the PAHs anthracene and pyrene [18],  $\text{Co}^{2+}$  and  $\text{Ni}^{2+}$  [19], and silver nanoparticles [20]. Some chemicals can even be chemo-attractants for one bacterial species and be repellent for another [5,21,22]. The physical state of the chemical also appears to influence the type of response, since it was shown that the naphthalene degrader *P. putida* G7 was repelled by naphthalene in the vapor phase, whereas it was attracted

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Conceptual framework of this review. Biodegradation, bioavailability and chemotaxis are interconnected through a variety of concepts related to bacterial physiology and genetics, bioremediation performance and environmental risk.

when the compound was dissolved in the aqueous phase  $[23^{\bullet\bullet}]$ . In the light of such results one has to keep in mind that an observed chemotaxis phenotype can be the result of the action of several, potential antagonistic chemoreceptors that differ in their sensitivity to a given compound.

The complexity of the chemotactic reactions to pollutants is also illustrated by the dissimilar reactions exhibited by P. putida to silver nanoparticles and Ag<sup>+</sup> ions [20]. Nanoscale silver induced a repellent response, possibly due to a direct effect of the nanoparticles on bacterial cells, and not due to the release of soluble Ag<sup>+</sup> from the particles. Indeed, the bacterium did not show any repellent response to soluble Ag<sup>+</sup>, which is in agreement with earlier reports, which demonstrate that some harmful chemicals, such as  $Cu^{2+}$ , do not induce negative taxis in bacteria [19]. Furthermore a positive tactic response was detected at low concentrations of silver nitrate, that probably reflects the physiological role of Ag<sup>+</sup> ions and other metal ions like Mn<sup>3+</sup> and Fe<sup>3+</sup> towards which taxis was observed [24-26]. The bacterial repellent responses can be considered as a prelude of toxicity because it is often observed only at sub-lethal pollutant concentrations, which suggests that taxis may indeed be a part of survival strategies aimed on minimizing the deleterious effects of toxic compounds. In addition, the molecular machinery for the detection of chemicals for tactic purposes can also be employed for analytical applications, namely the development of alternative bioassay methods.

## Pollutant chemoreceptors so far identified are present on plasmids

The specificity of a chemotactic response is determined by chemoreceptors. Two chemoreceptors for aromatic pollutants have so far been described, which are NahY [27] of the naphthalene degrading *P. putida* G7 and McpT of the toluene, benzene and ethylbenzene degrading P. putida DOT-T1E [6<sup>•</sup>]. Both receptors mediate chemoattraction towards their respective degradation substrates. Interesting parallels exist between both receptors. NahY and McpT are encoded on plasmids pNAH7 [28] and pGRT1 [29], respectively. Both plasmids contain genes that are related to either degradation of or resistance to aromatic pollutants. The pNAH7 plasmid contains genes that encode the naphthalene degradation route and the nahY gene is co-transcribed with part of these genes [27]. Plasmid pGRT1 contains two mcpT alleles which are both in the vicinity of *ttgGHI* operon that encodes the primary efflux pump responsible for solvent resistance [30]. McpT was found to mediate an extreme form of chemoattraction, termed hyperchemotaxis, towards a wide range of mono-and biaromatic compounds [6<sup>•</sup>]. The capacity of McpT to mediate a hyperchemotaxis response towards crude oil samples is illustrated in Figure 2. Due to taxis and a very high solvent resistance, cells were able to assemble on the surface of this toxic mixture of compounds. The deletion of the mcpT gene (Figure 2) abolished this capacity.

Although NahY and McpT exert a similar function, the sequence alignment of their ligand binding regions (LBRs) reveals no significant identity. However, close homologues of both receptors are found on other degradation plasmids. For example receptors showing 99% sequence identity with McpT are found on the carbazole degradation plasmid pCAR1 of P. resinovorans CA10 [31] or the toluene degradation plasmid pWW53 of P. putida MT53 [32]. In analogy, NahY homologues are found on the naphthalene degradation plasmids pDTG1 [33] or pND6-1 [34]. Due to the high sequence similarities these homologues are likely to carry out the analogous functions as McpT and NahY, which, however, has not been verified experimentally. These data suggest that there are at least two different families of pollutant chemoreceptors, of which NahY and McpT are representative members. A chemosensory signaling cascade is formed by chemoreceptors and cytosolic signaling proteins [1]. Plasmids mentioned contain chemoreceptor genes but lack those of signaling proteins. This implies that chemotaxis is mediated by the concerted action of plasmid encoded receptors and genome encoded signaling proteins. A transfer of the *mcpT* gene into strains *P. putida* KT2440 and F1 conferred the hyperchemotaxis phenotype to both strains [6<sup>•</sup>]. This indicates that McpT is able to interact with the signaling proteins that offers the possibility of conferring pollutant chemotaxis to other bacteria by chemoreceptor gene transfer. Although these pollutant

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