



Research paper

Diverse stacked and entangled topologies in cadmium tricarballylate coordination polymers with nitrobenzene detection capability

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ARTICLE INFO

Keywords:

Cadmium
Coordination polymer
X-ray crystallography
Entanglement
Nitrobenzene detection

ABSTRACT

Hydrothermal reaction of cadmium salts, tricarballylic acid (H_3tca) and dipyriddy ligands resulted in generation of four coordination polymers that were characterized by single-crystal X-ray diffraction. The nature of the ancillary dipyriddy ligand plays a primary role in enforcing differing structural topologies across the series of new phases. $[Cd_2(tcaH)_2(dpe)_2]_n$ (**1**, $dpe = 1,2$ -di(4-pyridyl)ethane) displays a 2-fold interpenetrated 3D $4^{12}6^3$ pcu net based on 6-connected $\{Cd_2(O)(OCO)\}$ cluster nodes. $[Cd(tcaH)(dpp)(H_2O)]_n$ (**2**, $dpp = 1,2$ -di(4-pyridyl)propane) manifests a 1D + 1D \rightarrow 3D interlocked system of loop-containing perpendicular ribbon motifs. $\{[Cd_4(tca)_2(Htca)(bpmp)_5(H_2O)_2] \cdot 27H_2O\}_n$ (**3**, $bpmp = \text{bis}(4\text{-pyridylmethyl})\text{piperazine}$) has intersecting $[Cd_4(tca)_2(Htca)(H_2O)_2]_n$ layer, $[Cd(bpmp)(H_2O)]_n$ chain, and $[Cd_2(bpmp)_3]_n$ ladder submotifs and forms a complicated 3,4,5-connected trinodal 3D self-penetrated network with $(6.9^2)(4^26^38)(4^26^38.9^310)$ topology. $\{[Cd(tcaH)(Hbpmp)](ClO_4) \cdot 5.5H_2O\}_n$ (**4**) exhibits stacked cationic bilayer (4,4) grid slabs. All compounds showed potential as potential detectors for nitrobenzene in ethanol suspension via luminescence quenching, with **1** showing the highest uptake of analyte.

1. Introduction

Crystalline coordination polymers remain an important research focus, as due to their significant capabilities in gas storage [1], shape-selective separations [2], ion exchange [3], catalysis [4], and detection of nitroaromatic explosives traces [5]. Additional impetus for exploratory synthesis in this subgenre of chemistry is offered by the aesthetic beauty of their molecular networks [6], including interpenetrated [7] or self-penetrated topologies [8]. One of the most efficacious anionic ligand choices for the construction of functional coordination polymers is 1,3,5-benzenetricarboxylate (btc), which has afforded the classic and well-studied absorbent metal-organic framework material $[Cu_3(btc)_2]_n$ [9]. Inclusion of dipyriddy-type coligands and use of different metal atoms has expanded the structural scope and utility of this class of materials [10–13], such as extensive gas absorption and bright luminescence in a zinc 1,3,5-benzenetricarboxylate coordination polymer incorporating a dipyriddy derivatized BODIPY-type ligand [10].

Less commonly employed have been aliphatic tricarballylate ligands, which can induce greater degrees of conformational freedom due to their flexible C–C σ bonds. A series of 2D and 3D gadolinium and europium coordination polymers based on the methanetricarboxylate ligand

