

Available online at www.sciencedirect.com



SPECTROCHIMICA ACTA PART A

Spectrochimica Acta Part A 61 (2005) 2444-2447

www.elsevier.com/locate/saa

Infrared (1.2–1.6 μm) luminescence in Cr⁴⁺:Yb₃Al₅O₁₂ single crystal with 940 nm diode pumping

Xiaodong Xu^{a,b}, Zhiwei Zhao^{a,*}, Pingxin Song^{a,b}, Guoqing Zhou^a, Jun Xu^a, Peizhen Deng^a, Gilbert Bourdet^c, Jean Christophe Chanteloup^c, Ji-Ping Zou^c, Annabelle Fulop^c

^a Shanghai Institute of Optics and Fine Mechanics, Chinese Academy of Sciences, P.O. Box 800-211, Shanghai 201800, People's Republic of China ^b Graduate School of Chinese Academy of Sciences, Beijing 100039, People's Republic of China

^c Laboratoire pour l'Utilisation des Lasers Intenses, UMR7605, CNRS-CEA-Université Paris VI, Ecole Polytechnique, 91128 Palaiseau Cedex, France

Received 12 August 2004; accepted 10 September 2004

Abstract

Infrared (1.2–1.6 μ m) luminescence in a ytterbium aluminium garnet (YbAG) crystal, doped with Cr (0.05 at.%) ions, was investigated under CW laser diode pumping ($\lambda = 940$ nm). The Cr⁴⁺ emission band was observed with its peak at 1.34 μ m and measured to be about 1.3 times with respect to Yb³⁺ IR luminescence ($\lambda = 1.03 \ \mu$ m). We demonstrate that for the excitation wavelength of 940 nm Yb³⁺ ions act as sensitizers of the ${}^{3}B_{2}({}^{3}T_{2})-{}^{3}B_{1}({}^{3}A_{2})$ emission of Cr⁴⁺ ions. This crystal is promising as a high-efficient system for tunable laser (1.2–1.6 μ m) output.

© 2004 Elsevier B.V. All rights reserved.

PACS: 42.70.Hj; 87.64.Ni

Keywords: Cr⁴⁺:Yb₃Al₅O₁₂ crystal; Luminescence; Energy transfer; Diode pumping

1. Introduction

Tetravalent chromium ion (Cr^{4+}) has been recognized as a popular and efficient ions for obtaining tunable laser operating in the near-infrared region (1.2–1.6 µm) [1–10]. Operating around 1.3 and 1.5 µm, the second and third telecommunication windows respectively, Cr:YAG lasers should find applications in spectroscopy, remote sensing, medicine and in the laboratory characterization of fibre communication systems [11].The material has a broad absorption peak around 1 µm and is usually pumped by Nd:YAG lasers operating at 1.06 µm [12–13].

It is well-known that rare-earth and transition metal elements are used both as active centers and sensitizers. The ytterbium ions have been studies as sensitizers in yttrium alumonum garnet (YAG) crystals [14–16]. Yb₃Al₅O₁₂ and $Y_3Al_5O_{12}$ belong to the crustal family of the rare-earth alu-

minium garnets (RE₃Al₅O₁₂, Re=rare earth or yttrium). They are isostructural with only about a 1.5% difference in unit-cell size and a 200 °C separation in their melting points, and they have a cubic symmetry with space group *Ia3d* [17]. In our recent letter [18] we showed that chromium can be easily doped into YbAG crystals to form tetrahedral Cr⁴⁺ ions for the purpose of obtaining a compact, efficient, highly stable, passive self-Q-switched thin chip solid-state lasers, and the Yb³⁺ ions act as active centres and Cr⁴⁺ ions act as excellent saturable absorbers. In this letter, we report on the experimental investigation of NIR luminescence of Cr⁴⁺ doped ytterbium aluminium garnet crystal under 940 nm excitation. The results demonstrate that it can act as a suitable material for Cr⁴⁺ tunable laser output.

2. Experiments

Raw materials of 5N pure Yb₂O₃, Al₂O₃, Cr₂O₃ and CaCO₃ were weighed, according to the stoichiometric composition. After the compounds were ground and thoroughly

^{*} Corresponding author. Tel.: +86 21 69918482; fax: +86 21 69918607. *E-mail address:* zzw8006@sina.com (Z. Zhao).

