



# Nitrate deposition following an astrophysical ionizing radiation event

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## Abstract

It is known that a gamma ray burst (GRB) originating near the Earth could be devastating to life. The mechanism of ozone depletion and subsequent increased UVB exposure is the primary risk, but models also show increased nitrification culminating in nitric acid rain-out. These effects are also expected after nearby supernovae and extreme solar proton events. In this work we considered specifically whether the increased nitric acid rainout from such events is a threat to modern terrestrial ecosystems. We also considered its potential benefit to early terrestrial Paleozoic ecosystems. We used established critical loads for nitrogen deposition in ecoregions of Europe and the US and compared them with previously predicted values of nitric acid rainout from a typical GRB within our galaxy. The predicted rainout was found to be too low to harm modern ecosystems, however, it is large compared with probable nitrate flux onto land prior to the invasion of plants. We suggest that this flux may have contributed nutrients to this invasion if, as hypothesized, the end-Ordovician extinction event were initiated by a GRB or other ionizing radiation event.

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## 1. Introduction

Bursts of ionizing radiation episodically alter the Earth's atmosphere. Sources include gamma ray bursts (GRBs), supernovae, and probably the Sun (Melott and Thomas, 2011; Thomas et al., 2013), although recent claims of numerous recent supernovae are probably incorrect (Melott et al., 2015). Astrophysical ionizing radiation can provide sufficient energy to break the normally stable N<sub>2</sub> triple bond which facilitates the formation of so-called "odd nitrogen" compounds (NO<sub>x</sub>). NO<sub>x</sub> can, among other things, react with ozone, catalyzing its conversion back to molecular oxygen. As stratospheric ozone is consumed, additional solar ultraviolet-B (UVB) radiation passes to the surface resulting in increased mortality of exposed organisms, particularly phytoplankton despite their ability

to avoid and repair UV damage (Melott and Thomas, 2011). This process has been hypothesized as a link to mass extinctions, particularly the end-Ordovician extinction, which fits some of the patterns expected for such an event (Melott et al., 2004; Melott and Thomas, 2009). A GRB of sufficient luminosity is expected, based on rates, to have occurred multiple times during the history of life on Earth. However, due to very different atmospheric chemistry and the possible lack of a UVB shield (Melott et al., 2005; Piran and Jimenez, 2014), the ozone depletion mechanism is probably largely irrelevant prior to the advent of free oxygen.

The presence of NO<sub>x</sub> in the atmosphere also eventually leads to increased nitrate rainout in the form of nitric acid (HNO<sub>3</sub>) (Thomas et al., 2005). This has both the potential to harm or benefit life. Life requires that nitrogen be oxidized or reduced from its gaseous form (N<sub>2</sub>), called nitrogen fixation, before it is available for biologic use. Prior to the Haber–Bosch ammonia manufacturing process

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which began in the early 1900's most nitrogen fixation was done by biologic sources in the form of nitrogenase-containing bacteria and archaea, with much lower contributions from abiotic sources which include lightning, volcanism and bolide impacts (Schlesinger, 1997; Galloway et al., 2004). Astrophysical ionizing radiation events produce intermittent doses of nitrate flux in addition to normal biotic and abiotic sources, as shown specifically for a GRB by Thomas et al. (2005). Although nitrogen fixation is necessary for life, it is known that excess nutrient deposition, or eutrophication, can be harmful (Galloway et al., 2004). Typically, eutrophication from nitrogen deposition is caused by anthropogenic sources such as fertilizer and fossil fuel combustion, however, the increased nitrate from ionizing radiation could also cause a eutrophic response. We addressed specifically whether the increased nitrate rainout from an intense event like a nearby GRB could be a threat to modern terrestrial ecosystems. In addition, given the likelihood that an event happened in the oxygenated but nitrogen limited past, it could have been an important timely source of biologically available nitrogen in the late-Ordovician.

