

## Accepted Manuscript

### Article

Atomically dispersed Au<sub>1</sub> catalyst towards efficient electrochemical synthesis of ammonia

Xiaoqian Wang, Wenyu Wang, Man Qiao, Geng Wu, Wenxing Chen, Tongwei Yuan, Qian Xu, Min Chen, Yan Zhang, Xin Wang, Jing Wang, Jingjie Ge, Xun Hong, Yafei Li, Yuen Wu, Yadong Li

PII: S2095-9273(18)30325-6  
DOI: <https://doi.org/10.1016/j.scib.2018.07.005>  
Reference: SCIB 455

To appear in: *Science Bulletin*



Please cite this article as: X. Wang, W. Wang, M. Qiao, G. Wu, W. Chen, T. Yuan, Q. Xu, M. Chen, Y. Zhang, X. Wang, J. Wang, J. Ge, X. Hong, Y. Li, Y. Wu, Y. Li, Atomically dispersed Au<sub>1</sub> catalyst towards efficient electrochemical synthesis of ammonia, *Science Bulletin* (2018), doi: <https://doi.org/10.1016/j.scib.2018.07.005>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

## Atomically dispersed Au<sub>1</sub> catalyst towards efficient electrochemical synthesis of ammonia

Xiaoqian Wang,<sup>1†</sup> Wenyu Wang,<sup>1†</sup> Man Qiao,<sup>2</sup> Geng Wu,<sup>1</sup> Wenxing Chen,<sup>3</sup> Tongwei Yuan,<sup>4</sup> Qian Xu,<sup>5</sup> Min Chen,<sup>1</sup> Yan Zhang,<sup>1</sup> Xin Wang,<sup>1</sup> Jing Wang,<sup>1</sup> Jingjie Ge,<sup>1</sup> Xun Hong,<sup>1</sup> Yafei Li,<sup>2</sup> Yuen Wu<sup>1\*</sup>, Yadong Li<sup>3</sup>

<sup>1</sup>Department of Chemistry, *iChEM* (Collaborative Innovation Center of Chemistry for Energy Materials), University of Science and Technology of China, Hefei 230026, China

<sup>2</sup>Jiangsu Collaborative Innovation Centre of Biomedical Functional Materials, School of Chemistry and Materials Science, Nanjing Normal University, Nanjing 210023, China.

<sup>3</sup>Department of Chemistry and Collaborative Innovation Center for Nanomaterial Science and Engineering, Tsinghua University, Beijing 100084, China

<sup>4</sup>NEST Lab, Department of Chemistry, College of Science, Shanghai University, Shanghai 200444, China

<sup>5</sup>National Synchrotron Radiation Laboratory, University of Science and Technology of China, Hefei 230029, China

<sup>†</sup>Xiaoqian Wang and Wenyu Wang contributed equally to this work.

\*Correspondence: [yuenwu@ustc.edu.cn](mailto:yuenwu@ustc.edu.cn)

### Abstract

Tremendous efforts have been devoted to explore energy-efficient strategies of ammonia synthesis to replace Haber-Bosch process which accounts for 1.4% of the annual energy consumption. In this study, atomically dispersed Au<sub>1</sub> catalyst is synthesized and applied in electrochemical synthesis of ammonia under ambient conditions. A high NH<sub>4</sub><sup>+</sup> Faradaic efficiency of 11.1% achieved by our Au<sub>1</sub> catalyst surpasses most of reported catalysts under comparable conditions. Benefiting from efficient atom utilization, an NH<sub>4</sub><sup>+</sup> yield rate of 1,305 μg h<sup>-1</sup> mg<sub>Au</sub><sup>-1</sup> has been reached, which is roughly 22.5 times as high as that by supported Au nanoparticles. We also demonstrate that by employing our Au<sub>1</sub> catalyst, NH<sub>4</sub><sup>+</sup> can be electrochemically produced directly from N<sub>2</sub> and H<sub>2</sub> with an energy utilization rate of 4.02 mmol kJ<sup>-1</sup>. Our study provides a possibility of replacing the Haber-Bosch process with environmentally benign and energy-efficient electrochemical strategies.

### Keywords

NH<sub>3</sub> synthesis, metal single sites, electrocatalysis, Haber–Bosch process, nitrogen reduction

### 1. Introduction

As one of the most essential industrial chemicals, ammonia (NH<sub>3</sub>) is currently produced on an enormous scale of over 150 megatons per year by the Haber–Bosch process which requires pressures of 200 to 300 atmospheres and temperatures from 300 to 500 °C. To date, this energy- and capital-intensive process accounts for 1.4% of the annual energy consumption and around 3% of global CO<sub>2</sub> emissions [1-5]. Electrocatalytic approach, especially driven by renewable energy,

Download English Version:

<https://daneshyari.com/en/article/11017643>

Download Persian Version:

<https://daneshyari.com/article/11017643>

[Daneshyari.com](https://daneshyari.com)