## Accepted Manuscript

Solvent screening and liquid-liquid measurement for extraction of phenols from aromatic hydrocarbon mixtures

Xingkun Liu, Xianglan Zhang

PII: S0021-9614(18)30536-6

DOI: https://doi.org/10.1016/j.jct.2018.09.006

Reference: YJCHT 5535

To appear in: J. Chem. Thermodynamics

Received Date: 22 May 2018

Revised Date: 11 September 2018 Accepted Date: 12 September 2018



Please cite this article as: X. Liu, X. Zhang, Solvent screening and liquid-liquid measurement for extraction of phenols from aromatic hydrocarbon mixtures, *J. Chem. Thermodynamics* (2018), doi: https://doi.org/10.1016/j.jct. 2018.09.006

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

**ACCEPTED MANUSCRIPT** 

Solvent screening and liquid-liquid measurement for extraction of phenols

from aromatic hydrocarbon mixtures

Xingkun Liu, Xianglan Zhang\*

College of Chemistry and Environmental Engineering, China University of Mining and Technology, Beijing 100083, China

**ABSTRACT:** The separation of phenolic compounds from low temperature coal tar has great

significance for further application of the coal tar. In this paper, the separation of special phenolic

compounds from model coal tar (phenols + toluene) was studied. The universal quasi chemical

functional group activity coefficient (UNIFAC) and conductor-like screening model COSMO-SAC

(segment activity coefficient) was carried out to screen suitable solvent. Ethylene glycol was selected

as the potential solvent to extract the phenols from aromatic hydrocarbon, evaluated by area of

two-phase region and the criterion of solvent power  $(SP_i^{\infty})$ , selectivity  $(S^{\infty})$  and performance index (PI).

Liquid-liquid equilibria of the ethylene glycol + phenols + toluene ternary systems were measured in

the temperature range (303.15-323.15) K under 101.3 kPa, and the results showed that ethylene glycol

has a high extraction efficiency with distribution coefficient (between 2 and 12) and selectivity

(between 7 and 255). Moreover, the NRTL and UNIQUAC models were successfully applied to

correlate the experimental LLE data with RMSD less than 1.85%, indicating that both models can

accurately describe LLE behaviour of this system. Thus, the corresponding binary interaction

parameters would be helpful for separation process designing or optimizing.

**Keywords:** Liquid-liquid equilibria; toluene; ethylene glycol; phenols; COSMO-SAC.

1. Introduction

\* Corresponding author.

E-mail address: <u>zhxl@cumtb.edu.cn</u> (X.L. Zhang)

## Download English Version:

## https://daneshyari.com/en/article/11028241

Download Persian Version:

https://daneshyari.com/article/11028241

<u>Daneshyari.com</u>