## 2. GRB nitrate prediction

A nearby GRB represents an extreme case of ionizing radiation making it a good example to use for nitrate producing potential. The Goddard Space Flight Center (GSFC) 2-D atmospheric model was used in previous work to predict the atmospheric effects, including nitrate rainout, of a typical GRB within 2 kpc of the earth (Melott et al., 2005; Melott and Thomas, 2009). It considers the atmosphere in two dimensions, latitude and altitude, and varies the angle of incidence of the burst as well as the time of year. It is based on the modern atmosphere, but since the oxygen fraction has generally been  $20 \pm 10\%$  during the Phanerozoic (Berner et al., 2003; Berner, 2009), it can be used as a semi-quantitative guide to what is expected over this period—with a possible exception during the excursion to very high oxygen levels during the Permo-Carboniferous. For this work we used the highest predicted nitrate rainout rates which were for a September burst striking the earth at  $90^\circ$  N latitude. The burst intensity is one which might be expected to induce a mass extinction, and to occur from a GRB or supernova approximately every 200 Myr (Melott et al., 2005; Piran and Jimenez, 2014; Melott and Thomas, 2011). An important feature of the prediction is that nitrate rainout had a 12 month periodicity and occurred almost exclusively between  $30^\circ$  N and  $60^\circ$  N latitude, making the maximum rainout values correlate well geographically with critical load data compiled for Europe and the US.

The peak month for GRB sourced nitrate rainout rate between  $30^\circ$  N and  $60^\circ$  N latitude occurs 20 months following the burst and is reported in kg of nitrate as  $3 \times 10^{-12} \text{ kg}_{\text{NO}_3} \text{ m}^{-2} \text{ s}^{-1}$  (Melott et al., 2005). Since the rainout during winter months is much less

( $1 \times 10^{-12} \text{ kg}_{\text{NO}_3} \text{ m}^{-2} \text{ s}^{-1}$ ), we used a conservative estimate for the average monthly value for the year of  $2 \times 10^{-12} \text{ kg}_{\text{NO}_3} \text{ m}^{-2} \text{ s}^{-1}$ ; which in terms of kg of nitrogen per hectare is  $1 \times 10^{-1} \text{ kg}_N \text{ ha}^{-1} \text{ yr}^{-1}$ . This localized concentration represents the highest potential value of additional terrestrial nitrogen deposition after an ionizing event. Note that these levels are insignificant for the vast reservoir of the oceans, which in any case have been fixing nitrogen since long before the time we are considering (Raven and Yin, 1998; Navarro-González et al., 2001).

## 3. Critical loads

While Thomas and Honeyman already showed that the nitrate rainout is too small to be harmful to amphibians, it is important to establish more generally whether increased nitrate from a GRB would be harmful (Thomas and Honeyman, 2008). While the complexities involved prevent a general predictive model for the evolution of an ecosystem under nitrate addition, the scientific interest in monitoring modern anthropogenic sources of nitrogen deposition, primarily industrial pollution and agricultural inputs, has produced a wealth of data to evaluate its impact. The concept of a critical load is one way that has been developed to systematize the large amount of resulting scientific data. It is defined as “a quantitative estimate of an exposure to one or more pollutants below which significant harmful effects on specified sensitive elements of the environment do not occur according to present knowledge” (Nilsson and Grennfelt, 1988). The critical loads established for total nitrogen deposition include both oxidized and reduced forms, notably nitrate and ammonium which are the primary counter ions in fertilizer. However, the distinction is not typically made in critical load studies since both forms are bioavailable and readily interconverted, therefore, they often give an equivalent eutrophic response. However, their effect on soil pH may differ making it also necessary to evaluate critical loads for acidification.

The use of critical loads to quantify the deposition of pollutants began in the 1980s and has been used extensively in Europe and Canada, and increasingly in the US. The idea is to set a threshold level of input below which there is no known harm on an ecosystem. The metrics used to evaluate harm from eutrophication (excess nutrient deposition) are changes in species composition (which includes decrease in species diversity), plant development, and/or biogeochemistry. In general, the first and most sensitive changes occur with species composition where those adapted for nitrogen starved environments give way to more nitrophilic species, while other changes can include increased mineralization, increased nitrification, nitrate leaching, increased species susceptibility to stress, altered growth patterns, increased N uptake, increased mortality, and increased growth. A synthesis of data for the effects of nitrogen deposition are detailed in Pardo et al. (2011), where the range of compiled critical loads for inland US

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