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A unique highly hydrophobic anticancer prodrug self-assembled 01 nanomedicine for cancer therapy

Guolian Ren^{a,b}, Mengjuan Jiang^a, Peng Xue^a, Jing Wang^a, Yongjun Wang, PhD^{a,*}, Bo Chen^{c,**}, Zhonggui He^a 03 02

> ^aSchool of Pharmacy, Shenyang Pharmaceutical University, Shenyang, China ^bSchool of Pharmacy, Shanxi Medical University, Shanxi, China

^cDepartment of Breast Surgery, The First Hospital of China Medical University, Shenyang, China

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Abstract 9

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In contrast with common thought, we generated highly hydrophobic anticancer prodrug self-assembled nanoparticles without the aid of 10 11 surface active substances, based on the conjugation of docetaxel to $D-\alpha$ -tocopherol succinate. The reduction-sensitive prodrug was synthesized with a disulfide bond inserted into the linker and was compared with a control reduction-insensitive prodrug. The morphology 12 and stability of self-assembled nanoparticles were investigated. Cytotoxicity and apoptosis assays showed that the reduction-sensitive 13 nanoparticles had higher anticancer activity than the reduction-insensitive nanoparticles. The reduction-sensitive nanoparticles exhibited 14 favorable in vivo antitumor activity and tolerance compared with docetaxel Tween80-containing formulation and the reduction-insensitive 15nanoparticles. Taken together, the unique nanomedicine demonstrated a number of advantages: (i) ease and reproducibility of preparation, 16 (ii) high drug payload, (iii) superior stability, (iv) prolonged circulation, and (v) improved therapeutic effect. This highly reproducible 17 molecular assembly strategy should motivate the development of new nanomedicines. 18

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20Key words: Docetaxel; Prodrug; Self-assemble; Nanomedicine; Disulfide bond

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Cancer is a leading cause of morbidity and mortality 2223worldwide, and recent advancements have resulted in a modest impact on patient survival. Nanomedicine has played an 24 important role in developing anticancer drug delivery system 25with improved efficacy and/or reduced toxicity.1 However, few 26 nanotherapeutics have been approved by the Food and Drug 27Administration for cancer therapy because there is a huge gap 28between the technical requirements and clinical translation.^{2,3} A 29wide range of nanomaterials have been employed for the 30

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*Correspondence to: Y. Wang, School of Pharmacy, Shenyang Pharmaceutical University, Shenyang, China.

**Correspondence to: B. Chen, Department of Breast Surgery, The First Hospital of China Medical University, Shenyang, China.

E-mail addresses: wangyjspu@163.com (Y. Wang), chbyxl@163.com (B. Chen).

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development of cancer therapeutics, including lipids,^{4,5} 31 inorganics,^{6,7} polymers,^{8,9} or glycan compounds.^{10,11} Further- 32 more, to improve therapeutic performance, increasingly compli- 33 cated nanostructures have been fabricated. The added 34 complexity of nanocarriers further deters their translation to 35 clinics.³ Thus the balance between the medical and societal 36 benefits should be taken into account when developing 37 anticancer nanomedicines. Anticancer nanomedicines with 38 good druggability should: (i) by simply and reproducibly 39 prepared, (ii) have high drug loading efficiency to avoid the 40 use of excessively high amounts of carrier materials, (iii) possess 41 good stability ensuring storage quality, and (iv) achieve 42 favorable therapeutic outcomes with low undesired toxicity. 43 However, the nanomedicines meeting the above criteria are 44 limited.¹² 45

In most cases, previous carrier-based nanomedicines are 46 inherently multicomponent systems, where in addition to loaded 47 active pharmaceutical ingredients, some organic or inorganic 48 materials serve as a delivery vehicle, and sometimes stealth and/ 49 or targeting functional materials are employed. The preparation 50

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of these nanomedicines is accomplished through a series of 51physical and/or chemical procedures.^{13,14} Although many 52studies have reported improved in vivo efficacy, only a few of 53have achieved clinical translation because of their unpredictable 54and inconsistent formulation outcomes.¹⁵ In recent years, the use 55of drug molecules to directly create well-define nanostructures has 5657 gained increasing interest due to their simplified formulation optimization.¹⁵ Through precise design, these drug molecules, 58including polymer-drug conjugates,¹⁶ polypeptide-drug 59 conjugates,^{17,18} and small molecule prodrugs,^{19,20} could 60 self-assemble and form one-component nanomedicines with 61 controllable and distinct physicochemical properties. The chemical 62 flexibility of small molecule prodrugs and the simplicity of 63 nano-assembly make it appealing for the development of new 64 nanomedicine. 65

