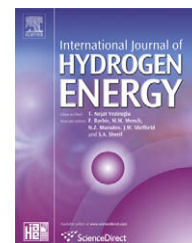


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A pilot-scale study of biohydrogen production from distillery effluent using defined bacterial co-culture

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ABSTRACT

We evaluated the feasibility of improving the scale of hydrogen (H₂) production from sugar cane distillery effluent using co-cultures of *Citrobacter freundii* O1, *Enterobacter aerogenes* E10 and *Rhodospseudomonas palustris* P2 at 100 m³ scale. The culture conditions at 100 ml and 2 L scales were optimized in minimal medium and we observed that the co-culture of the above three strains enhanced H₂ productivity significantly. Results at the 100 m³ scale revealed a maximum of 21.38 kg of H₂, corresponding to 10692.6 mol, which was obtained through batch method at 40 h from reducing sugar (3862.3 mol) as glucose. The average yield of H₂ was 2.76 mol mol⁻¹ glucose, and the rate of H₂ production was estimated as 0.53 kg/100 m³/h. Our results demonstrate the utility of distillery effluent as a source of clean alternative energy and provide insights into treatment for industrial exploitation.

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1. Introduction

Hydrogen (H₂) is a sustainable, non-polluting source of energy. Owing to its potential use, it has been considered as a prominent future fuel and energy vector. The United States and the European countries have already committed to establish an energy economy based on H₂ from fully sustainable or renewable sources. Based on the national H₂ program of the United States, the contribution of H₂ to total energy market will be ~8–10% by 2025 [1]. The market for H₂ and technologies for its production are potentially huge, and the level of investment into H₂ research has increased significantly in recent years,

and H₂ may, in the long-term, replace all other fuels. However, cost is currently the biggest impediment to H₂ fuel production. H₂ is produced from nonrenewable energy sources, such as oil, natural gas, and coal that require expensive energy input [2]. Thus, the development of cost-effective and/or economically viable processes is necessary for increasing the achievable dominance of H₂ as an energy source. As a consequence, scientists are pursuing biological systems as a way to meet increasing energy demand. Biological H₂ production stands out as an environmentally friendly process carried out under mild operating conditions that include fermentation of sugars, photolysis of water and water gas shift reactions [3–6].

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Biological H₂ production is the most challenging and exigent prefecture of renewable energy technology development, especially where it involves the treatment of organic waste. H₂ can also be produced by bacteria from renewable biomass. Several studies on biohydrogen production from various substrates, including wastes [7–10], have been reported at laboratory level. Vatsala and Seshadri [7] have studied the enzymatic hydrolysis of silk-cotton waste into H₂ and volatile fatty acids (VFAs) using *Rhodospirillum rubrum* ATCC 11170. Kapdan and Kargi [8] have published an excellent review on the use of various substrates, particularly simple sugars, corn starch wastes, sweet potato, food industrial wastes, wastewater, and waste sludge, for H₂ production under dark and phototrophic conditions. While, chemical wastewater was used successfully for biohydrogen production at lab-scale [9]. Yu et al. [10] have demonstrated the continuous production of H₂ from anaerobic acidogenesis of a high-strength rice winery wastewater by a mixed bacterial flora at 3 L upflow reactor. Various types of bioreactors with different capacities have also been used for H₂ production, including batch, fed-batch and continuous-flow, stirred tank reactors [11–14]. The concept of waste utilization through a microbial fermentation process represents an ideal alternative technology for coordinating pollutant elimination and energy production. In this sense, biohydrogen is considered an intriguing “future fuel”. The energy required for aerobic wastewater treatment is huge, but can be reduced through economically viable biohydrogen production processes.

We have been studying H₂ production from various waste materials [15,16]. Among the waste materials examined, distillery effluent was the most suitable substrate for H₂ production [17]. H₂ production from distillery effluent was investigated at lab-scale and pilot-scale [17]. The seasonal variations on the molasses quality, particularly sugar content, sulfur content, and their influences on H₂ production using individual and/or co-culture of *Citrobacter freundii* and/or *Rhodospseudomonas palustris* P2 were investigated [17]. We also reported H₂ production from distillery effluent at 10 m³ volume and the produced H₂ was subsequently tested for fuel cell application [18]. Here, the present investigation focuses on developing a cost-effective, viable process for producing clean H₂ on a pilot-scale (100 m³) from distillery effluent. Apart from H₂ as an energy source, the treatment of distillery waste was also accomplished, with a drastic reduction in chemical oxygen demand (COD) and biological oxygen demand (BOD). To the best of our knowledge, this is the first report on H₂ production at 100 m³ scale.