has been reported by Ruiz-Pérez and coworkers [14,15]. The tricarballylate ligand (1,2,3-propanetricarboxylate, tca) has been found to regulate the biomineralization of calcite [16] and can also act as a carbon source for the growth of the fungus *Exophiala spinifera* [17]. Although the use of tricarballylate in the generation of coordination polymers has been infrequent, its derivatives can show extremely intriguing interlocked and self-penetrated topologies due to its conformational flexibility and multifarious possible binding and bridging modes [17–22]. $\{[Zn_3(tca)_2(bpmp)(Hbpmp)_2](ClO_4)_2 \cdot 5H_2O\}_n$ ($bpmp = \text{bis}(4\text{-pyridylmethyl})\text{piperazine}$) possessed a unique self-penetrated 2-D layered topology with threaded-loop type linkages, while the oxoanion-free phase $\{[Zn_3(H_2O)_4(tca)_2(bpmp)_2] \cdot 8H_2O\}_n$ manifested a three-fold interpenetrated $(6^3)(6^58)$ 3,4-connected 3-D binodal network [18]. $\{[Cd_2(tca)_2(Hdpa)_2] \cdot 7H_2O\}_n$ ($hdpa = 4,4'$ -dipyriddyamine) exhibited 1-D ribbon motifs with pendant $Hdpa^+$ ligands. $\{[Zn_3(tca)_2(dpa)_2] \cdot 2H_2O\}_n$ displayed a new 4,4-connected binodal $(4^26^28^2)_4(6^28^4)$ topology, while the copper derivative $[Cu(Htca)(dpa)]_n$ manifested a 3,5-connected binodal net with rare $(4.6^2)(4.6^68^3)$ topology [19]. To the best of our knowledge, no tricarballylate-containing coordination polymers have as yet been prepared using the simple dipyriddy linkers 1,2-di(4-pyridyl)ethane (dpe) and 1,3-di(4-pyridyl)propane (dpp).

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Table 1
Crystal and Structure Refinement Data for 1–4.

Data	1	2
Empirical Formula	C ₃₆ H ₃₆ Cd ₂ N ₄ O ₁₂	C ₁₉ H ₂₂ CdN ₂ O ₇
Formula Weight	941.49	502.79
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> ₂ ₁ / <i>c</i>	<i>C</i> ₂ / <i>c</i>
<i>a</i> (Å)	13.9255(10)	18.8260(14)
<i>b</i> (Å)	12.7541(9)	14.6412(11)
<i>c</i> (Å)	20.0983(14)	15.4657(11)
β (°)	96.4030(10)	112.5690(10)
<i>V</i> (Å ³)	3547.3(4)	3936.4(5)
<i>Z</i>	4	8
<i>D</i> _{calc} (g cm ⁻³)	1.763	1.697
μ (mm ⁻¹)	1.270	1.154
Min./max. trans.	0.8170	0.8968
<i>hkl</i> ranges	-16 ≤ <i>h</i> ≤ 16 -13 ≤ <i>k</i> ≤ 15 -24 ≤ <i>l</i> ≤ 24	-10 ≤ <i>h</i> ≤ 10 -15 ≤ <i>k</i> ≤ 15 -16 ≤ <i>l</i> ≤ 16
Total reflections	21,732	33,712
Unique reflections	6541	4830
<i>R</i> (int)	0.0288	0.0454
Parameters	493	271
<i>R</i> ₁ (all data)	0.0298	0.0369
<i>R</i> ₁ (<i>I</i> > 2σ(<i>I</i>))	0.0241	0.0257
<i>wR</i> ₂ (all data)	0.0598	0.0588
<i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.0568	0.0544
Max/min residual (e ⁻ /Å ³)	0.496/-0.412	0.472/-0.303
G.O.F.	1.027	1.041