A commonly used strategy to produce self-assembling prodrugs 66 couples drugs with an auxiliary segment, which endows the 67 conjugate with amphiphilicity to form a nanoassembly.²¹ Here, we 68 propose a proof-of-concept study using a highly hydrophobic 69 70 compound to self-assemble and form stable nanoparticles without any surface active agents. The self-assembled nanomedicine is 7172based on the conjugation of docetaxel (DTX) to $D-\alpha$ -tocopherol succinate (VE) with a disulfide bond (DTX-S-S-VE) or an ester 7374 bond (DTX-VE) as linker. The disulfide bond, a reduction-sensitive 75bond, was selected with the aim of enabling the controlled release of DTX from the prodrug after its passage from blood circulation 76(1-10 µM glutathione (GSH)) into intracellular compartments 77 (1-10 mM GSH).²² The ester bond, a reduction-insensitive bond, 78was chosen as a control. DTX-S-S-VE and DTX-VE both led to the 79construction of nanoassemblies by a simple solvent-displacement 80 procedure, and were termed DSNPs and DNPs, respectively. Their 81 characterization, stability in various media, DTX release, in vitro 82 cell activity, and in vivo pharmacokinetic and tissue distribution 83 behavior were investigated. In vivo antitumor efficacy and toxicity 84 were evaluated using a murine xenograft model. It was expected 85 that the unique nanomedicine might enhance the antitumor efficacy, 86 and that side effects could be reduced simultaneously. 87

88 Methods

89 Synthesis and purification of DTX-VE and DTX-S-S-VE

For DTX-VE, DTX was reacted with vitamin E (VE) succinate directly. For DTX-S-S-VE, VE was firstly reacted with dithiodiglycolic acid anhydride, then, DTX was added to the intermediate product. Both prodrugs were purified by column chromatography. Detailed procedures are shown in Supplementary Material.

95 Preparation and characterization of nanoassemblies

96 Briefly, a solution of prodrug in ethanol (10 mg/mL) was added slowly to water under stirring (800 rpm) at room 97 temperature. Ethanol was completely evaporated under a 98 vacuum by a rotary evaporator. The aqueous suspension of the 99 self-assembled nanoparticles (NPs) was obtained with light blue 100 opalescence, and stored at 4 °C. PEGylated NPs (DSNPs and 101 DNPs) were achieved just by mixing NPs with DSPE-PEG2000 102solution. 103

The particle size, zeta potential and polydispersity index 104 (PDI) of the resulting NPs were evaluated using a Zetasizer 105 (Nano ZS, Malvern, UK). The measurements were performed in 106 triplicate, and the results were expressed as mean \pm standard 107 deviation (SD). The morphology and size of samples were 108 characterized by transmission electron microscopy (TEM) 109 (Hitachi, Japan). A drop (10 µL) of NPs solution was deposited 110 on a carbon-coated copper grid. After 5 min, the excess solution 111 was removed by filter paper. Grids were then stained with 1% 112 uranyl acetate (5 µL) for 15 s and air dried. The samples were 113 imaged using a Hitachi H600 TEM operated at 100 KeV.

In vitro release study of DSNPs

The in vitro release of DSNPs was carried out by dialysis 116 method.²³ Briefly, 200 µL DSNPs (2 mg/mL) were transferred 117 into a dialysis bag. The sealed bags were immediately put into 118 conical flasks containing phosphate buffer saline (PBS) at 119 pH 7.4 with different concentrations of GSH: (i) PBS with 120 10 mM GSH, (ii) PBS with 0.1 mM GSH, (iii) PBS with 121 0.01 mM GSH, (iv) PBS with 0.002 mM GSH, and (v) pH 7.4 122 PBS, which were placed in conical flasks containing 20% 123 ethanol as solubilizer in order to improve the solubility of DTX. 124 The flasks were placed in an air bath with shaking at 100 rpm at 125 37 °C. At predesigned time intervals, 100 µL sample was 126 withdrawn and analyzed by ultra-high performance liquid 127 chromatography (Waters ACQUITY UPLC) on a reverse 128 ODS Cosmosil-C18 column (150 mm \times 4.6 mm, 5 μ m) with 129 acetonitrile/water (55:45, v/v) as eluting solution. The samples 130 were detected using an ultraviolet-visible detector at 230 nm 131 wavelength, and delivered at a flow rate of 0.8 mL/min. 132

In vitro conversion in rat plasma

The solutions of DSNPs and DNPs were incubated in rat 134 plasma at 37 °C under the shaking of 100 rpm for 24 h. Then, 135 50 μ L samples were taken for analysis at 0, 1, 2, 3, 5 and 24 h. 136 The plasma samples were processed according to the operation 137 steps of the pharmacokinetic study, and the DTX content was 138 measured by UPLC–MS/MS. 139

Physical stability

DSNPs were incubated in distilled water, RPMI 1640 141 medium, PBS at pH 7.4, and different percentages of bovine 142 serum albumin (BSA) at 37 ± 2 °C (100 rpm) for 24 h. Samples 143 were taken for analysis at pre-determined intervals. Changes in 144 particle size and PDI were chosen to evaluate the stability of 145 DSNPs. 146

In vitro anticancer activity studies

Human lung adenocarcinoma cells (A549) and human 148 prostate cancer cells (PC-3) were cultured in DMEM and 149 RPMI-1640 medium respectively, supplemented with 10% (v/v) 150 heat-inactivated fetal bovine serum (FBS), 100 μ g/mL strepto- 151 mycin and 100 U/mL penicillin. Cells were grown in a 152 humidified atmosphere with 5% CO₂ at 37 °C. The cells were 153 subcultured until 80% confluence by treatment with trypsin- 154 EDTA. 155

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