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Deng-Hui Liu, Cheng-Peng Yao, Ming Yang, Heng-Jiang Zhu

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ACCEPTED MANUSCRIPT The electron properties of infinite length single-walled silicon nanotubes are studied by density functional theory

Deng-Hui Liu Cheng-Peng Yao* Ming Yang^b and Heng-Jiang Zhu^a

Abstract:

In this article, we aim to use LGTO-PBC-DFT calculation method with PBEPBE/ 6-31G(d) basis sets and PBC model to optimize geometric structure, obtain electron characteristics and band gaps of (n, m) chiral (6 $\leq n \leq 16$, $2 \leq m \leq 8$); (n, 0) zigzag ($3 \leq n \leq 15$) and (n, n) armchair $(2 \le n \le 8)$ pure infinite length single-walled Silicon nanotubes.

We have obtained that the HOCO-LUCO gaps decrease with the radii of the SWSiNTs increase by using the PBEPBE functional and 6-31G(d) basis set. most of the infinite length SWSiNTs which we studied are narrow band gaps semiconductors. perhaps the surface structure is the most important factor affecting the band gap of the nanotubes. there have been a research on the infinite silicon nanotubes and with the increase of the tube radius, an indirectdirect band gap transition has been revealed.

For all the armchair SWSiNTs, the infinite zigzag SiNTs (n,0), $(3 \le n \le 9)$ and the infinite chiral SWSiNTs (6, 2), (9, 3), (10, 5), (12, 6) are semiconductors with indirect band gaps. While for the infinite zigzag SiNTs (n, 0) 15≥n≥10, and the infinite chiral SWSiNT (12, 4) are semiconductors with direct gaps at X point, the direct gaps open at X point. It is possible that direct band gap will become potential building blocks for electronic and optoelectronic devices.

Keywords Silicon nanotubes; Density functional theory; Periodic boundary conditions; Band gap ; Semiconductor

Deng-Hui Liu, Cheng-Peng Yao* Ming Yang^b and Heng-Jiang Zhu⁶

a: College of Physics and Electronic Engineering, Xinjiang Normal University, Urumuqi, 830054, P.R. China.

b: School of materials science and engineering. Jilin University. Chanachun. P.R. China

*Correspondence E-mail: cpyao1990@yahoo.com;

Introduction

For almost thirty years, with the development of science and technology, nanomaterials¹⁻⁴ have gradually become the focus of researchers.

From graphene⁵⁻⁸, diamond⁹⁻¹¹, C₆₀ fullerene¹²⁻¹⁴to nanowires $^{15-17}$, nanobelt 18 19 , nanotubes $^{20-23}$. the successful synthesis of graphene 24 has attracted widespread attention to carbon-based nanomaterials²⁵²⁶. later, people gradually increased their attention from carbon based nanomaterials 27 28 to non-carbon nanomaterials²⁹³⁰, and gradually increased from zerodimensional nanoparticles ³¹ to one-dimensional nanotubes³²⁻³⁴ and nanowires³⁵⁻³⁸ to two-dimensional graphene³⁹⁻⁴¹ and then extended to three-dimensional graphite⁴²⁻⁴⁴.

Nano-materials with its unique physical and chemical properties in lithium batteries 45 - 48, optoelectronic nanodevices⁴⁹⁻⁵², industrial areas⁵³ and even global sustainability⁵⁴ have a lot of value. the most famous graphene is a strong material with good electrical conductivity, like rare metals (Rickard Arvidsson and Bjorn Sanden et al⁵⁵ Have studied new ways to replace rare metals with carbon nanomaterials. Rare metals are widely found in our daily necessities around us, computers, mobile phones, etc. almost all other electronic devices and around you found in the plastic, our society is highly dependent on rare metals, which ensures a high degree of pursuit of rare metals, but Because rare metals such as tin, silver, tungsten and indium are rare and difficult to extract because their concentration is very small).

In addition to two-dimensional graphene⁵⁶⁵⁷, onedimensional nanotube⁵⁸⁻⁶¹ and nanowires⁶²⁶³ have also attracted widespread attention from scientists, especially carbon nanotubes ⁶⁴ ⁶⁵ and non-carbon nanotubes ⁶⁶⁻⁶⁸, which exhibit different physical and chemical properties, the application of drugs 69, catalysis⁷⁰ and nano-electronic⁷¹ equipment and other fields⁷².

There are some articles on the theory and experiment of silicon nanotubes. J Sha et al⁷³ using nanochannel Al 2 O 3 as a template and a chemical vapor deposition (CVD) synthesize silicon nanotubes, SB Fagan et al⁷⁴ using first-principles calculations based on density functional theory, RQ Zhang et al⁷⁵ using the semiempirical molecular orbital PM3 method, M Zhang

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