# **ARTICLE IN PRESS**

[Biochimica et Biophysica Acta xxx \(2014\) xxx](http://dx.doi.org/10.1016/j.bbabio.2014.06.003)–xxx



Contents lists available at ScienceDirect

## Biochimica et Biophysica Acta



journal homepage: www.elsevier.com/locate/bbabio

## Dynamics of the active site architecture in plant-type ferredoxin-NADP<sup>+</sup> <sup>2</sup> reductase catalytic complexes

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### 8 ARTICLE INFO ABSTRACT

9 Article history: 10 Received 24 March 2014 11 Received in revised form 5 June 2014

12 Accepted 11 June 2014

13 Available online xxxx

 Keywords: 15 Ferredoxin-NADP<sup>+</sup> reductase Flavoenzyme Kinetic isotope effect Hydride transfer Charge–transfer complex

20 Plastidic-type FNR<br>21 Bacterial-type FPR Bacterial-type FPR

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ano a. Kinetic isotope effects in reactions involving hydride transfer and their temperature dependence are powerful 22 tools to explore dynamics of enzyme catalytic sites. In plant-type ferredoxin-NADP+ reductases the FAD cofactor  $23$ exchanges a hydride with the NADP(H) coenzyme. Rates for these processes are considerably faster for the 24 plastidic members (FNR) of the family than for those belonging to the bacterial class (FPR). Hydride transfer 25 (HT) and deuteride transfer (DT) rates for the NADP<sup>+</sup> coenzyme reduction of four plant-type FNRs (two repre- $26$ sentatives of the plastidic type FNRs and the other two from the bacterial class), and their temperature depen- 27 dences are here examined applying a full tunnelling model with coupled environmental fluctuations. 28 Parameters for the two plastidic FNRs confirm a tunnelling reaction with active dynamics contributions, but iso- 29 tope effects on Arrhenius factors indicate a larger contribution for donor–acceptor distance (DAD) dynamics in 30 the Pisum sativum FNR reaction than in the Anabaena FNR reaction. On the other hand, parameters for bacterial 31 O5 FPRs are consistent with passive environmental reorganisation movements dominating the HT coordinate and 32 no contribution of DAD sampling or gating fluctuations. This indicates that active sites of FPRs are more organised 33 and rigid than those of FNRs. These differences must be due to adaptation of the active sites and catalytic mech- 34 anisms to fulfil their particular metabolic roles, establishing a compromise between protein flexibility and func- 35 tional optimisation. Analysis of site-directed mutants in plastidic enzymes additionally indicates the requirement 36 of a minimal optimal architecture in the catalytic complex to provide a favourable gating contribution. 37 38 © 2014 Published by Elsevier B.V.

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**1. Introduction** 43

 $\overline{Abbreviations}$ : FNR, ferredoxin-NADP<sup>+</sup> reductase; FPR, bacterial-type FNR; AnFNR, FNR from the cyanobacterium Anabaena PCC 7119; PsFNR, FNR from Pisum sativum; XaFPR, FPR from Xanthomonas axonopodis pv. citri; EcFPR, FPR from Escherichia coli; FNRox, FNR in the fully oxidised state; FNRhq, FNR in the anionic hydroquinone (fully reduced) state; HT, hydride transfer; DT, deuteride transfer; WT, wild-type; CTC, charge–transfer complex; CTC-1, FNR<sub>ox</sub>:NADPH CTC; CTC-2, FNR<sub>hq</sub>:NADP<sup>+</sup> CTC; 2'-P-AMP, 2′-P-AMP moiety of NADP(H); N5-FAD, N5 hydride donor/acceptor of the FADH−/ FAD isoalloxazine ring of FNR; C4-NADP(H), C4 hydride acceptor/donor of the NADP<sup>+</sup>/ NADPH nicotinamide ring; NADPD, (4R)-4-<sup>2</sup>H-NADPH;  $k_{A \rightarrow B}$ ,  $k_{B \rightarrow C}$ , apparent/observed rate constants obtained by global analysis of spectral kinetic data;  $k_{\text{HT}}$ ,  $k_{\text{DT}}$ ,  $k_{\text{obsFT}}$ ,  $k_{\text{obsDT}}$ , limiting hydride and deuteride transfer first-order rate constants for the reduction of FNR and their corresponding observed values under particular conditions; KIE, kinetic isotope effect on rate constants;  $A_H$ ,  $A_D$ , Arrhenius pre-exponential factors for hydride and deuteride, respectively;  $E_{\text{aH}}$ ,  $E_{\text{aD}}$ , activation energies for hydride transfer and deuteride transfer, respectively; DAD, donor–acceptor distance

