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1,3-Dipolar cycloaddition of diaryldiazomethanes across *N*-ethoxy-carbonyl-*N*-(2,2,2-trichloroethylidene)amine and reactivity of the resulting 2-azabutadienes towards thiolates and cyclic amides

Rodolphe Kinghat^a, Gérard Schmitt^a, Kabula Ciamala^a, Abderrahim Khatyr^{a,*}, Michael Knorr^{a,*}, Sandrine Jacquot-Rousseau^a, Yoann Rousselin^b, Marek M. Kubicki^{b,*}

^a Institut UTINAM UMR CNRS 6213, Université de Franche-Comté, 16 Route de Gray, 25030 Besançon, France

^b Institut de Chimie Moléculaire de l'Université de Bourgogne UMR 6302, Université de Bourgogne, 9 Avenue A. Savary, 21078 Dijon, France

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ABSTRACT

1,3-dipolar cycloaddition of diaryldiazomethanes $\text{Ar}_2\text{C}=\text{N}_2$ across $\text{Cl}_3\text{C}-\text{CH}=\text{N}-\text{CO}_2\text{Et}$ **1** yields Δ^3 -1,2,4-triazolines **2**. Thermolysis of **2** leads, via transient azomethine ylides **3**, to diaryldichloroazabutadienes $[\text{Ar}(\text{Ar}')\text{C}=\text{N}-\text{CH}=\text{CCl}_2]$ **4**. Treatment of **4a** ($\text{Ar} = \text{Ar}' = \text{C}_6\text{H}_5$) and **4c** ($\text{Ar} = \text{Ar}' = p\text{-ClC}_6\text{H}_4$) with NaSR in DMF yields 2-azabutadienes $[\text{Ar}_2\text{C}=\text{N}-\text{C}(\text{H})=\text{C}(\text{SR})_2]$ **5**. In contrast, nucleophilic attack of NaStBu on **4** affords azadienic dithioethers $[\text{Ar}_2\text{C}=\text{N}-\text{C}(\text{S}^i\text{Bu})=\text{C}(\text{H})(\text{S}^i\text{Bu})]$ (**7a** $\text{Ar} = \text{C}_6\text{H}_5$; **7b** $\text{Ar}' = p\text{-ClC}_6\text{H}_4$). The reaction of **4a** with NaSEt conducted in neat EtSH produces $[\text{Ph}_2\text{C}=\text{N}-\text{C}(\text{H})(\text{SEt})-\text{CCl}_2\text{H}]$ **8**, which after dehydrochlorination by NaOMe and subsequent addition of NaSEt is converted to $[\text{Ph}_2\text{C}=\text{N}-\text{C}(\text{SEt})=\text{C}(\text{H})(\text{SEt})]$ **7c**. Upon the reaction of **4c** with NaS^{*i*}Pr, the intermediate dithioether $[(p\text{-ClC}_6\text{H}_4)_2\text{C}=\text{N}-\text{CH}=\text{C}(\text{S}^i\text{Pr})_2]$ **5k** is converted to tetrakisithioether $[(p\text{-}^i\text{PrSC}_6\text{H}_4)_2\text{C}=\text{N}-\text{CH}=\text{C}(\text{S}^i\text{Pr})_2]$ **6**. Treatment of **4a** with the sodium salt of piperidine leads to $[\text{Ph}_2\text{C}=\text{N}-\text{CH}=\text{C}(\text{NC}_5\text{H}_{10})_2]$ **10**. The coordination of **6** on CuBr affords the macrocyclic dinuclear Cu(I) complex **11**. The crystal structures of **5i**, **7a,b**, **10** and **11** have been determined by X-ray diffraction.

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1. Introduction

The reactive imine ethoxycarbonyl-*N*-(2,2,2-trichloroethylidene)amine (**1**) (also named anhydrochloral-urethane), first described by Feist and Ulrich *et al.* [12], has been used in the past by several groups as a versatile starting material for a number of transformations. For example, treatment of $\text{Cl}_3\text{C}-\text{CH}=\text{N}-\text{CO}_2\text{Et}$ (**1**) with Grignard reagents

gives $\text{Cl}_3\text{C}-\text{CHRNHCO}_2\text{Et}$ ($\text{R} = \text{Me}, \text{Et}, \text{Pr}, \text{Bu}, \text{PhCH}_2, \text{CH}_2=\text{CHCH}_2$), which after hydrolysis and decarboxylation leads to α -amino acids [3]. Hetero-Michael addition of pyrazolinones to $\text{Cl}_3\text{CCH}=\text{NCO}_2\text{R}$ is reported to yield novel 1,3,4-substituted 2-pyrazolin-5-ones [4]. The synthesis of 5,6,7-substituted 1,2,4-triazolo[1,5-*a*]pyrimidines was achieved by the treatment of triazolopyrimidinols with **1** [5]. Several papers deal also with the utilisation of this imine as a heterodienophile for [4 + 2] Diels-Alder cycloadditions [6–9].

Our laboratory investigated in the late 90s the potential of the title compound as a reagent for the synthesis of heterocycles via 1,3-dipolar cycloaddition [10–12]. In this

* Corresponding authors.

E-mail addresses: michael.knorr@univ-fcomte.fr (M. Knorr), marek.kubicki@u-bourgogne.fr (M.M. Kubicki).

