

Contents lists available at ScienceDirect

### Journal of Fluorine Chemistry



journal homepage: www.elsevier.com/locate/fluor

# Regioselective synthesis of 5-trifluoromethyl-1,2,3-triazole nucleoside analogues via TBS-directed 1,3-dipolar cycloaddition reaction

Zhiru Xiong<sup>a</sup>, Xiao-Long Qiu<sup>b</sup>, Yangen Huang<sup>a</sup>, Feng-Ling Qing<sup>a,b,\*</sup>

<sup>a</sup> College of Chemistry, Chemical Engineering and Biotechnology, Donghua University, 2999 North Renmin Road, Shanghai 201620, China <sup>b</sup> Key Laboratory of Organofluorine Chemistry, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China

#### ARTICLE INFO

Article history: Received 18 September 2010 Received in revised form 14 December 2010 Accepted 15 December 2010 Available online 23 December 2010

Keywords: Nucleoside analogues Triazole Trifluoromethyl group 1,3-Dipolar cycloaddition Regioselective synthesis

#### ABSTRACT

Herein described was a straightforward method for the highly regioselective synthesis of 5-trifluoromethyl-1,2,3-triazole nucleoside analogues, which featured the utilization of tert-butyldimethylsilyl (TBDMS) group as the directing group in the 1,3-dipolar cycloaddition reactions. 4-tert-Butyldimethylsilyl-5-trifluoromethyl-1,2,3-triazole nucleoside analogues were generated as the only cycloaddition products in moderate yields (15–79%) via the treatment of glycosyl azides with 3,3,3trifluoro-1-tert-butyldimethylsilylpropyne **1** in toluene at 85 °C. Removal of TBS groups in these triazole cycloadducts with tetrabutylammonium fluoride (TBAF) smoothly afforded the various 5-trifluoromethyl-1,5-disubstituted 1,2,3-triazole nucleoside analogues in good yields (40–88%).

© 2010 Elsevier B.V. All rights reserved.

#### 1. Introduction

Naturally occurring nucleosides and their synthetic analogues have been the intensive research interest due to their highly potential biological activity as antitumor and antiviral agents [1]. Among them nucleosides bearing a five-member ring nucleobase are of great interest due the fact that they have exhibited unique biological activities [2]. In the past years, many highly bioactive triazole nucleosides have been synthesized and biologically evaluated for virus and hepatitis therapies [3–6]. For example, 1,2,3-triazole TSAO analogue displayed very strong bioactivity for inhibition of HIV-1 in CEM and MT-4 cells [4a] and 1,2,4-triazole ribavirin derivative has been reported to be very potent for treatment of drug-resistant pancreatic cancer [6a] (Fig. 1).

On the other hand, introduction of a trifluoromethyl group into some nucleosides has drawn considerable attention in view of the remarkable changes in the bioactivity and stability of the corresponding compounds [7]. Thus, development of methodology for synthesizing trifluoromethylated triazolo nucleoside analogues has been of interest of organic chemists, medicinal chemists and pharmacologists. So far, the most versatile method to access triazole-based nucleoside analogues has been the Huisgen 1,3dipolar cycloaddition of glycosyl azides with alkynes [8]. This classic thermal cycloaddition, however, usually suffers from the formation of regioisomeric mixture of products when azides were treated with unsymmetrical acetylene [9]. Although the regioselective synthesis of the 4-trifluoromethyl-substituted 1,2,3-triazoles has been made by several groups [10], to the best of our knowledge the regioselective incorporation of a CF<sub>3</sub> group into 5position of 1,2,3-triazolo moiety of nucleosides has never been explored [11]. Indeed, the strong electron-withdrawing property of trifluoromethyl group makes it more accessible to 4-position of triazole ring. Recently, Hlasta et al. [12] reported the use of trimethylsilyl group as regiodirecting group to control the regioselectivity in synthesizing 1,5-disubstituted 4-trimethylsilyl 1,2,3-triazoles via the cycloaddition reaction of 1-trimethylsilylacetylenes and organoazides. Inspired by Hlasta's work, we were interested to investigate the viability of utilizing the trifluoromethylated acetylene as an efficient building block to regioselective preparation of 5-trifluoromethylated 1,2,3-triazoles. Herein described was our methodology of practically constructing 5trifluoromethyl 1,5-disubstituted 1,2,3-triazole nucleoside analogues via tert-butyldimethylsilyl(TBS)-directed 1,3-dipolar cycloaddition reaction.

#### 2. Results and discussion

We first prepared the 3,3,3-trifluoro-1-tert-butyldimethylsilylpropyne **1** as the dipolarophile for cycloaddition reaction. According to a modified procedure reported by Hanamoto and

<sup>\*</sup> Corresponding author at: Shanghai Inst. of Organic Chemistry, Lab. of Organofluorine Chemistry, Chinese Academy of Sciences, 345 Lingling Road, Shanghai 200032, China. Tel.: +86 21 54925187; fax: +86 21 64166128.

