

A divalent cadmium coordination polymer featuring a complex tetranodal 3-D topology and hexadecameric water clusters

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ABSTRACT

The luminescent coordination polymer $\{[\text{Cd}_2(\text{dms})_2(\text{bpmp})_{1.5}] \cdot 8\text{H}_2\text{O}\}_n$ (**1**, dms = 2,2-dimethylsuccinate, bpmp = bis(4-pyridylmethyl)piperazine) exhibits $[\text{Cd}_2(\text{dms})_2(\text{bpmp})]_n$ layers partially pillared by other crystallographically distinct bpmp ligands into a complex 3-D 3,4,5-connected tetranodal network. Hydrogen-bonded hexadecameric water molecule clusters lie in incipient channels.

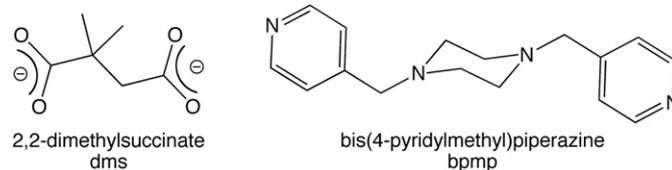
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Divalent metal dicarboxylate coordination polymers have been widely studied in recent years, because of their potentially useful properties such as gas storage [1], gas separation [2], ion exchange [3], catalysis [4], non-linear optics [5], and luminescence [6]. Additionally, the aesthetics of their underlying network topologies provides a significant impetus for continued exploratory synthesis. Synergistic cooperation between metal coordination geometry, specific carboxylate binding and bridging modes, and the geometric disposition of carboxylate donor groups during self-assembly is reflected in the wide variety of coordination polymer topologies that have been observed to date [7]. Given the challenges inherent in *a priori* prediction and design of specific networks, exploratory synthesis continues to develop the scope of coordination polymer topologies. A recent successful approach towards the generation of new topologies has been the simultaneous employment of aliphatic α,ω -dicarboxylate dicarboxylate and dipyrindyl tethering ligands. For example, $\{[\text{Cd}(\text{succinate})(N,N'-bispyridin-4-ylmethylsuccinamide)] \cdot \text{H}_2\text{O}\}_n$ possesses a very rare 6⁶ self-penetrated layered topology [8], and $\{[\text{Cd}(\text{succinate})(\text{bpmp})(\text{CH}_3\text{OH})] \cdot 2\text{H}_2\text{O}\}_n$ (bpmp = bis(4-pyridylmethyl)piperazine) manifests a chiral layered network with a unique 6⁵8 topology [9]. In light of the unprecedented topologies of these two succinate/long-spanning dipyrindine dual-ligand cadmium coordination polymers, we aimed to examine the effect of *gem*-dimethyl group substitution on the underlying topology. We

present here a discussion of the synthesis, structural characterization, luminescence, and thermal properties of $\{[\text{Cd}_2(\text{dms})_2(\text{bpmp})_{1.5}] \cdot 8\text{H}_2\text{O}\}_n$ (**1**, dms = 2,2-dimethylsuccinate, Scheme 1), which manifests a complex 3-D tetranodal network topology. The water molecules of crystallization in **1** also form an intriguing hydrogen-bonded cluster morphology.

Colorless blocks of compound **1** were prepared in moderate yield from the hydrothermal reaction of cadmium nitrate, H_2dms and bpmp in a 1:1:2 mol ratio [10]. It is plausible that some quantity of bpmp in the reaction mixture acts to deprotonate the H_2dms precursor, and is therefore unable to bind to cadmium. The infrared spectrum of polycrystalline **1** indicated the presence of antisymmetric and symmetric carboxylate stretching bands at 1599/1565 cm^{-1} and 1426/1412 cm^{-1} , respectively. A moderately intense, broad band centered on 3396 cm^{-1} marks the presence of the hydrogen-bonded water molecules of crystallization.

Single-crystal X-ray diffraction revealed that the asymmetric unit of **1** (Fig. S1) contains two divalent cadmium atoms, two dms ligands (*dms-A*, O1–O4; *dms-B*, O5–O8), halves of three crystallographically distinct bpmp ligands (*bpmp-A*, N1–N2; *bpmp-B*, N3–N4; *bpmp-C*, N5–N6), and



Scheme 1. Ligands used in this study.

