



Short communication

A novel bilayered $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3/\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ interconnector for anode-supported tubular solid oxide fuel cell via slurry-brushing and co-sintering process

Yanjie Xu, Shaorong Wang*, Renzhu Liu, Tinglian Wen, Zhaoyin Wen

CAS Key Laboratory of Materials for Energy Conversion, Shanghai Institute of Ceramics, Chinese Academy of Sciences (SICCAS), 1295 Dingxi Road, Shanghai 200050, PR China

ARTICLE INFO

Article history:

Received 29 June 2010

Received in revised form 27 July 2010

Accepted 27 July 2010

Available online 6 August 2010

Keywords:

Tubular solid oxide fuel cell

Anode-supported

Interconnector

Slurry-brushing

Co-sintering

ABSTRACT

Considering that conventional lanthanum chromate (LaCrO_3) interconnector is hard to be co-sintered with green anode, we have fabricated a novel bilayered interconnector which consists of La-doped SrTiO_3 ($\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3$) and Sr-doped lanthanum manganite ($\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$). $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3$ is conductive and stable in reducing atmosphere, locating on the anode side; while $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ is on the cathode side. A slurry-brushing and co-sintering method is applied: the $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3$ and $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ slurries are successively brushed onto green anode specimen, followed by co-firing course to form a dense bilayered $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3/\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ interconnector. For operating with humidified hydrogen and oxygen at 900°C , the ohmic resistances between anode and cathode/interconnector are $0.33\ \Omega\ \text{cm}^2$ and $0.186\ \Omega\ \text{cm}^2$, respectively. The maximum power density is $290\ \text{mW}\ \text{cm}^{-2}$ for a cell with interconnector, and $420\ \text{mW}\ \text{cm}^{-2}$ for a cell without it, which demonstrates that nearly 70% of the power output can be achieved using this bilayered $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3/\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ interconnector.

© 2010 Elsevier B.V. All rights reserved.

1. Introduction

Solid oxide fuel cells (SOFC) represent an advanced technology for clean, high efficient and reliable energy conversion, which has attracted more and more attentions worldwide [1–3]. Tubular SOFC have many advantages such as the ease of sealing, the ability to endure the thermal stress caused by rapid heating [3–6]. However, interconnector still remains to be a main challenge holding back single fuel cells from being compiled into stacks [7,8]. Interconnector provides the conductive path for electrical current to pass from the anode of one cell to the cathode of the next one, as well as separating the fuel gas from the oxidant. So it should be completely dense, sufficiently conductive, and quite stable in both oxidizing and reducing atmospheres. Due to these tough requirements, only few materials can be adequate for SOFC interconnector. Lanthanum chromate (LaCrO_3) based perovskite materials are the most promising one which have high thermal and chemical stability in dual oxidation-reduction atmosphere as well as good electrical conductivity, thus have been extensively investigated [9–12]. However, their poor sintering ability makes them unable to be co-sintered with green tubular anode, resulting expensive cost for manufacturing.

This communication reports our efforts to prepare a novel bilayered $\text{Sr}_{0.6}\text{La}_{0.4}\text{TiO}_3/\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ interconnector via co-sintering with green tubular anode. La-doped SrTiO_3 (SLT) is studied by some researchers for SOFC anode material due to its good electrical conductivity and stabilization in reducing atmosphere [13–16]. Thus we employ it to be connected with NiO + YSZ anode. $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ (LSM) is on the cathode side due to its high electrical conductivity and good stabilization in oxidation atmosphere. The SLT and LSM slurries were successively brushed onto green anode specimen, followed by co-firing course to form a dense bilayered SLT/LSM interconnector. Microstructure and combination between the bilayered interconnector and the anode were studied by SEM, AC impedance spectra and current–voltage (I – V) curves were employed to determine the properties of single cells with and without interconnector.