E-mail address: flq@mail.sioc.ac.cn (F.-L. Qing).

<sup>0022-1139/\$ –</sup> see front matter © 2010 Elsevier B.V. All rights reserved. doi:10.1016/j.jfluchem.2010.12.012



Fig. 1. Highly bioactive triazolo nucleoside analogues.



Scheme 1. Synthesis of 1 from 2-bromo-3,3,3-trifluoropropene.

Yamada [13], **1** could be provided by addition of commercially available 2-bromo-3,3,3-trifluoropropene to a solution of in situ generated lithium hexamethyldisilazide (LiHMDS) in THF using hexamethylphosphoramide (HMPA) as additive, followed by addition of tert-butyldimethylsilyl chloride(TBDMSCI). <sup>19</sup>F NMR spectroscopy of the reaction mixture illustrated that the reaction proceeded smoothly. However, it was unexpectedly difficult to isolate **1** from other components by fractional distillation due to their close boiling points. <sup>1</sup>H, <sup>13</sup>C and <sup>19</sup>F NMR showed that **1** was contaminated by the hexamethyldisilazane (HMDS) and other unknown compounds, which were also confirmed by the GC–MS analysis (Scheme 1) [14].

Although the purification of the compound **1** was unsuccessful, we believed that the cycloaddition between organic azides and the crude compound **1** should be chemospecific, thus purification of **1** was not necessary [15]. To investigate the reactivity of **1** for 1,3-dipolar cycloaddition, readily available benzyl azide **2** was chosen as the model substrate. We were pleased to find that the cycloaddition, when performed in toluene at 85 °C for 30 h, provided the desired triazole cycloadduct **3** as the single regioisomer product (Scheme 2). The regiochemistry assignment of the compound **3** was determined by HMBC experiment [16]. Increasing reaction temperature from 85 °C to reflux temperature or performing the reaction in THF provided the product in very low yield along with some complicated byproducts.

The promising result obtained from the cycloaddition of benzyl azide **2** with **1** stimulated us to extend the reaction to glycosyl azide substrates. Thus, a series of O-protected glycosyl azides **6a–6c** [17], **6d** [18], **6e–6g** [19] and **6h** [20] were prepared according to the reported procedures and used for cycloaddition reaction. It should be pointed out that for non-participating benzyl group protected glycosyl azide **6c**, anomeric mixture ( $\alpha/\beta$  1:1) was obtained. O-Acyl glycosyl azide **6e** was also contaminated by about 10% of 1,2-cis anomer [19b]. <sup>n–</sup>Octyl azide **4** was used as a control and the results were summarized in Table 1. We were pleased to find that all the glycosyl azides provided the desired cycloadducts



Fig. 2. ORTEP drawing of the X-ray crystallographic structure of 7a.

in moderate yields (Table 1, entries 2-9), albeit lower yields compared to **4** (Table 1, entry 1). Of all the O-protected glycosyl azides, glycopyranosyl azides **6a** and **6b** (Table 1, entries 2 and 3) were more reactive in the cycloaddition reaction than glycofuranosyl azides 6e and 6f (Table 1, entries 6-8), although higher potential bioactivity should be expected from the latter ones. The lower reactivity of **6e** and **6f** was ascribed to the larger steric hindrance of furanose ring in the transition state of cycloaddition than that of their pyranose counterparts. Additionally, it was found that all the O-acyl glycosyl azides reacted smoothly to afford the corresponding cycloadducts with retention of configuration at the anomeric carbon (Table 1, entries 2 and 3 and entries 5-8). Lower yield (37%) was resulted for O-benzyl glucosyl azide 6c (anomeric mixture 1:1) to afford the corresponding mixture **7c** of two  $\alpha/\beta$ anomers (Table 1, entry 4) which may due to more crowded steric environment of azido group caused by benzyl group in 6c than that caused by acetyl group in 6a or 6b [21]. It should be noted that cycloaddition reaction also proceeded well for the azides 6d and 6h, where azido groups were not attached to anomeric carbon (Table 1, entries 5 and 9), and the corresponding 1,2,3-triazole nucleosides 7d and 7h were provided in 70% and 54% yield, respectively. Most importantly, all the cycloaddition reactions exhibited very excellent regioselectivity and exclusively delivered the 5-trifluoromethyl-1,4,5-trisubstituted 1,2,3-triazoles. The absolute structure of compound 7a was confirmed by X-ray diffraction analysis (Fig. 2) [22]. The excellent regioselectivities undoubtedly illustrated that the transition state of herein described 1,3-dipolar cycloaddition was mainly directed by TBS



Scheme 2. Model cycloaddition reaction between 1 and benzyl azide 2.

Download English Version:

## https://daneshyari.com/en/article/10572694

Download Persian Version:

https://daneshyari.com/article/10572694

Daneshyari.com