

## Paper derived biomorphic porous titanium carbide and titanium oxide ceramics produced by chemical vapor infiltration and reaction (CVI-R)

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### Abstract

Chemical vapor infiltration (CVI) is used for producing biomorphic porous TiC ceramics derived from paper. The paper samples are first carbonized in inert atmosphere to yield biocarbon template structures ( $C_b$ -template). Subsequently, three routes for converting the  $C_b$ -templates into TiC ceramics are studied. The first route includes CVI with  $TiCl_4-H_2$ . The effect of methane as additional carbon source is investigated on the second route ( $TiCl_4-H_2-CH_4$ ). Finally, a two step CVI process (Route 3), first  $TiCl_4-H_2$  and subsequent  $TiCl_4-H_2-CH_4$ , is performed in order to improve both the grade of conversion of the  $C_b$ -template into TiC and the mechanical properties of the resulting porous TiC ceramics. Furthermore, porous  $TiO_2$  ceramics are produced by high temperature oxidation of the TiC ceramics in air flow.

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### 1. Introduction

Biomorphic cellular ceramics are a new class of ceramic materials with natural designed porous structure having recently attained particular interest. They are obtained by carbonization and consecutive conversion of biological preforms into ceramics. During the conversion their microstructure is neither destroyed nor altered, only a homogeneous shrinkage in all directions takes place during carbonization.<sup>1,2</sup> Various biotemplate processing technologies are developed for manufacturing of biomorphic SiC-based ceramics. Among these methods the most investigated are the Si liquid infiltration,<sup>3–5</sup> the Si gas infiltration,<sup>5–7</sup> the SiO vapor infiltration,<sup>6,8</sup> the polymer infiltration<sup>9</sup> and, the method used in this paper, the CVI-R method.<sup>10–13</sup> Compared to the other infiltration methods the chemical vapor infiltration (CVI) has the advantage of using relatively low processing temperatures, retaining the initial structure of the biotemplate on a micrometer level.

Biomorphic SiC porous ceramics could be applied as high temperature filters or catalytic support structures due to their high thermal conductivity, good oxidation and corrosion resistance as well as high strength at elevated temperatures.

The properties of porous TiC ceramics are inferior to those of SiC, however, their improved corrosion resistance in phosphoric acid, the high electrical conductivity as well as the very good wet ability by metal melts made these materials interesting candidates for specific applications such as ceramic-metal composites<sup>14</sup> or catalyst supports for chemical and biochemical reactions.<sup>15,16</sup>  $TiO_2$  ceramics are used as catalysts in photochemical and in biochemical reactions, for example in the purification of water.<sup>17,18</sup>

There exist only few reports about production of porous TiC and  $TiO_2$  ceramics. The infiltration of dried wood or charcoal with titanium tetra-isopropoxide, followed by high temperature treatment yields highly porous  $TiO_2$  ceramics.<sup>19,20</sup> A porous TiC ceramic is prepared by liquid vacuum infiltration of wood derived carbon structures with tetrabutyl-titanat.<sup>21</sup> First, the tetrabutyl-titanat decomposes to  $TiO_2$ , and then reacts with the carbon biotemplate at temperatures of about 1400 °C forming TiC.<sup>21</sup> In<sup>22</sup> our first results about the processing of TiC porous ceramic from wood by chemical vapor infiltration are reported.

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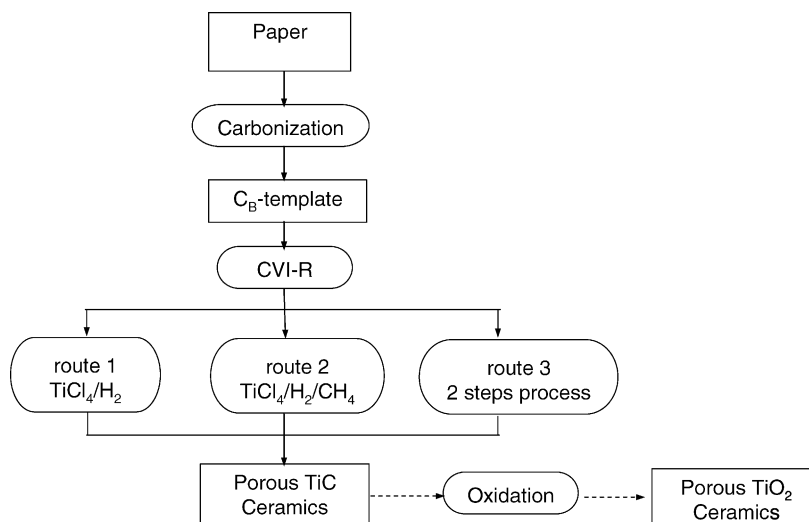


Fig. 1. Flow chart of the CVI-R ceramization process with three different routes.