^{1386-1425/\$ –} see front matter 2004 Elsevier B.V. All rights reserved. doi:10.1016/j.saa.2004.09.007

mixed, they were pressed into pieces. The pieces were heated to 1200 °C and kept at the temperature for 20 h. Then the mixture was charged into an Iridium crucible of 50 mm in diameter for crystal growth. The pulling and rotation rates were 1 mm h⁻¹ and 10-20 rpm, respectively, under the nitrogen or argon atmosphere, respectively. The initial growth boundary in solid-melt was convex towards the melt so that dislocation and impurities were reduced or eliminated from the crystal. After that, the growth boundary became flat. In order to prevent the crystal from cracking, the crystal was cooled to room temperature slowly after growth. The crystal was blue-green and free from crack, inclusions and precipitations. When the 20 mW He-Ne laser beam passed through the as-grown Cr⁴⁺:YbAG, the light beam was almost unseen, which indicated very few scattering particles in Cr⁴⁺:YbAG crystal. The crystal changed to brown after annealing in oxygen atmosphere at 1400 °C for 24 h.

Samples for spectroscopic measurements were cut out of the boules and surfaces perpendicular to the $\langle 1 \ 1 \ 1 \rangle$ growth axis were polished. The thickness of the samples was 0.35 mm. A JASCO V-570 UV–vis/NIR Spectrophotometer was employed for acquisition of the absorption spectra at room temperature. The fluorescence spectra were acquired by a TRIAX 550 spectrophotometer with InGaAs LD as the pump source (excited at 940 nm). The decay time was measured by a computer controlled transient digitizer.

3. Results and discussion

The room temperature absorption spectra of YbAG and Cr^{4+} :YbAG crystals are displayed in Fig. 1. As it is seen from the spectra, the main absorption bands in YbAG and Cr^{4+} :YbAG crystals lie between 900 and 1050 nm (Yb³⁺, transition ${}^{2}F_{7/2}-{}^{2}F_{5/2}$), which is characteristic for all Yb³⁺ activated materials. Each absorption peak of Cr^{4+} :YbAG in-



Fig. 1. The absorption spectra of YbAG and Cr⁴⁺:YbAG crystals.

creases in intensity compared with YbAG. The absorption coefficient at 1030 nm increases from 15.6 to 17.5 cm⁻¹, and the increase of the absorption is due to the increase of the concentration of Cr^{4+} ions. The band centered at 1030 nm is believed to be caused by the ${}^{3}A_{2} \rightarrow {}^{3}T_{1}$ transition of Cr^{4+} ions. The absorption spectrum of Cr^{4+} , as shown in Fig. 2, can be calculated by subtracting the absorption spectrum of YbAG from the absorption spectrum of Cr^{4+} :YbAG. It can be seen that Cr^{4+} ions have broad absorption band, and the absorption is similar to that of Cr^{4+} :YAG crystal [4].

Fig. 3 shows the fluorescence spectra of Cr^{4+} :YbAG crystal under 940 nm diode pumping. There is a emission peak located at wavelength 1032 nm, which is the characteristic emission of Yb³⁺. The fluorescence band peaks at ~1.34 µm and covers a wide range from 1.2 to 1.6 µm. The spectrum is similar to that of Cr^{4+} :YAG crystal previously reported [2]. In our experimental conditions, the intensity of the luminescence centered at 1.34 µm was notably high and was estimated to be 1.3 times of the emission at 1032 nm. This fact supports the prospect of achieving lasing in Cr^{4+} :YbAG under 940 nm diode pumping.

In Cr^{4+} :YbAG crystal, which is closely analogous to Cr^{4+} :YAG, the Cr^{4+} ions occupy trivalent cation sites with tetragonally distorted tetrahedral coordination, and are charge compensated by substitutional Ca^{2+} ions. Accordingly, the relevant Cr^{4+} energy levels, all of which are derived from the ³F ground term of the free ion, are labeled by the irreducible representations of tetragonal group D_{2d} followed in parentheses by the corresponding irreducible representations of tetrahedral group T_d . Three such levels are shown in Fig. 4. According to the Tanabe-Sugano diagram for tetrahedrally coordinated Cr^{4+} :YAG [6], the ${}^{3}B_{1}({}^{3}A_{2}) \rightarrow {}^{3}A_{2}({}^{3}T_{1})$ transition is the only fully electric-dipole allowed transition, to the extent that the tetrahedral crystal field relaxes the parity selection rule, and is therefore a target for energy transfer since it is in resonance with the emission of Yb³⁺. The



Fig. 2. The absorption of Cr⁴⁺ in Cr⁴⁺: YbAG crystal.

Download English Version:

https://daneshyari.com/en/article/1237116

Download Persian Version:

https://daneshyari.com/article/1237116

Daneshyari.com