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Authors: Jishen Li, Xiyun Yang, Zhoulan Yin

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Recovery of manganese from sulfuric acid leaching liquor of spent lithium-ion batteries and synthesis of lithium ion-sieve

Jishen Li^a, Xiyun Yang^{*a, b}, Zhoulan Yin^b

^aSchool of Metallurgy and Environment, Central South University, Changsha 410083, China

^bCollege of Chemistry and Chemical Engineering, Central South University, Changsha 410083, China

Highlights

- Mn was recovered as lithium ion-sieve from leaching solution of spent LIBs.
- Mn was separated from Ni and Co with D2EHPA extraction following two-step stripping.
- Co doping improves structure stability and adsorption capacity of lithium ion-sieve.

ABSTRACT

A novel process of recovery of Mn as lithium ion-sieve from H_2SO_4 leaching liquor of spent lithium-ion batteries was developed. After precipitation of Fe³⁺ and Cu²⁺ in the leaching liquor, Mn was recovered with D2EHPA extraction followed by two-step stripping with H_2SO_4 solution. Lithium ion-sieve was synthesized by oxidation of the stripping solution using KMnO₄, then roasting at atmosphere followed by acid leaching. The effects of solvent extraction parameters on Mn recovery rate and Co doping content on the performance of lithium ion-sieve were investigated. The result indicates that through secondary countercurrent solvent extraction, Mn is fully separated from Ni and a MnSO₄ solution containing a small amount of Co is obtained. Co can incorporate into the lattice of lithium ion-sieve and improve the structure Download English Version:

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