2. Experimental

2.1. Powder and slurry preparation

To prepare SLT powder by solid-state reaction method, stoichiometric ratio of SrCO_3 , La_2O_3 and TiO_2 was mixed by ball-milling. After that, the mixture was dried, die-pressed, and heated at 1200°C for 5–8 h. Then, the powder was ball-milled again for further use. LSM powder was provided by Fuel Cell Materials Corporation.

* Corresponding author. Tel.: +86 21 52411520; fax: +86 21 52411520.
E-mail address: srwang@mail.sic.ac.cn (S. Wang).

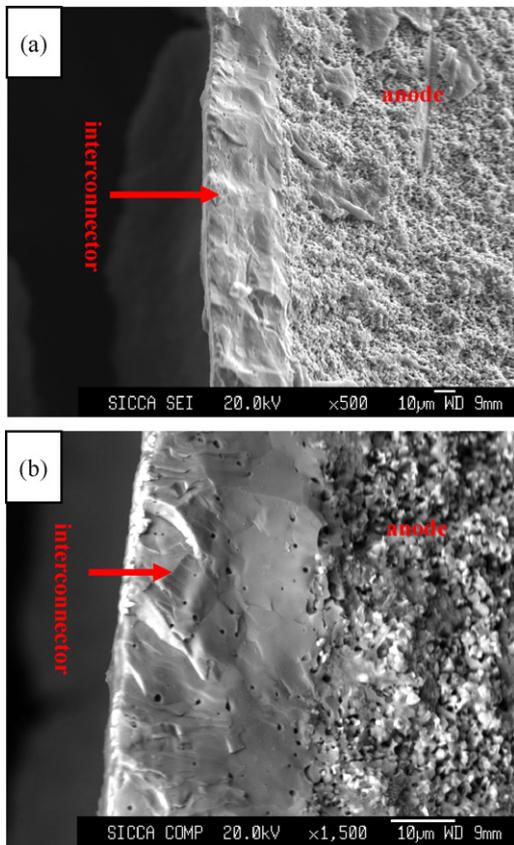


Fig. 1. Cross-section SEM micrographs of (a) co-sintered interconnector and anode, (b) enlarged (a).

To prepare the slurries for brushing, the powders of SLT and LSM were mixed with an appropriate binder system (i.e. ethanol, PVB, triethanolamine), respectively. The triethanolamine (TEA) was helpful for the powders to disperse homogeneously, with the mass percentage of 3% to the powders; while PVB was added in order to enhance the viscosity of slurries, with the mass percentage of 2% to the powders. The mass ratio of powder to ethanol was 1:2 for SLT slurry, and 1:1.5 for LSM slurry. Then the two slurries were ball-milled for about 3 h, respectively, followed by vacuum pumping for several minutes to eliminate air bubbles.

2.2. Slurry-brushing process

Firstly, green tubular anode consisting of NiO and YSZ was fabricated via dip-coating process (details can be found elsewhere [17]). Secondly, a SSZ (Sc doped ZrO_2 , from Daiichi Kigenso Kagaku Kogyo, Japan) electrolyte layer was coated onto green tubular anode by dip-coating too. Thirdly, erase a rectangular area of SSZ along the axial direction by $0.6\text{ cm} \times 4.5\text{ cm}$, in order to brush SLT slurry onto the exposed anode area.

Then, SLT slurry was brushed onto green anode by a brush pen, until the SSZ-free area was completely covered by SLT slurry, especially the boundary of green anode and SSZ electrolyte. Otherwise, fuel gas or oxygen could penetrate through the boundary which would reduce the open circuit voltage of SOFC (OCV). The LSM slurry was brushed onto the SLT layer subsequently. Finally, the green tubular anode with SSZ electrolyte and bilayered SLT/LSM were co-fired at 1400°C for 3–5 h.

