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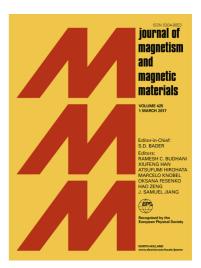
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Study of structural, morphological and magnetic properties Ag substituted cobalt ferrite nanoparticles prepared by honey assisted combustion method and evaluation of their antibacterial activity

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Abstract

Pure and Ag substituted cobalt ferrite nanoparticles (NPs) baring the composition (1-x)CoFe₂O₄: xAg (x = 0.0.2) were synthesized by a novel honey assisted combustion method in order to investigate their structural and magnetic properties along withtheir antibacterial activity. XRD patters confirm the spinel phase of CoFe₂O₄ and the presence of silver (Ag) nanoparticles in the spinel network. The incorporation of Ag in CoFe₂O₄ spinel structure enhanced the size of the unit cell, resulting to higher value of lattice parameter (a) compared to the pure CoFe₂O₄. The sintering process promoted the growth of the crystallite sizes (D). The crystallite sizes of the synthesized and annealed powders were found in the range of 24–41 nm. From the EDX studies, it seemed that the distribution of Ag nanoparticles was non-uniform. The saturation magnetization (M_s) and coercivity (H_c) of the powders were influencedbyannealing as well as with the substitution of Ag. The highest value of saturation magnetization (60emu/g) was obtained by the CoFe₂O₄ nanoparticles with the coercivity value 1358 Oe. The saturation magnetization and coercivity of Ag doped CoFe₂O₄were less than that of pure CoFe₂O₄. The present cobalt ferrite nanoparticles and Ag doped cobalt ferrite nanoparticles have showed good antibacterial activities. But Ag doped cobalt ferrite

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