ELSEVIER

Contents lists available at ScienceDirect

Journal of Environmental Management

journal homepage: www.elsevier.com/locate/jenvman



Review

Overview of microbes based fabricated biogenic nanoparticles for water and wastewater treatment



Imran Ali^a, Changsheng Peng^{a,b,c,**}, Zahid M. Khan^d, Iffat Naz^e, Muhammad Sultan^{d,*}, Mohsin Ali^f, Irfan A. Abbasi^a, Tariqul Islam^a, Tong Ye^a

- ^a College of Environmental Science and Engineering, Ocean University of China, Qingdao 266100, China
- b The Key Lab of Marine Environmental Science and Ecology, Ministry of Education, Ocean University of China, Qingdao 266100, China
- ^c School of Environmental and Chemical Engineering, Zhaoqing University, Zhaoqing 526061, China
- ^d Department of Agricultural Engineering, Bahauddin Zakariya University, Bosan Road, Multan 60800, Pakistan
- ^e Department of Biology, Qassim University, Buraidah 51452, Saudi Arabia
- f Department of Environmental Engineering, Middle East Technical University, Ankara 0600, Turkey

ARTICLE INFO

Keywords: Microbes Biogenic nanoparticles Wastewater Water Treatment

ABSTRACT

Treatment of toxic and emerging pollutants (T&EPs) is increasing the threats to the survival of conventional wastewater treatment (WWTs) technologies. The high installation and operational costs of advanced treatment technologies have shifted the research interest to the development of economical and reliable technology for management of T&EPs. Thus, recently biogenic nanoparticles (BNPs) fabricated using microbes/microorganisms are getting tremendous research interest due to their unique properties (i.e. high specific surface area, desired morphology, catalytic reactivity) for the biodegradation and biosorption of T&EPs. In addition, BNPs can be manufactured using metal contaminated water which indicates a hidden potential for resource recovery and utilization. Therefore, the present study discusses the adsorptive and catalytic performance of BNPs in the removal of T&EPs from water (W) and wastewater (WW). In addition, inspired by the superior performance of BNPs in advance WWT, a model of BNPs based WWT resource recovery and utilization process is also proposed. Finally, main issues i.e. mass production, leaching, poisoning/toxicity, regeneration, reusability and fabrication costs and process optimization are discussed which are main hinders in the transfer of BNPs based WWT technologies from laboratory to commercial scale.

1. Introduction

Currently, the discovery of additional toxic and emerging pollutants (T&EPs) is increasingly sabotaging the survival of conventional wastewater treatment technologies (Puvol et al., 2016; Zhuang et al., 2015). Emerging or priority pollutants (E/PPs), such as toxic heavy metals, organic and inorganic micro-pollutants, pharmaceuticals and halogenated compounds, pigments, toxic dyes and pathogenic microorganisms are increasingly reported in many studies (Xiao et al., 2017; Kim et al., 2016; Jain et al., 2016). The concentration and volume of these pollutants are increasing continuously due to a rapid increase in domestic, industrial and agricultural activities (Zhou et al., 2016). On the other hand, the stringent limits of these T&EPs are creating limitations for the existing water and wastewater treatment technologies (W/WWTT) to discharge these contaminants safely into the natural water environment

(Liu et al., 2015). Thus, there is a need to discover the alternate options that can address these issues.

Since the development of the conception of microbial interaction with metals, biogenic nanoparticles (BNPs) have been getting tremendous research interest from last few decades (Gillan, 2016). Owing to their unique properties, BNPs have been employed in the applications of biomedical, electronic, and W/WWT approaches (Ali et al., 2017a, 2017b). The engineering of BNPs fabrication via microbial interaction is now well documented (Holmes and Gu, 2016). In contrast, the fabrication of nanoparticles (NPs) via conventional methods is often expensive and hazardous due to the involvement of toxic reducing and stabilizing agents (Rezić, 2011). However, BNPs can be manufactured in a safe environment using microbes as reductant (Kitching et al., 2015; Kharissova et al., 2013). In addition, heavy metal polluted water can also be used instead of salts to fabricate BNPs. It is also a method for

^{*} Corresponding author. Department of Agricultural Engineering, Bahauddin Zakariya University, Bosan Road, Multan 60800, Pakistan.

^{***} Corresponding author. The Key Lab of Marine Environmental Science and Ecology, Ministry of Education, Ocean University of China, Qingdao 266100, China. E-mail addresses: cspeng@ouc.edu.cn (C. Peng), muhammadsultan@bzu.edu.pk (M. Sultan).

