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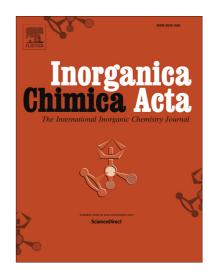
PII: S0020-1693(18)30875-2

DOI: https://doi.org/10.1016/j.ica.2018.08.029

Reference: ICA 18429

To appear in: Inorganica Chimica Acta

Received Date: 6 June 2018
Revised Date: 20 July 2018
Accepted Date: 20 August 2018



Please cite this article as: D. Lomjanský, C. Rajnák, J. Titiš, J. Moncoľ, L. Smolko, R. Boča, Impact of tetrahedral and square planar geometry of Ni(II) complexes with (pseudo)halide ligands to magnetic properties, *Inorganica Chimica Acta* (2018), doi: https://doi.org/10.1016/j.ica.2018.08.029

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Impact of tetrahedral and square planar geometry of Ni(II) complexes with (pseudo)halide ligands to magnetic properties

Dominik Lomjanský^a, Cyril Rajnák^{a*}, Ján Titiš^{a*}, Ján Moncoľ^b, Lukáš Smolko^c, and Roman Boča^a

Abstract

Four tetracoordinate Ni(II) complexes have been prepared, structurally characterized, and subjected to magnetometric studies. The complexes $[Ni(PPh_3)_2(NCS)_2]$, $[Ni(dppp)(NCS)_2]$, and $[Ni(dppm)Br_2]$ are planar and thus diamagnetic. The complex $[Ni(biqu)Br_2]$ is quasi-tetrahedral, with the geometry close to C_{2v} symmetry, and paramagnetic. While on one side it resembles a prolate bisphenoid (the angle N-Ni-N = 83 deg), on the second side it mimics an oblate bisphenoid (Br-Ni-Br = 126 deg). It exhibits a zero-field splitting of the ground term 3A_2 into three crystal-field multiplets that can be described by D and E parameters within the spin Hamiltonian formalism. The ab initio calculations confirm this interpretation; however, the evaluation of the spin-Hamiltonian parameters meets difficulties owing to the quasi-degeneracy of the electronic terms.

Key words

Ni(II) complexes, P-donor ligands, N-donor ligands, (pseudo)halide ligands, paramagnetism, diamagnetism.

1. Introduction

Magnetic anisotropy of transition metal complexes attracts much attention in the recent period because this is a key factor that induces the slow magnetic relaxation and consequently a behavior of the metal complexes as single molecule (single ion) magnets – SMMs, SIMs. The SIM behavior has been observed among several first-row transition metal complexes such as V(IV), Cr(III), low-spin Mn(IV), Mn(III), Fe(III), Fe(II), Fe(I), Co(II), Ni(II), Ni(I), and Cu(II) [1-11]. Though some families are extremely rich, such as tri-, tetra-, penta-, hexa-, hepta- and octacoordinate Co(II) complexes, some are represented only by a few or a single example [12, 13]. The class of Ni(II) SIMs is limited only to three compounds so far of which $[Ni(pydca)(dmpy)] \cdot H_2O$ and $[Ni(NCS)_2(nqu)_2(H_2O)_2] \cdot 2nqu$ are hexacoordinate, and $[Ni(mdabco)_2Cl_3]ClO_4$ is pentacoordinate [14, 15]. All of them possess negative axial zero-field splitting parameter D < O(-13.7, -5.9, and -311 cm⁻¹) as a prerequisite of the barrier to spin reversal that facilitates the SIM behavior. The class of hexacoordinate Ni(II) complexes is a widely studied and well understood case where the magnetic anisotropy parameter can be predicted and tuned on the basis of the magnetostructural D-correlation [16].

The class of tetracoordinate Ni(II) complexes is studied in a lower extent. There is one obstacle: owing to the Jahn-Teller effect the ground electronic term 3T_1 in the tetrahedral geometry is unstable and the geometry is stabilized along the e-mode of distortion leading either to the 3A_2 or 3E state of the D_{2d} geometry. The latter is also unstable and the distortion continues to the stable geometry of the C_{2v} symmetry (or lower) with 3A_1 , 3B_1 , 3B_2 terms. Another obstacle originates in weaker crystal-field strength of the tetrahedral systems when compared to the octahedral ones: the quasi-degeneracy is a common feature since the energy gap Δ_{ax} between the ground and the first excited terms is rather low. This causes that the evaluation of the traditional spin-Hamiltonian parameters (D and E) could violate because the perturbation theory tends to diverge.

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