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# Capturing transiently charged states at the $C_{60}/TiO_2(110)$ interface by time-resolved soft X-ray photoelectron spectroscopy

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# ABSTRACT

Time-resolved soft X-ray photoelectron spectroscopy is utilized to determine an energy level alignment and the photoexcited carrier dynamics at a  $C_{60}$ /TiO<sub>2</sub>(110) interface. The interface electronic structure is characterized by a type II junction, which favors an injection of photoexcited electrons from  $C_{60}$  to TiO<sub>2</sub>. Ultraviolet (UV) laser pulse irradiation induces transient shifts of both C 1s and Ti 2p core levels towards the higher binding energies. These energy shifts are caused by a laser-induced charge transfer between the  $C_{60}$  layer and the TiO<sub>2</sub>(110) surface. Upon UV absorption, valence electrons of  $C_{60}$  are promoted to unoccupied levels, followed by a resonant transfer to TiO<sub>2</sub>, leaving  $C_{60}$  in a cationized state. On the TiO<sub>2</sub>(110) side, the electrons are injected into the conduction band to raise the carrier density so that downward bending of the TiO<sub>2</sub> band is induced. The UV-excited states of  $C_{60}$  and TiO<sub>2</sub> have sufficiently longer lifetime than the lifetime of the electron–hole pairs in solid  $C_{60}$ . The  $C_{60}$ /TiO<sub>2</sub>(110) interface is, thus, proved to be efficient for separating the electron–hole pairs generated within the  $C_{60}$  layer.

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## 1. Introduction

One of key technologies for realizing sustainable society is an efficient conversion of the sunlight into chemical and electrical energies. Titanium dioxide (TiO<sub>2</sub>) exhibits a high photocatalytic activity for water splitting to produce a zero-emission fuel, H<sub>2</sub> [1]. TiO<sub>2</sub> is also used as a transparent electrode in photovoltaic cells [2]. One of the extensively studied TiO<sub>2</sub>-based photovoltaic cells is a dye-sensitized organic cell [3–5], in which visible-light sensitive dye molecules or polymers are loaded onto the TiO<sub>2</sub> electrodes. The TiO<sub>2</sub> electrode collects photoexcited electrons from the light absorbing layer to facilitate electron—hole separation. In such a photovoltaic cell, a charge transfer across the interface is an important elemental step to define the overall performance of the cell.

Many theoretical and experimental studies have been devoted to elucidate the interface electronic structure of model photovoltaic

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ture and the charge transfer time has not been established yet. The time-resolved soft X-ray photoelectron spectroscopy (TRPES) utilizing a pump-probe method has been proved to be a powerful tool to simultaneously determine the electronic structure and the dynamics of the photoexcited carriers at semiconductor surfaces such as silicon [12–14], GaAs [9], ZnO [10,14,15] and TiO<sub>2</sub> [16] as well as at a semiconductor–semiconductor interface [9,10]. In these studies, transient changes of the surface photovoltage (SPV) effect are determined by following temporal binding-energy (BE) shifts of the core-level and/or valence-band photoemission

cells, because one can predict the charge transfer efficiency by assessing electron- and hole-injection barriers at the interface

[6-8]. On the other hand, real-time observations of the carrier

dynamics across the interface have been limited [9-11], unfortu-

nately, so that the relation between the interface electronic struc-

spectra of the semiconductor surfaces. In the present study, we apply TRPES to the  $C_{60}/\text{TiO}_2(110)$  heterojunction in order to elucidate how the excited carriers behave at the junction.  $C_{60}$  is an important component of dye-sensitized photovoltaic cells, where the  $C_{60}$  layer is used as an electron acceptor or as an electron transport layer [17,18].  $C_{60}$  also serves to enhance photocatalytic activity of TiO<sub>2</sub> by promoting the





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separation of the photoexcited electron—hole pairs [19,20]. We show that the electronic structure of the  $C_{60}$ /TiO<sub>2</sub>(110) interface is characterized by a type II junction, which favors an electron transfer from  $C_{60}$  to TiO<sub>2</sub>, and that an efficient  $C_{60} \rightarrow$  TiO<sub>2</sub> electron transfer actually occurs upon ultraviolet (UV) light irradiation.

#### 2. Experimental details

A single crystal of rutile TiO<sub>2</sub> with (110) orientation (Earth Chemical Co.) was used as a substrate. The sample surface was cleaned in an ultrahigh vacuum (UHV) chamber by cycles of Ar<sup>+</sup> sputtering (1 kV) and annealing at 950 K in Ref.  $1 \times 10^{-4}$  Pa until a sharp (1×1) low energy electron diffraction pattern was obtained. No carbon-related peak was detected in the photoemission spectra of the clean surface. C<sub>60</sub> (> 99% purity, Kanto Chemical Co.) was deposited on TiO<sub>2</sub>(110) using a home-made evaporation cell, consisted of a quart crucible around which a tungsten wire for resistive heating and a thermocouple for temperature monitoring were wound. Thoroughly out-gassed C<sub>60</sub> was sublimated at 680 K while keeping the substrate at room temperature. An average thickness of the deposited C<sub>60</sub> overlayer, estimated from intensity attenuation of

the Ti 2p core-level peak (Fig. 1b), was 0.57 nm [21], which is ~ 70% of a monolayer thickness of the  $C_{60}$  layer on TiO<sub>2</sub>(110) [22].

The TRPES measurements employing a pump-probe method were carried out at SPring-8 BL07LSU [23]. For the pump light, a second harmonic of an amplified Ti:sapphire laser pulse was used. The pulse duration and a repetition rate were 35 fs and 1 kHz, respectively. The photon energy was set at 3.06 eV, which exceeds both the band gap of rutile TiO<sub>2</sub> (3.0 eV) [24] and the energy difference between the highest occupied molecular orbital (HOMO) level and the lowest unoccupied molecular orbital (LUMO) level of C<sub>60</sub> (2.3 eV) [25,26].

For the probe light, synchrotron radiation (SR) lights with the energies of 253 eV and 600 eV were used to measure valence-band spectra and Ti 2p and C 1s core-level spectra, respectively. The SR lights were provided by a single bunch and a 11/29-filling bunch train of an H-mode operation. SR pulses from the single bunches were used for the pump-probe measurements. A width of the pulse was 50 ps, and a time interval between the pulses was 4.79  $\mu$ s. A time-of-flight electron energy analyzer (VG Scienta ARTOF 10 k) was used to acquire photoelectron spectra. The BEs of the photoelectron spectra were referenced to the Au 4f<sub>7/2</sub> core-level peak



**Figure 1.** (a) Valence-band spectra of the clean and  $C_{60}$ -covered TiO<sub>2</sub>(110) surfaces. (b) A magnified view of the spectra in the band gap region and the VBM region. The spectra in the band gap region are displayed with intensities magnified by 10 times. In the spectrum of the clean TiO<sub>2</sub> surface, an O-vacancy induced state, whose lineshape is reproduced by a Gaussian peak on a polynomial background, is observed at just below the Fermi level. The VBM positions of TiO<sub>2</sub> and the leading edge of the HOMO level of  $C_{60}$ , which are determined by extrapolating the leading edges to the base lines, are indicated by thick vertical bars. (c) Ti  $2p_{3/2}$  spectra of the clean and  $C_{60}$ -covered TiO<sub>2</sub>(110) surfaces. The intensity is reduced to 27% of that of the clean surface by  $C_{60}$  deposition. (d) An energy level diagram of the  $C_{60}/TiO_2$  heterojunction. The energy positions of  $C_{60}$  are the upper and lower edges of the HOMO and LUMO levels, respectively.

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