2. Materials and methods

2.1. Materials

The biochemical characterization kits for bacterial identification were purchased from Hi-Media (Mumbai, India). The genomic DNA isolation kit was procured from Promega (Madison, WI, USA). DNA gel purification kit was purchased from Qiagen (Mannheim, Germany). Primers were synthesized by Cosmotech Co. Ltd., Korea. All other chemicals, unless otherwise indicated, were obtained from E-Merck (Mumbai, India) and/or Sigma–Aldrich (St. Louis, MO, USA).

2.2. Microorganisms and culture conditions

Heterotrophic bacterial strains (HTB) were isolated from the effluent treatment plant, EID Parry (I) Ltd., Nellikuppam, Tamil Nadu, India. Photoheterotrophic bacterial strains (PTB) were isolated from pond soil, Chengalpet, Tamil Nadu, India. Ormerod's minimal medium [19] was used for the regular maintenance of photoheterotrophic isolates. To measure the H₂ production efficiency of the PTB isolates, experiments were conducted in serum bottles (125 ml total volume (tv); 100 ml working volume (wv)) at 37 °C, under photoheterotrophic conditions at 7000 lux intensity with an incandescent light source. Malate at 50 mM was used as a carbon source for PTB isolates. Heterotrophic isolates were cultured in H₂ production medium, which contained (per litre): 1.0 g yeast extract, 1.0 g NaCl, 1.0 g sodium thiosulphate and 10 g glucose, under dark anaerobic conditions for 24 h. This medium was fortified with a potassium phosphate buffer (50 mM; pH 7.0), and the experiments were conducted in serum bottles at 37 °C. Before inoculating the seed culture (HTB and/or PTB), the bottles were flushed with argon (Ar) gas (99.9%) for 10 min to ensure that the bottles were completely deprived of oxygen (O₂). Bottles were then sealed with butyl rubber septa and aluminum caps. Isolates were tested for H₂ production efficacy with specific media. The best H₂ producing strains, HTB01, HTB10, and PTB2, were subjected to physiochemical characterization using the methodologies described previously [20].

To further confirm the identification of bacterial isolates, the 16S rRNA genes from the strains HTB01, HTB10, and PTB2 were amplified by using the primers: 27F (5'-AGA GTT TGA TCM TGG CTC AG-3') and 1492R (5'-TAC GGY TAC CTT GTT ACG ACT T-3') and sequenced by Cosmotech, Co. Ltd., Korea. Sequence alignment and analysis of similarity of the 16S rRNA genes were performed with CLUSTAL W [21] and DNA base RC2 (v2.9.97) programs. A similarity search was done in the Ribosomal Database Project (RDP) [22].

To determine the efficiency of the selected microorganism for H₂ production, several experiments were conducted at a lab-scale with 125 ml serum bottles and a glass bioreactor (Bioengineering, KLF, 2000, SW). Axenic strains were acclimatized on distillery effluent in the laboratory under dark anaerobic conditions to grow and produce H₂ for several generations.

2.3. Bioreactors designing, fabrication and operation

The feasibility of H₂ production and effluent treatment was evaluated at the 100 m³ scale. The lab-scale process of H₂ production was then scaled-up from 1 L to 100,000 L. As shown in Fig. 1, a sequence of bioreactors was constructed for inoculum preparation for the 100 m³ reactor. The total volumes of the reactors were 0.125, 1.25, 12.5 and 125 m³, and the working volumes were set from 0.1 to 100 m³. The height-to-diameter (H/D) ratio of each reactor was set to 1.28. All the reactors were fabricated using mild steel with epoxy coating to prevent chemical and/or biological corrosion. Temperature and pH were monitored using specialized probes for 10 and 100 m³ reactors. However, temperature and pH were not controlled. A 10% inoculum was used for each reactor. The

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