In this study, chemical vapor infiltration-and reaction (CVI-R) process with  $\text{TiCl}_4\text{-H}_2$  and  $\text{TiCl}_4\text{-H}_2\text{-CH}_4$  systems is investigated to produce biomorphic cellular TiC ceramics by converting carbon fibers from paper biotemplates into TiC. The processing scheme is presented in Fig. 1. Three routes for converting the carbon biotemplates ( $\text{C}_b$ ) into TiC ceramics are investigated. The first route includes chemical vapor infiltration with  $\text{TiCl}_4\text{-H}_2$ . The effect of methane as additional carbon source is investigated on Route 2. Finally, on the third route, two step CVI process is performed with  $\text{TiCl}_4\text{-H}_2$  and subsequently with the  $\text{TiCl}_4\text{-H}_2\text{-CH}_4$  system. The grade of conversion of the  $\text{C}_b$ -template into TiC as well as the mechanical properties of the resulting porous TiC ceramics should be improved with the two step process. Furthermore, first results on the processing of porous  $\text{TiO}_2$  ceramics by oxidation of the TiC ceramics in air flow are presented.

## 2. Experimental procedure

The substrates used in this study are flat paper preforms of 0.80 mm thickness with a geometrical density of  $0.22 \text{ g/cm}^3$ . The initial porosity of the papers is 82% with a mean pore size of  $25 \mu\text{m}$ . The papers are cut to  $40 \text{ mm} \times 40 \text{ mm}$  squares and carbonized at the conditions described below.

The carbonization of the paper performs and the chemical vapor infiltration of the resulting carbon biotemplates is performed in a horizontal hot-wall tubular flow reactor operated at atmospheric pressure.

The paper is carbonized in inert atmosphere ( $\text{He}$ ,  $5 \text{ cm/s}$ ) at the following conditions:  $1 \text{ K/min}$  ramp up to  $350^\circ\text{C}$ ,  $1 \text{ h}$  dwell time, followed by  $2 \text{ K/min}$  ramp up to  $850^\circ\text{C}$  and additional also  $1 \text{ h}$  dwell time.

$\text{TiCl}_4$  is used as a titanium source for converting  $\text{C}_b$  into TiC. It is vaporized in a bubbler and carried into the reactor

by a carrier gas  $\text{H}_2$  or  $\text{He}$ , dosed by mass flow controllers. The gaseous precursors diffuse into the pores of the carbon biotemplate, where reductive decomposition of  $\text{TiCl}_4$  in excess of hydrogen and chemical reaction to TiC takes place simultaneously. The effect of the process parameters such as temperature ( $1000\text{--}1200^\circ\text{C}$ ), gas flow velocity ( $5\text{--}18 \text{ cm/s}$ ), molar fraction of the precursor ( $x_{\text{TiCl}_4} = 0.01\text{--}0.06$ ) and  $\text{H}_2/\text{TiCl}_4$  molar ratio ( $7\text{--}55$ ) on the mass gain and the porosity of the samples after infiltration is investigated. Three pieces of carbonized paper are infiltrated and converted into TiC ceramics in the isothermal zone of the reactor for each experiment.

Titanium oxide porous ceramics are produced by oxidation of TiC ceramics at  $850^\circ\text{C}$  for  $6 \text{ h}$  in air flow ( $10 \text{ cm/s}$ ) in the same equipment.

The composition of the porous ceramics is determined by XRD and EDX. The X-ray diffraction spectroscopy (Phillips PW 3040) is performed with rotation of the samples and  $\text{Cu K}\alpha$  radiation in angle range  $2\theta = 5\text{--}85^\circ$ . The morphology of the ceramics is investigated by Scanning Electron Microscopy coupled with an Energy Dispersive X-ray Analysis (SEM/EDX, JSM-6400) to determine the element distribution across the infiltrated layer. The porosity of the ceramics is determined by Hg-porosimetry (Carlo Erba Mercury Intrusion Porosimeter 2000).

According to German Standard Code DIN 52 292, coaxial double ring bending test (INSTRON Model 4204) is carried out to measure the bending strength of the ceramic specimen. A scheme of the experimental arrangement is presented in Fig. 2.

The bending strength  $\sigma_b$  of the specimen is related to the maximum load force  $F_{\text{max}}$  before crushing and the sample thickness  $s$ . For rectangular specimen, bending strength  $\sigma_{b,\text{max}}$  is given by:

$$\sigma_{b,\text{max}} = 1.04 \frac{F_{\text{max}}}{s^2}.$$

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