



Calendar and PHEV cycle life aging of high-energy, lithium-ion cells containing blended spinel and layered-oxide cathodes

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ARTICLE INFO

Article history:

Received 23 June 2011

Received in revised form 15 August 2011

Accepted 16 August 2011

Available online 22 August 2011

Keywords:

PHEV battery testing

Performance degradation

Curve-fitting

Calendar life

Charge-sustaining cycling

Charge-depleting cycling

ABSTRACT

One hundred seven commercially available, off-the-shelf, 1.2-Ah cells were tested for calendar life and CS cycle- and CD cycle-life using the new USABC PHEV Battery Test Manual. Here, the effects of temperature on calendar life, on CS cycle life, and on CD cycle life; the effects of SOC on calendar life and on CS cycle life; and the effects of rest time on CD cycle life were investigated. The results indicated that the test procedures caused performance decline in the cells in an expected manner, calendar < CS cycling < CD cycling. In some cases, the kinetic law changed with test type, from linear-with-time to about t^2 . Additionally, temperature was found to stress the cells more than SOC, causing increased changes in performance with increasing temperature.

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1. Introduction

Lithium-ion batteries continue to attract much interest in applications where high specific or volumetric power or energy is required. High-energy lithium-ion batteries are also being considered for automotive applications by the U.S. Department of Energy-supported U.S. Advanced Battery Consortium (USABC) [1]. The batteries usually consist of a metal-oxide positive electrode, a carbon negative electrode, and an organic electrolyte containing dissolved lithium salts.

Layered-oxide cathodes, such as $\text{Li}(\text{Ni}_{1/3}\text{Mn}_{1/3}\text{Co}_{1/3})\text{O}_2$, represent good candidates for automotive applications because of their high specific capacity [2]. However, the cycle life of this oxide was not as high as desired [3]. On the other hand, spinel oxides, such as LiMn_2O_4 , are also viable candidates. These oxides are low-cost and have fast kinetics, which makes the oxide suitable for high-power applications [2]. The spinels were reported to have lower specific capacity than the layered oxides [2] and to degrade rapidly due to manganese dissolution [3–6]. There have been reports in the literature that blending these two materials produced a composite with the benefits of both [7–12].

Two U.S. Department of Energy national laboratories, Argonne National Laboratory (ANL) and Idaho National Laboratory (INL), continue to collaborate to understand the causes of performance decline in lithium-ion batteries. Results from this collaboration using three positive electrode materials— $\text{LiNi}_{0.8}\text{Co}_{0.2}\text{O}_2$, $\text{LiNi}_{0.8}\text{Co}_{0.1}\text{Al}_{0.1}\text{O}_2$, and $\text{Li}_{1.05}(\text{Mn}_{1/3}\text{Co}_{1/3}\text{Ni}_{1/3})_{0.95}\text{O}_2$ —are given in Refs. [13–18].

The procedures outlined in the USABC test manuals [19–21] are intended to show the promise of a technology versus a set of performance and cost targets. No knowledge of the actual battery chemistry is needed. Thus, the evaluation of cells concentrates on their performance and life and how their life is affected by SOC, time, temperature and type of test. Calendar and cycle life tests were performed to determine the aging characteristics of the blended cathode material under plug-in hybrid electric vehicle (PHEV) testing conditions [19]. It should be noted that the calendar tests in the PHEV and hybrid-electric vehicle (HEV) [20] manuals are essentially the same. The cycle life test is more complex, depending on what is to be learned. Instead of just one cycling mode of operation, charge-sustaining, as in an HEV, the PHEV operates in charge-depleting (CD) as well as charge-sustaining (CS) modes, as shown schematically in Fig. 1. In the CD mode of operation, the battery powers the vehicle directly; the internal-combustion engine is not used at all. After a while, the battery energy becomes exhausted. At this low state of charge, the PHEV will operate in CS mode, similar to that of a hybrid-electric vehicle.

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