



Effects of various sulphuring methods and storage temperatures on the physical and chemical quality of dried apricots



Ali Levent Coşkun^a, Meltem Türkyılmaz^{b,1}, Özge Turfan Aksu^c, Betül Erkan Koç^d, Oktay Yemiş^e, Mehmet Özkan^{f,*}

^a Department of Food Engineering, Gaziosmanpaşa University, Taşlıçiftlik, 60150 Tokat, Turkey

^b Institute of Food Safety, Ankara University, Diskapi, 06110 Ankara, Turkey

^c Municipality of Antalya, Water Control Laboratory, 07100 Antalya, Turkey

^d Turkish Statistical Institute, Çankaya, 06100 Ankara, Turkey

^e Department of Food Engineering, Faculty of Engineering, Pamukkale University Kınıklı, 20020 Denizli, Turkey

^f Department of Food Engineering, Faculty of Engineering, Ankara University, Diskapi, 06110 Ankara, Turkey

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ABSTRACT

The effects of different sulphuring methods, i.e. sulphuring by “burning the elemental sulphites (BES),” “SO₂ gas from liquified SO₂ tank (SG)” and “dipping into sodium metabisulphite solution (DSM)” on the colour (brown colour formation and carotenoid degradation) and the loss of SO₂ in apricots from Hacıhaliloğlu and Kabaası varieties stored at 5, 20 and 30 °C for a year were investigated. There were significant effects of variety, sulphuring method and especially storage at 30 °C on the brown colour formation and loss of SO₂ ($P < 0.05$). As storage temperature-time increased, β -carotene content decreased. Sulphuring methods and variety did not show significant effect on β -carotene content ($P > 0.05$). The changes in L^* , b^* and C^* values were directly associated with β -carotene content and browning values. The most suitable method for all samples, except for Hacıhaliloğlu variety stored at 30 °C (BES), is SG, because the samples retained their attractive golden yellow colour during storage.

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1. Introduction

Dietary carotenoids provide health benefits in decreasing the risk of some diseases, particularly certain cancer and eye disease (Krinsky & Jhonson, 2005). Therefore, there is an intense interest in apricots which contain high amount of β -carotene whose beneficial effects are due to its role as an antioxidant (Krinsky & Jhonson, 2005). Apricots are grown in many countries in the world. The major apricot producers are Turkey, Iran, Pakistan, Uzbekistan and Italy. Turkey produces approximately 20% of the world's fresh apricot (716000 metric tons, Mt) and 84% of the dried apricots (DAs) (98000 Mt) (FAOSTAT, 2011). The primary importer countries of Turkish DAs with a share of ca. 60% are the USA, England, Russia, Germany, France and Australia (Asma, 2011).

Apricots have a short harvest season and a limited time of storage even under suitable conditions (Ihns, Diamante, Savage, & Vanhanen, 2011). To supply the apricots to consumers during throughout the year, different preservation methods, i.e. freezing,

canning (Ihns et al., 2011), drying without any treatment and drying after sulphuring, are commonly applied. Compared to DAs without any treatment, sulphured-DAs are much more preferred due to their characteristic golden yellow colour and taste. Sulphur dioxide (SO₂) could retard the β -carotene breakdown in DAs during drying and storage (Eheart & Sholes, 1945; Türkyılmaz, Tağı, & Özkan, 2013). Additionally, SO₂ provides both inhibition of the enzymatic browning during drying and protection from non-enzymatic browning reactions as well as prevention of microbial deterioration during drying and storage.

In Turkey, almost all of the commercial DA production is carried out in the city of Malatya, located in the south-eastern part of Turkey. In Malatya, Hacıhaliloğlu and Kabaası varieties are the main apricot varieties destined for drying. Among these two varieties, Hacıhaliloğlu cultivar is the most commonly grown cultivar in Malatya and is very suitable for drying due to its high soluble solid content (°Brix) which is between 24 and 28. In Malatya, 70% of the apricots destined for drying belong to this cultivar. The fruits are medium-sized; 25–35 g, oval-shaped, and skin and flesh colours are yellow. Fruit has a tendency to be red blushed. Flesh is firm-textured, low in water content, and very sweet and aromatic (Asma, 2011). Kabaası is the second most important cultivar for drying, which has larger fruit in size and weight (30–35 g), has

* Corresponding author.