Data	3	4
Empirical Formula	C ₉₈ H ₁₇₄ Cd ₄ N ₂₀ O ₄₇	C ₂₂ H ₃₈ CdClN ₄ O ₁₆
Formula Weight	2834.18	762.41
Crystal system	Monoclinic	Monoclinic
Space group	<i>P</i> ₂ ₁ / <i>n</i>	<i>P</i> ₂ ₁ / <i>c</i>
<i>a</i> (Å)	12.3556(10)	17.1148(13)
<i>b</i> (Å)	12.0599(10)	20.5960(16)
<i>c</i> (Å)	40.705(3)	8.7252(7)
β (°)	90.5390(10)	98.8400(10)
<i>V</i> (Å ³)	6065.1(9)	3039.1(4)
<i>Z</i>	2	4
<i>D</i> _{calc} (g cm ⁻³)	1.551	1.666
μ (mm ⁻¹)	0.787	0.886
Min./max. trans.	0.8846	0.6920
<i>hkl</i> ranges	-14 ≤ <i>h</i> ≤ 14 0 ≤ <i>k</i> ≤ 14 0 ≤ <i>l</i> ≤ 49	-20 ≤ <i>h</i> ≤ 20 -24 ≤ <i>k</i> ≤ 24 -10 ≤ <i>l</i> ≤ 10
Total reflections	137,985	24,378
Unique reflections	11,266	5540
<i>R</i> (int)	0.0702	0.0360
Parameters	974	435
<i>R</i> ₁ (all data)	0.0640	0.0612
<i>R</i> ₁ (<i>I</i> > 2σ(<i>I</i>))	0.0531	0.0538
<i>wR</i> ₂ (all data)	0.1264	0.1613
<i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	0.1217	0.1530
Max/min residual (e ⁻ /Å ³)	1.600/-1.245	1.871/-2.140
G.O.F.	1.098	1.071

The zinc and cadmium tricarballoylate coordination polymers in these previous studies all exhibited fluorescent behavior upon irradiation with ultraviolet light, due to π - π^* transitions within the aromatic dipyriddy ligands, but were not assayed as detectors or sensors. Nevertheless, the visible light spectral window of this class of material could prove beneficial for the detection of nitroaromatics via luminescence quenching. In this contribution we present the synthesis, single-crystal structural determination, and nitrobenzene detection assays for four new cadmium-based coordination polymers containing the tricarballoylate ligand: [Cd₂(tcaH)₂(dpe)₂]_n (1), [Cd(tcaH)(dpp)(H₂O)]_n (2), {[Cd₂(tca)₂(Htca)(bpmp)₅(H₂O)₂·27H₂O]_n (3), and {[Cd(tcaH)(Hbpmp)](ClO₄)·5.5H₂O]_n (4). Thermal properties of the non-perchlorate containing materials are also discussed herein.

Table 2
Selected Bond distance (Å) and angle (°) data for 1.

Cd1–O7	2.2337(16)	N2 ^{#2} –Cd1–O1	85.41(7)
Cd1–O12 ^{#1}	2.2810(16)	N1–Cd1–O1	97.83(7)
Cd1–N2 ^{#2}	2.3029(19)	O2–Cd1–O1	54.96(6)
Cd1–N1	2.3085(19)	O8–Cd2–N4 ^{#2}	88.34(7)
Cd1–O2	2.3451(18)	O8–Cd2–N3	88.39(7)
Cd1–O1	2.4310(18)	N4 ^{#2} –Cd2–N3	176.32(7)
Cd2–O8	2.3262(17)	O8–Cd2–O3 ^{#3}	85.49(6)
Cd2–N4 ^{#2}	2.330(2)	N4 ^{#2} –Cd2–O3 ^{#3}	95.54(7)
Cd2–N3	2.3367(19)	N3–Cd2–O3 ^{#3}	85.87(7)
Cd2–O3 ^{#3}	2.3439(18)	O8–Cd2–O11 ^{#1}	137.87(7)
Cd2–O11 ^{#1}	2.3554(16)	N4 ^{#2} –Cd2–O11 ^{#1}	89.49(7)
Cd2–O4 ^{#3}	2.5023(18)	N3–Cd2–O11 ^{#1}	91.81(6)
Cd2–O12 ^{#1}	2.5746(18)	O3 ^{#3} –Cd2–O11 ^{#1}	136.56(6)
O7–Cd1–O12 ^{#1}	111.03(6)	O8–Cd2–O4 ^{#3}	138.01(6)
O7–Cd1–N2 ^{#2}	95.41(7)	N4 ^{#2} –Cd2–O4 ^{#3}	85.40(7)
O12–Cd1–N2 ^{#2}	85.19(7)	N3–Cd2–O4 ^{#3}	98.17(7)
O7–Cd1–N1	86.71(7)	O3 ^{#3} –Cd2–O4 ^{#3}	54.03(6)
O12 ^{#1} –Cd1–N1	90.91(7)	O11 ^{#1} –Cd2–O4 ^{#3}	83.61(6)
N2 ^{#2} –Cd1–N1	176.02(7)	O8–Cd2–O12 ^{#1}	86.18(6)
O7–Cd1–O2	146.16(6)	N4 ^{#2} –Cd2–O12 ^{#1}	98.45(6)
O12 ^{#1} –Cd1–O2	102.01(6)	N3–Cd2–O12 ^{#1}	79.65(6)
N2 ^{#2} –Cd1–O2	94.39(7)	O3 ^{#3} –Cd2–O12 ^{#1}	163.47(6)
N1–Cd1–O2	85.64(7)	O11 ^{#1} –Cd2–O12 ^{#1}	52.62(5)
O7–Cd1–O1	93.70(6)	O4 ^{#3} –Cd2–O12 ^{#1}	135.81(5)
O12 ^{#1} –Cd1–O1	154.24(6)		