Nomenclature	Fe ₃ O ₄ TW Tea waste
	FTIR Fourier transformed infrared analysis
ABCSBR Aerobic bubble column sequencing batch reactor	HAuCl4•3H2O Hydrogen tetrachloroaurate (III) hydrate
AFM Atomic force microscope	HRTEM High resolution transmission electron microscopy
AI-FeNPs Azadiracta indica	INPB Ironoxide nanoparticle beads
AMMC-Fe ₃ O ₄ Acid-modified maize cob impregnated with ferroso-	MB Methylene blue
ferric oxide	MG Malachite green
BET Braunauer–Emmett–Teller	MO Methyl orange
BioMnOx Biogenic manganese oxide nanoparticles	MOB Manganese oxidizing bacteria \
Bio-Pd Biogenic nano-palladium	MRGO Magnetite/reduced graphene oxide
BioSeNPs Biogenic elemental selenium nanoparticles	Ms Magnetic saturation
BMNPs Biogenic magnetic nanoparticles	Na2PdCl4Sodium tetrachloropalladate
BMO Biogenic Mn oxide; BnM: Biogenic magnetite	Na2-SeO3 Sodium selenite
CSNPs-Fe ₃ O ₄ -Ag Core shell nanoparticles	NaBH4 Sodium boro-hydride
CueO Biocatalyst multicopper oxidase	NMR Nuclear magnetic resonance
DEAMTPP@Fe ₃ O ₄ MNP Diethyl-4-(4-amino-5-mercapto-4H-1,2,4-	ORS/Fe3O4/PCL-NCs Modified rice straw/Fe ₃ O ₄ /polycaprolactone
triazol-3-yl)phenyl phosphonate	nanocomposites
DHPCT@Fe ₃ O ₄ MNP 3,4-dihydroxyphenethylcarbamodithioate	PXRD Powder X-ray Diffraction
(DHPCT)	RhB Rhodamine B
DLS Dynamic light scattering DMSA@Fe3O4 MNRs	SAED Selected-area electron diffraction
Dimercaptosuccinic acid (DMSA) anchored Fe3O4 mag-	SEM Scanning Electron Microscopy
netic nanorods (MNRs)	SMNPs Bio-inspired Fe ₃ O ₄ spherical magnetic nanoparticles
DRS Diffuse reflectance spectroscopy	SQUID Superconducting quantum interference device
DSC Differential scanning calorimetry	TCE Trichloroethylene
ECueO cueO-overexpressing Escherichia coli strain	TEM Transmission Electronic Microscopy
EDDS (S,S)-ethylenediamine-N,N0-disuccinic acid ()	TGA Thermal gravimetric analysis
EDS Energy dispersive X-ray spectroscopy	UASB Up-flow anaerobic sludge blanket
EDTA Ethylenediamine tetraacetate	UV-VS UV-visible Spectrometer
EL-Fe Eucalyptus leaves extract	VSM Vibrating sample magnetometer
Fe ₃ O ₄ @2D-CF composite Bioinspired graphene oxide-like 2D-	XPS X-ray photoelectron spectroscopy
carbon flake (CF) Fe ₃ O ₄ nanoparticles	XRF X-ray fluorescence
Fe ₃ O ₄ @DAPF-CSFMNR Core–shell ferromagnetic nanorods	ZP Zeta-potential

resource recovery and the safe utilization of domestic and industrial wastewaters (Meerburg et al., 2012). The rapid increase in price of metals which are being used as a raw material in most industries including leather tanning, cement production, metallurgy, electroplating, electronics, has propelled our research interest to recover these metals from polluted waters and soils (Quan et al., 2015; Hennebel et al., 2011a,b).

Apart from the safe fabrication protocol, BNPs have been exploited for the generation of electricity, remediation of polluted soils and waters, and removal/recovery of scarce and precious metals (Martins et al., 2017; Qu et al., 2017). In this regard, Fig. 1 shows an increasing research interest in BNPs applications in W/WWT. Owing to the presence of a unique bacterial carrier matrix, BNPs have been used as biocatalysts, adsorbents, oxidants, and reductants in the removal of T&

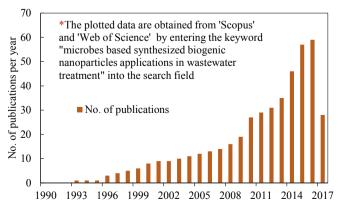


Fig. 1. Literature survey and research developments about biogenic nanoparticles (BNPs) applications in water and wastewater treatment (W/WWT).

EPs from drinking water and wastewater (Gusseme et al., 2012). Most studies stated that the presence of specific functional groups (i.e. hydroxyl, carboxylate, methyl and amide I, II &II) and the variety of reducing substance generated by the bacterial cells (charged opposite to targeted pollutants) are primarily responsible for the removal of contaminants via electrostatic interactions and the ion exchange process (Ahluwalia et al., 2016; Qu et al., 2016; Yue et al., 2016; Furgal et al., 2015). In addition, BNPs can be activated by providing hydrogen gas (H₂) or formate as an electron donor for the degradation of T&EPs (Suja et al., 2014; De et al., 2012; Hennebel et al., 2010). Few studies have demonstrated that molecular oxygen can be produced using Mn-oxidizing bacteria for the bio (degradation) of recalcitrant pollutants via the combination of electrostatic interaction, co-precipitation and bacterial co-metabolism (Su et al., 2014; Meerburg et al., 2012). Although BNPs have shown superior performance in the removal of T&EPs, however, certain aspects (detail in section 4) are mainly creating barriers in the market penetration of BNP based W/WWT technologies.

1.1. Scope of the study

Several studies have reviewed different features of BNPs based on the fabrication and extensive application of BNPs in different fields of modern science. The reviews on BNPs fabrication and application can be found in literature (Zan and Wu, 2016; Shamaila et al., 2016, Siddiqi and Husen, 2016a,b; Huang et al., 2015, Sharma et al., 2015, Moritz and Geszke, 2013; Narayanan and Sakthivel, 2010). Different kinds of BNPs e.g. organic, inorganic, semiconductor, chalcogenide quantum dots, can be manufactured using different microbial strains and their applications would also vary. For instance, Rubilar et al. (2013) reported a detailed description of the applications of BNPs fabricated using copper, copper sulphides and copper oxides. Quantum dots have

Download English Version:

https://daneshyari.com/en/article/11023229

Download Persian Version:

https://daneshyari.com/article/11023229

<u>Daneshyari.com</u>