2.3. Cathode fabrication

The LSM/SSZ composite cathode was fabricated by dip-coating method. 70 g LSM and 30 g SSZ were mixed with 60 g methyl ethyl

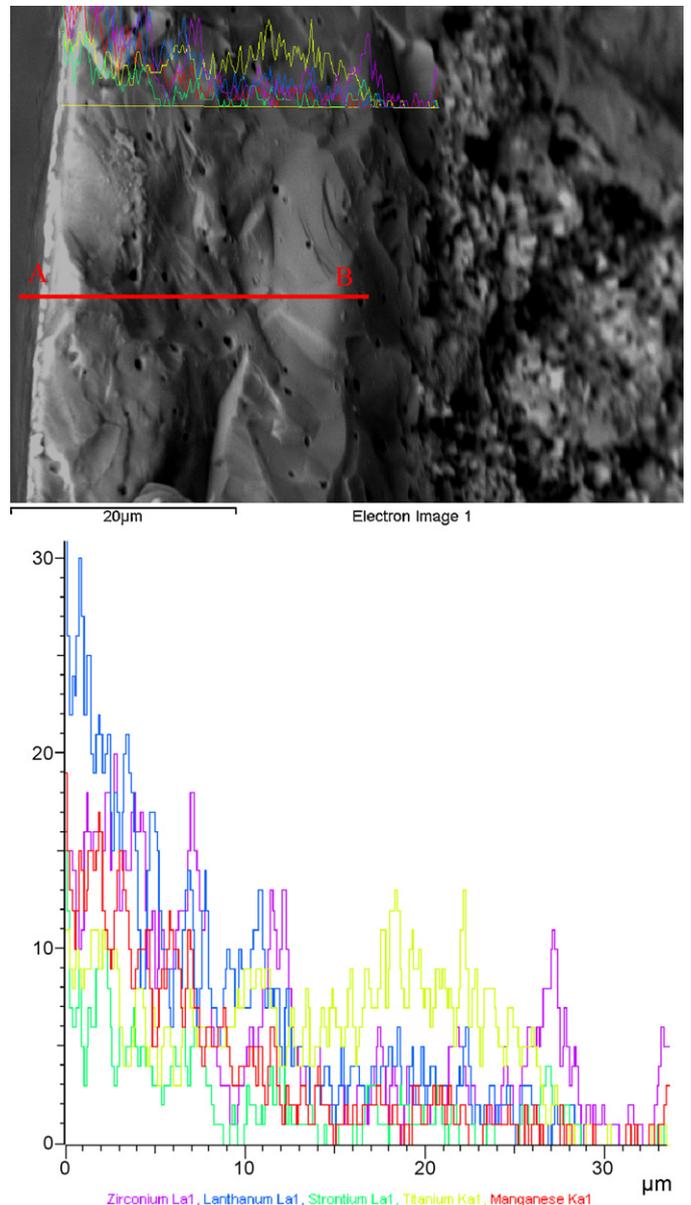


Fig. 2. EDS line-scan images through cross-section of co-sintered interconnector and anode.

ketone (MEK) and 50 g ethanol by ball-milling for 1–2 h, using 5 g TEA as dispersant. Then, 8 g PVB, 10 g polyethylene glycol and 10 g dibutyl-O-phthalate were added into the slurry in order to enhance the viscosity and avoid the coating cracking. The slurry was ball-milled for another 1–2 h, followed by vacuum pumping for several minutes to eliminate air bubbles. Before dip-coating, the interconnector was covered by adhesive tape, in order to separate it from LSM/SSZ cathode slurry. There was an empty area between cathode and interconnector serving for insulating. Otherwise, cathode would be connected with anode via interconnector, leading to inner short-circuit. The cathode was heated at 1200°C for 3 h, and a single tubular SOFC with interconnector was obtained. The diameters of tubular cells were about 11.16 mm, and the length of cathode was about 40–60 mm which depended on the dip-coating method.

2.4. Cell performance test

Tubular SOFC tests were carried out by four-probe method. There were two Pt-wires fixed on anode, cathode and intercon-

Download English Version:

<https://daneshyari.com/en/article/1289629>

Download Persian Version:

<https://daneshyari.com/article/1289629>

[Daneshyari.com](https://daneshyari.com)