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Plant-type ferredoxin-NADP<sup>+</sup> reductases (FNRs) have evolved from  $44$ a common ancestor into two classes: plastidic FNRs, which are found in 45 plastids of plants, algae and cyanobacteria, and bacterial FNRs, herein 46 known as FPRs [1,2]. The main role of plastidic FNR is the photosynthetic 47 transfer of reduction equivalents from ferredoxin (Fd) to NADP<sup>+</sup> via its  $48$ non-covalently bound FAD redox cofactor, although in vivo this reaction 49 is reversible and FNR can provide electrons to different electron carrier 50 proteins using NADPH as reductant [3]. The bacterial counterparts catal- 51 yse the non-photosynthetic reaction providing reducing power to de- 52 toxification and nitrogen fixation processes [\[1,2,4,5\].](#page--1-0) Differences in the 53 requirements of their biological functions are reflected in turnover 54 rates, substrate affinity and specificity, and, therefore, catalytic efficien- 55 cy, as a result of the divergence achieved along functional specialisation. 56 The lower catalytic efficiency of FPRs is related to structural differences 57 in the isoalloxazine active site environment: i) the FAD acquires an ex- 58 tended conformation in plastidic FNRs but it is folded in FPRs and, ii) the 59 isoalloxazine stacks between two Tyr residues (Tyr79 and the C- 60 terminal Tyr303 in Anabaena (AnFNR)) in plastidic FNRs, while FPRs 61

Please cite this article as: A. Sánchez-Azqueta, et al., Dynamics of the active site architecture in plant-type ferredoxin-NADP<sup>+</sup> reductase catalytic complexes, Biochim. Biophys. Acta (2014), <http://dx.doi.org/10.1016/j.bbabio.2014.06.003>

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 have a C-terminal extension where the residue facing the isoalloxazine 63 ring moiety is not the C-terminal  $[4,6–8]$  $[4,6–8]$  (Fig. 1). The length and se- quence of this extension are not homogeneous among FPRs. Subclass I, including Rhodobacter capsulatus (RcFPR) and Xanthomonas axonopodis pv. citri (XaFPR) enzymes among others, has an Ala instead of the plastidic C-terminal Tyr followed by a Phe and up to 5 additional resi- dues [9–[12\].](#page--1-0) Members of subclass II maintain a Tyr stacking the re- face of the isoalloxazine and it is followed by a Trp, as in Escherichia coli FPR (EcFPR) [13–[15\].](#page--1-0) In both subclasses, the abovementioned Phe or Trp residue located at the C-terminal extension stacks on the adeno- sine moiety of FAD, apparently contributing to its folded conformation. In plastidic FNRs the hydride transfer (HT) takes place between the N5 atom of the FAD(H−) isoalloxazine (N5-FAD) and the C4 atom of the nicotinamide ring of NADP(H) (C4-NADP(H)) [16,17]. Mutational and theoretical analysis focused on the displacement of the Tyr stacking on the re-face of the isoalloxazine, required to allow the approaching of the reacting atoms during HT, and have contributed to understand mechanistic details of this process as well as the role of the Tyr itself [17–[26\]](#page--1-0). Though the presence of this Tyr is not obligatory for HT, it is crucial for the high catalytic efficiency of FNRs. It modulates the FAD midpoint reduction potential, avoids a too strong interaction between the reacting rings that would be incompatible with product release, and contributes to the optimal geometry between the reacting atoms for HT, N5-FAD and C4-NADP(H). Finally, it provides the active site

86 with the required flexibility to allow the HT step occurring through

tunnelling [\[21,22\].](#page--1-0) It is accepted that in bacterial FPRs the HT also 87 takes place between C4-NADP(H) and N5-FAD. However, differences 88 in the side-chain stacking against the isoalloxazine and, particularly, 89 the presence of the C-terminal extension suggest a more complex 90 mechanism to attain the catalytically competent interaction (Fig. 1) 91 [\[6,7,11,15,27\].](#page--1-0) So far, no details about structural arrangement and dy- 92 namics of the active site during catalysis in FPRs have been provided. 93

The importance of slow (ms to s) protein flexibility in substrate rec- 94 ognition and allosterism is widely accepted [\[28\].](#page--1-0) Faster (fs to ps) mo- 95 tions coupled to the chemical step have also been pointed as 96 contributing to active site dynamics in enzyme catalysed reactions. Var- 97 iations of the Eyring's Transition State Theory have been used to treat 98 enzyme-catalysed reactions, including fast electron–proton coupled 99 transfers [29–39]. The most recent approaches postulate HT processes 100 as fully quantum-mechanical events modulated by dynamical motions 101 of the active site environment within the "environmentally coupled 102 full tunnelling model" that describes two types of protein motions puta- 103 tively linked to catalysis: pre-organisation and reorganisation [\[40\]](#page--1-0). Pre- 104 organisation motions are assumed to occur prior to the HT event, in 105 the ps to ns time scale, and involve large regions of the protein. 106 Reorganisation motions involve heavy atoms within the active site, 107 and constitute fast (ps to fs) nuclear fluctuations. Despite evidences 108 for a role of protein dynamics in accelerating HT reactions [41–[44\],](#page--1-0) 109 their contribution to catalysis is still on debate [\[45\]](#page--1-0). Studies on 110 dihydrofolate reductases and on some pyridine-nucleotide dependent 111



Fig. 1. The FAD environment in the plant-type FNR family. (A) Comparison of the FAD folding and environment in the crystal structures of AnFNR (PDB 1que, blue), XaFPR (PDB 4b4d, wine red) and EcFPR (PDB 1fdr, yellow). (B) Theoretical model of the allocation of the nicotinamide moiety of NADP<sup>+</sup> in the active site of AnFNR<sub>hq</sub> as obtained by MD simulations [\[23\]](#page--1-0). Protein chain is shown in green, FAD in orange and NADPH in blue. Detail of the active site configuration in (C) AnFNR, (D) XaFPR and (E) EcFPR. Key residues are shown as CPK coloured sticks.

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