E-mail addresses: mturkyilmaz@ankara.edu.tr (M. Türkyılmaz), mozkan@ankara.edu.tr (M. Özkan).

¹ Tel.: +90 312 596 1087; fax: +90 312 317 8711.

sweet taste and is also more resistant to spring late frosts than the other apricot cultivars, especially Hacıhaliloğlu (Asma, 2011).

In Malatya, Hacıhaliloğlu and Kabaası cultivars have always been sulphured by BES due to the simplicity of its application. However, there are some disadvantages of this sulphuring method. The maximum limit of 2000 mg SO₂ per kg of DAs is accepted by most countries, including Turkey (Codex Alimentarius Commission, 1989). However, DAs sulphured by BES method often contain much higher or lower SO₂ content than the legal limits because many factors affect the sulphuring process (e.g. variety, temperature and SO₂ gas content in the sulphur house). In fact, McBean, Johnson, and Pitt (1964) reported that the apricots sulphured by BES could contain SO₂ from less than 1000 to over 6000 mg kg⁻¹. In BES method, elemental sulphites could not always burn completely due to insufficient oxygen in sulphur house. Moreover, temperature and SO₂ concentration in sulphur house could not be controlled by BES method. As a result, losses of time, labour and quality occur with the BES method.

In this study, the effects of three sulphuring methods, i.e. sulphuring by SG and DSM as well as BES, on the colour and loss of SO₂ in Hacıhaliloğlu and Kabaası varieties during storage at 5, 20 and 30 °C for a year were investigated. Meanwhile, the correlations between reflectance colour values, total colour changes (ΔE) and browning index with browning values and β -carotene contents of the DAs were also determined.

2. Materials and methods

2.1. Materials

Fresh apricots (*Prunus armenica* L., var. Hacıhaliloğlu and var. Kabaası) were provided by the Institute of Fruit Research Center at İnönü University, Malatya.

2.2. Sulphuring methods and sun-drying

Taking into consideration the maximum limit (2000 mg SO₂ per kg of DAs) set by Codex Alimentarius Commission (1989), the production of the samples containing SO₂ at 2000 mg kg⁻¹ level by using three different sulphuring methods given below was intended (Table 1). However, since the processing parameters of the sulphuring methods were not known at the time of study and SO₂ absorption of each apricot was different, SO₂ concentrations of the samples were not accurately 2000 mg SO₂ per kg of DAs, but were close to this value.

2.2.1. Sulphuring by “burning elemental sulphites” (BES)

Fresh apricots that are in uniform size and free of physical damage were selected before sulphuring process. After sorting for

sound fruits, fresh fruits (20 kg) were placed in plastic crates (90 × 180 cm) in single layer and the crates were stacked on top of each other in the sulphur house. Then, the elemental sulphur was burned and the gate of sulphur house was closed. The amount of elemental sulphur burned was 1.8 kg ton⁻¹ and the apricots were exposed to the fumes of SO₂ in sulphur house for 12 h.

2.2.2. Sulphuring with “SO₂ gas from liquified SO₂ tank” (SG)

Similar to BES method, after sorting for sound fruits, fresh fruits (20 kg) were placed in plastic crates (90 × 180 cm) in single layer and the crates were stacked on top of each other in the sulphur house. Then, the apricots were sulphured by SG. The amount of SO₂ gas given to sulphur house was controlled by measuring the reduction in the weight of the liquified SO₂ tank on a steelyard. After target level (1–1.5 kg) of SO₂ was given to sulphur house, valve of the tank was closed. The apricots were exposed to the fumes of SO₂ in sulphur house for 3.5 h. In this method, a fan was used to ensure homogenous distribution of SO₂ gas in sulphur house, while a heater was used to accelerate the SO₂ absorption by the apricots.