Symmetry transformation to generate equivalent atoms: #1 *x*, -*y* + 3/2, *z* - 1/2; #2 *x* + 1, *y*, *z*; #3 *x*, *y* + 1, *z*.

2. Experimental section

2.1. General considerations

Cadmium salts, tricarballoylic acid, dpe, and dpp were purchased commercially. Bis(4-pyridylmethyl)piperazine was prepared using a published procedure [23]. Water was deionized above 3 MΩ-cm in-house. IR spectra were recorded on powdered samples using a Perkin Elmer Spectrum One instrument. The luminescence spectra were obtained with a Hitachi F-4500 Fluorescence Spectrometer. **CAUTION! Perchlorate compounds can be explosive! While no issues arose during this study, minimum quantities were used and no perchlorate-containing solids were heated above their synthesis temperature.**

2.2. Preparation of [Cd₂(tcaH)₂(dpe)₂]_n (1)

CdSO₄·8H₂O (63 mg, 0.18 mmol), dpe (68 mg, 0.37 mmol) and tricarballoylic acid (33 mg, 0.19 mmol) were placed into 5 mL distilled H₂O in a 15 mL screw-cap glass vial. The vial was sealed and heated in an oil bath at 80 °C for 72 hrs, whereupon it was cooled slowly to 25 °C. Colorless blocks of 1 (66 mg, 78% yield based on Cd) were isolated after washing with distilled water and acetone, and drying in air. Anal. Calc. for C₃₆H₃₆Cd₂N₄O₁₂ 1: C, 45.92; H, 3.85; N, 5.95% Found: C, 45.77; H, 3.73; N, 5.89%. IR (ν): 3251 (w, br), 1702 (m), 1612 (m), 1579 (s), 1426 (w), 1397 (s), 1312 (w), 1258 (w), 1226 (w), 1098 (w), 1018 (w), 832 (m), 808 (w), 722 (w), 675 (w) cm⁻¹.

2.3. Preparation of [Cd(tcaH)(dpp)(H₂O)]_n (2)

Cd(ClO₄)₂·6H₂O (56 mg, 0.18 mmol), dpp (73 mg, 0.37 mmol) and tricarballoylic acid (33 mg, 0.19 mmol) were placed into 5 mL distilled H₂O in a 15 mL screw-cap glass vial. The vial was sealed and heated in an oil bath at 80 °C for 72 hrs, whereupon it was cooled slowly to 25 °C. Colorless blocks of 2 (82 mg, 91% yield based on Cd) were isolated after washing with distilled water and acetone, and drying in air. Anal. Calc. for C₁₉H₂₂CdN₂O₇ 2: C, 45.39; H, 4.41; N, 5.57% Found: C, 45.31; H, 4.36; N, 5.92%. IR (ν): 3298 (w, br), 2926 (w), 2500 (w, br), 1717 (s), 1610 (m), 1577 (s), 1560 (s), 1542 (w), 1503 (w), 1450 (m), 1417 (s), 1359 (s), 1314 (m), 1278 (m), 1224 (m), 1193 (s), 1172 (m), 1097 (w),

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