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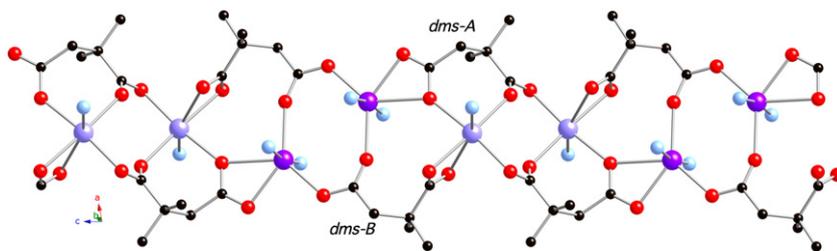


Fig. 1. A $[\text{Cd}_2(\text{dms})_2]_n$ chain motif in **1**. Cd1 and Cd2 atoms are shown in light and dark purple, respectively.

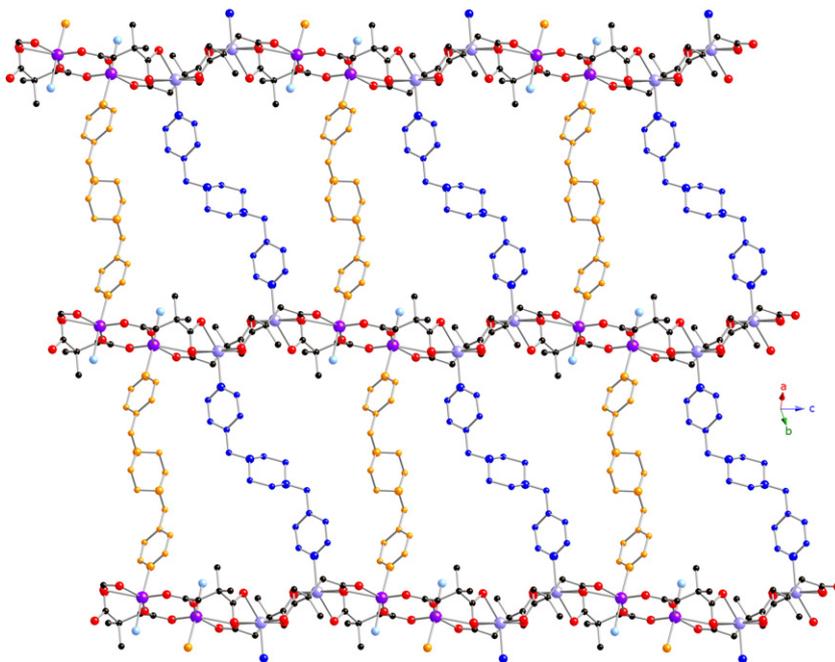


Fig. 2. A $[\text{Cd}_2(\text{dms})_2(\text{bpmp})]_n$ layer in **1**, with $[\text{Cd}_2(\text{dms})_2]_n$ chains linked by *bpmp-A* and *bpmp-B* ligands, which are drawn in blue and orange, respectively, in the online version of this article.

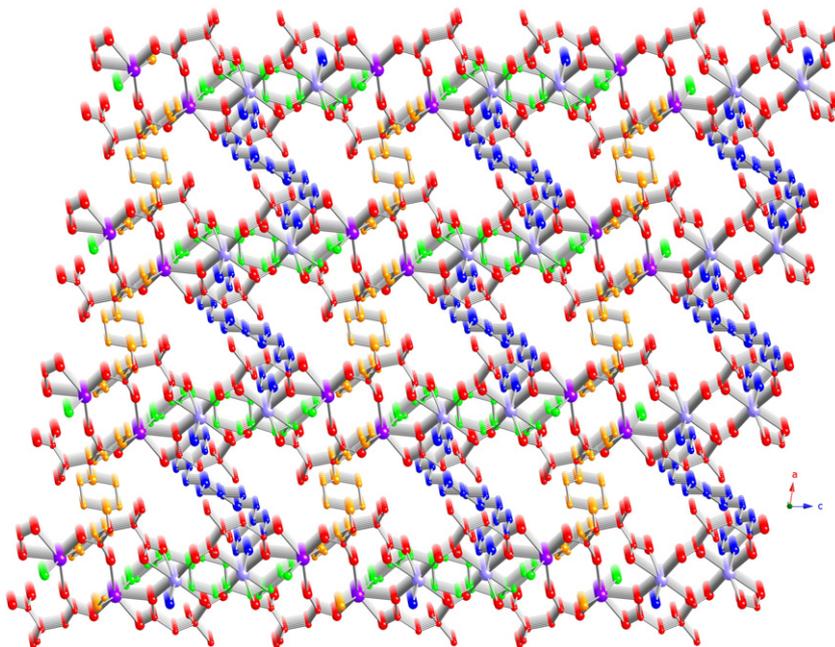


Fig. 3. Three-dimensional $[\text{Cd}_2(\text{dms})_2(\text{bpmp})_{1.5}]_n$ coordination polymer network in **1**, viewed down the *b* crystal direction. The *dms*, *bpmp-A*, *bpmp-B*, and *bpmp-C* ligands are drawn in red, blue, orange, and green, in the online version of this article.

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