2.2.3. Sulphuring by “dipping into sodium metabisulphite (Na₂S₂O₅) solution” (DSM)

In this method, the apricots (13 kg) were dipped to the Na₂S₂O₅ solution for 35 min. For the preparation of Na₂S₂O₅ solution, 3 kg of Na₂S₂O₅ was dissolved in 25 L water.

2.3. Drying

After sulphuring, the sulphured apricots in plastic crates in single layer were removed from the sulphur house and placed under direct sunlight. On the 3th day of drying, the pits were removed by squeezing the fruit by hand. At the end of 6 days of sun-drying, the drying process was terminated.

2.4. Sampling

The sulphured and sun-dried apricots were brought to Ankara University Food Engineering Department where all storage studies as well as physical and chemical analyses were carried out. The samples sulphured by different sulphuring methods were left in enclosed containers at 20 ± 0.5 °C for 2 weeks to equilibrate moisture content.

A 1-kg of the apricot samples were placed in each plastic bag made of polyethylene [gas permeability 7.5 dm³ O₂ (cm² day)⁻¹; water vapour permeability 15 g H₂O (m² 24 h)⁻¹]. The bags were then hot sealed and placed to carton package with a capacity of 2 kg. The selection of the package materials was made by taking into account of the package materials used by the packing houses in Malatya.

Table 1

Some physical and chemical properties of the dried apricot samples sulphured by three different methods before storage.

Properties	BES ^a		SG ^b		DSM ^c	
	Hacıhaliloğlu	Kabaası	Hacıhaliloğlu	Kabaası	Hacıhaliloğlu	Kabaası
Moisture (%)	22.58 ± 0.02Aa ^d	19.01 ± 0.00Ab	22.27 ± 0.10Aa	17.34 ± 0.29Bb	21.27 ± 0.03Ba	15.41 ± 0.23Cb
<i>a</i> _w (water activity)	0.69 ± 0.00Aa	0.62 ± 0.01Ab	0.68 ± 0.00Aa	0.60 ± 0.00Bb	0.66 ± 0.01Ba	0.56 ± 0.00Cb
pH	4.28 ± 0.04Ba	3.81 ± 0.03Bb	4.15 ± 0.02Ca	3.77 ± 0.04Bb	4.86 ± 0.02Aa	4.60 ± 0.02Ab
Titrateable acidity (g 100 g ⁻¹ dry weight)	1.73 ± 0.05Ab	2.04 ± 0.12Aa	1.79 ± 0.08Ab	2.06 ± 0.10Aa	1.21 ± 0.07Bb	1.76 ± 0.05Ba
SO ₂ content (mg kg ⁻¹)	2364 ± 20.2Bb	2609 ± 33.0Aa	2573 ± 17.3Aa	2218 ± 19.7Bb	1703 ± 27.6Ca	1456 ± 31.0Cb
Browning value (A ₄₂₀ g ⁻¹ dry weight)	0.27 ± 0.01Ba	0.24 ± 0.00Bb	0.25 ± 0.01Ba	0.21 ± 0.04Bb	0.37 ± 0.01Aa	0.28 ± 0.00Ab
β -Carotene content (mg 100 g ⁻¹ dry weight)	31.3 ± 1.02Aa	28.1 ± 0.94Aa	30.0 ± 1.01Ba	26.4 ± 1.11Bb	33.3 ± 0.9Aa	30.8 ± 1.27Ab

a–b: values with different letters within lines indicate significant difference between varieties (Duncan test, $P < 0.05$).

^a BES : sulphuring by “burning the elemental sulphites”.

^b SG : sulphuring by “SO₂ gas from liquified SO₂ tank”.

^c DSM: sulphuring by “dipping into sodium metabisulphite solution”.

^d A–C: values with different letters within lines indicate significant difference between sulphuring methods (Duncan test, $P < 0.05$).

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