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The electron properties of infinite length single-walled silicon nanotubes are studied by density functional theory

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Abstract:

In this article, we aim to use LGTO-PBC-DFT calculation method with PBEPBE/ 6–31G(d) basis sets and PBC model to optimize geometric structure, obtain electron characteristics and band gaps of (n, m) chiral (6 $\leq n \leq 16$, $2 \leq m \leq 8$); (n, 0) zigzag ($3 \leq n \leq 15$) and (n, n) armchair ($2 \leq n \leq 8$) pure infinite length single-walled Silicon nanotubes.

We have obtained that the HOCO-LUCO gaps decrease with the radii of the SWSiNTs increase by using the PBEPBE functional and 6–31G(d) basis set. most of the infinite length SWSiNTs which we studied are narrow band gaps semiconductors. perhaps the surface structure is the most important factor affecting the band gap of the nanotubes. there have been a research on the infinite silicon nanotubes and with the increase of the tube radius, an indirectdirect band gap transition has been revealed.

For all the armchair SWSiNTs, the infinite zigzag SiNTs (n,0), ($3 \le n \le 9$) and the infinite chiral SWSiNTs (6, 2), (9, 3), (10, 5), (12, 6) are semiconductors with indirect band gaps. While for the infinite zigzag SiNTs (n, 0) $15 \ge n \ge 10$, and the infinite chiral SWSiNT (12, 4) are semiconductors with direct gaps at X point, the direct gaps open at X point. It is possible that direct band gap will become potential building blocks for electronic and optoelectronic devices.

Keywords Silicon nanotubes; Density functional theory; Periodic boundary conditions; Band gap ; Semiconductor

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Introduction

For almost thirty years, with the development of science and technology, nanomaterials¹⁻²have gradually become the focus of researchers.

From graphene³⁻⁴, diamond⁵⁻⁶, C_{60} fullerene⁷⁻⁸to nanowires⁹⁻¹⁰, nanobelt¹¹¹², nanotubes¹³⁻¹⁴. the successful synthesis of graphene¹⁵ has attracted widespread attention to carbon-based nanomaterials¹⁶¹⁷. later, people gradually increased their attention from nanomaterials¹⁸¹⁹ based carbon to non-carbon nanomaterials²⁰²¹, and gradually increased from zerodimensional nanoparticles²² to one-dimensional nanotubes²³⁻²⁴ and nanowires²⁵⁻²⁶to two-dimensional graphene²⁷⁻²⁸and then extended to three-dimensional graphite²⁹⁻³⁰.

Nano-materials with its unique physical and batteries³¹⁻³², lithium chemical properties in optoelectronic nanodevices³³⁻³⁴, industrial areas³⁵ and even global sustainability³⁶ have a lot of value. the most famous graphene is a strong material with good electrical conductivity, like rare metals (Rickard Arvidsson and Bjorn Sanden et al³⁷ Have studied new ways to replace rare metals with carbon nanomaterials. Rare metals are widely found in our daily necessities around us, computers, mobile phones, etc. almost all other electronic devices and around you found in the plastic, our society is highly dependent on rare metals, which ensures a high degree of pursuit of rare metals, but Because rare metals such as tin, silver, tungsten and indium are rare and difficult to extract because their concentration is very small).

In addition to two-dimensional graphene³⁸³⁹, onedimensional nanotube⁴⁰⁻⁴¹and nanowires⁴²⁴³have also attracted widespread attention from scientists, especially carbon nanotubes⁴⁴⁴⁵ and non-carbon nanotubes⁴⁶⁻⁴⁷, which exhibit different physical and chemical properties, the application of drugs⁴⁸, catalysis⁴⁹ and nano-electronic⁵⁰ equipment and other fields⁵¹.

There are some articles on the theory and experiment of silicon nanotubes. J Sha et al^{52} using nanochannel Al $_2O_3$ as a template and a chemical vapor deposition (CVD) synthesize silicon nanotubes, SB Fagan et al^{53} using first-principles calculations based on density functional theory, RQ Zhang et al^{54} using the semiempirical molecular orbital PM3 method, M Zhang

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