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context we have described and discussed the unexpected formation of 4,4-dichloro-1,1-diphenyl-2-azabuta-1,3-diene [$\text{Ph}_2\text{C}=\text{N}-\text{CH}=\text{CCl}_2$] (**4a**) by cycloaddition of diphenyldiazomethane across the imine function of **1** at 65 °C in toluene [12]. We have demonstrated that the first step of the process consists of a dipolar 1,3-cycloaddition of the diazo-compound across the imine, leading to the Δ^3 -1,2,4-triazoline **2a**. Then, extrusion of dinitrogen gives the transient azomethine ylide **3a**. Subsequent elimination of ethyl chloroformate produces the 2-azadiene **4a** in 70% yield (Scheme 1). Investigating the electronic structure and chemical properties of **4a**, we have found that it presents an interesting reactivity towards various nucleophiles, such as alkoxides, thiolates, and the sodium salt of pyrrole or the cyanide anion [13–17]. More recently, some of these compounds were reacted as *S,N*-chelators with transition metal derivatives and allowed us to access new organometallic products *via* oxidative addition and cyclometallation reaction [18,19].

In order to modulate the electronic properties of this reactive 2-azadienic array, we now have investigated two different strategies to prepare functionalized derivatives of **4a** to have a feedstock for further organic or organometallic transformations:

- (i) introduction of electron-pushing or withdrawing substituents at the *para*-positions of the phenyl rings.
- (ii) systematic investigation of the reactivity towards a wide series of thiolates SR^- and amides NR_2 to obtain functional ligands possessing both *S,N* or *N,N* donor sites for complexation studies.

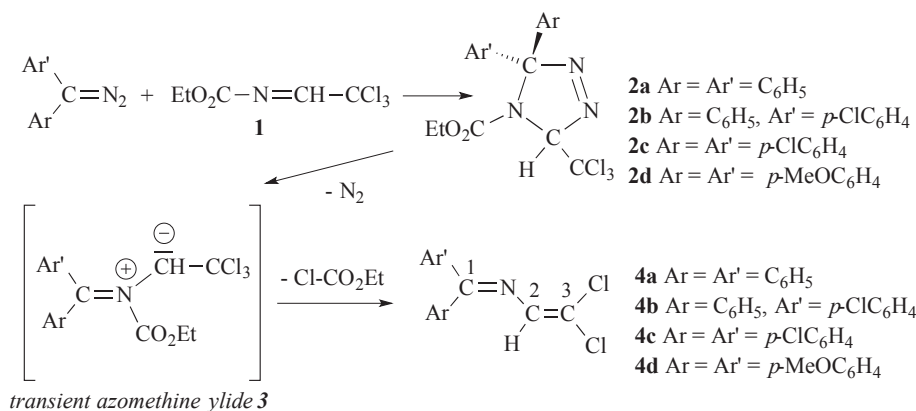
We report herein the preparation of some new functionalized 1,1-diaryl-4,4-dichloro-2-azabutadienes and the rich reactivity of these π -conjugated compounds [20] towards various thiolate nucleophiles of the type NaSR . The molecular structures of some azadienic dithioethers resulting from the unpredictable attack of thiolates have been determined by means of X-ray diffraction studies. This investigation includes also a reactivity study of **4a** toward the sodium salt of piperidine and the structural characterization of the resulting substitution product.

2. Results and discussion

2.1. Cycloaddition of diaryldiazomethanes $\text{Ar}_2\text{C}=\text{N}_2$ across $\text{Cl}_3\text{C}-\text{CH}=\text{N}-\text{CO}_2\text{Et}$

In analogy to the protocol using diphenyldiazomethane as a dipole, the cycloaddition of di(*p*-chlorophenyl)diazomethane, (*p*-chlorophenyl)phenyldiazomethane and di(*p*-methoxyphenyl)-diazomethane produces at 20 °C in toluene as solvent, the expected Δ^3 -1,2,4-triazolines **2b–d** as colourless powders (see Scheme 1). Subsequent thermolysis of these heterocyclic compounds in toluene at 60° leads to (most probably *via* the transient azomethine ylides **3** [21]) the corresponding 1,1-diaryl-4,4-dichloro-2-azabutadienes **4b–d** as pale-yellow stable solids. However, it is not necessary to isolate first the heterocyclic intermediates **2**. If the imine **1** is treated with $\text{Ar}_2\text{C}=\text{N}_2$ in warm toluene, *in situ* transformation affords straightforwardly the 2-azadienes **4**. The crude residue of **4c** is contaminated by 10–15% of tetrakis(*p*-chlorophenyl)diazine (*p*-ClC₆H₄)₂C=N–N=C(*p*-ClC₆H₄)₂ as a minor product, which can be separated by fractional crystallisation from hot ethanol [22].

As observed in the proton NMR spectrum of **4a**, the vinylic hydrogen resonances of **4c** and **4d** are quite low-field shifted and appear as singlets at δ 7.07. Furthermore, two distinct singlets due to the methoxy groups are found at δ 3.83 and δ 3.87 in the spectrum of **4d**. In the case of **4b**, two singlets in a 2:1 ratio are detected at δ 7.03 and δ 7.01 in CDCl₃ solution, respectively. Two singlets, attributed to the C=N resonance, are also observed in the ¹³C{¹H} spectrum at δ 167.0 and 166.9 in an approximate 2:1 ratio. Initially, we rationalized this finding by formation of two stereoisomers resulting from different approaches of the dissymmetric diaryldiazoalkane across the C=N bond of imine **1** in the transition state. A careful examination under an optical microscope confirmed that the morphology of all crystals of **4b** within this batch was quite homogenous, the melting point being sharp between 64–66 °C. Surprisingly, dissolving a single crystal in CDCl₃ again gave rise to two sets of signals in the 2:1 ratio. A survey of the literature revealed that this phenomenon is most probably due to a hindered inversion at the imine nitrogen. For example,



Scheme 1. Reaction of diaryldiazomethanes with imine **1**.

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