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Cadmium(II) mediated addition of methanol to nitrile-functionalized 1,4,7-triazacyclononanes



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ABSTRACT

The nucleophilic addition of methanol to two nitrile-functionalized triazamacrocyclic derivatives 1,4,7-tris(cyanomethyl)-1,4,7-triazacyclononane (L¹) and 1,4,7-tris(cyanoethyl)-1,4,7-triazacyclononane (L^2) were systematically studied under the mediation of various Cd^{2+} salts with different counter anions. When L^1 was reacted with Cd(OAc)₂·2H₂O in refluxing methanol for 4 h at a ratio of 1:1, complete solvolysis of three pendant –CH₂CN groups of L¹ was achieved and then a trigonal prismatic complex $[Cd(L^3)]^{2+}$ with the resulting imino ether ligand L^3 was in situ produced and crystallized upon the addition of stoichiometric NaClO₄. After raising the molar ratio of $Cd(OAc)_2 \cdot 2H_2O/L^1$ to 1.5:1 or higher and shortening the reaction time to 2 h, L^1 was methanolyzed selectively into L^5 , incorporating two $-CH_2C(OMe)=NH$ and one $-CH_2CN$ pendant arms because an acetate can still bind to Cd^{2+} and limits the conversion of the remaining nitrile arm. By adding NaClO₄, a mono-capped trigonal prismatic complex [Cd(L⁵) (OAc)](ClO₄) was easily precipitated from the methanolic solution. In contrast to L¹, perchlorate, nitrate, and acetate salts of Cd^{2+} were found to promote the transformation of L^2 bearing longer – CH_2CH_2CN pendant arms into the entire solvolysis product L^4 after refluxing for 4 h, which utilizes its three $-CH_2CH_2$ -C(OMe)=NH groups and tacn (tacn = 1,4,7-triazacyclononane) backbone to capture Cd^{2+} yielding an octahedral complex $[Cd(L^4)]^{2+}$. Additionally, L¹ and L² did not undergo methanolysis when treated with $CdCl_2 \cdot 4H_2O$ and two neutral compounds $[Cd(L^1)Cl_2]$ and $[Cd(L^2)Cl_2] \cdot 0.5H_2O$ containing the original nitrile ligands were obtained. The latter was recrystallized from a DMF/CH₃CN solution to afford an octahedral solvent-coordinated complex [Cd(L²)(DMF)Cl₂]·CH₃CN.

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1. Introduction

Polyazamacrocycles with diversified functional pendant arms are quite pursued by researchers in recent years because of their wide uses in enzyme mimics, medicinal diagnostics and catalyzed oxidations [1–10]. Beside direct attachment of the pendant functional groups onto the secondary amine N atoms of the parent azamacrocycles [11–13], addition of nucleophilic reagents to the nitrile-functionalized polyazamacrocycles in the presence of metal ions has been found to be a convenient means to acquire azamacrocyclic derivatives bearing pendant amide, imino ether, amidine and heterocycle groups [14–19]. The metal-catalyzed or metalmediated transformation shows some advantages over the traditional acid or base catalyzed ways, such as the moderately mild reaction conditions, without treatment of the troublesome acid or base byproducts, stopping the transformation at a special stage [20,21]. In our previous published investigations, we have found a tri-substituted N-nitrile functionalized 1,4,7-triazacyclononane is stepwise hydrolyzed into the respective tri-amide species in the wet acetonitrile under the mediation of Zn²⁺ [22]. Tei et al. have reported that nitrile pendant arm derivatives of tacn upon chelating to Cu²⁺ may undergo the solvolysis in refluxing methanol to afford triazamacrocyclic compounds bearing imino ether functionalities [23]. Their study revealed that selective addition of methanol to one or two of three pendant nitrile groups can be realized by choosing the starting Cu²⁺ salts with different counter anions. However, in such a Cu²⁺-containing system, complete solvolysis of all three nitrile groups attached to azamacrocycles was not fulfilled. The further studies about the nucleophilic addition of nitrile-functionalized tacn rings under the mediation of metals are necessary in consideration of the demands of looking for the convenient chemical procedures to obtain tacn derivatives with homo- or hetero- functional pendant arms.









Scheme 1. Summary of triazamacrocyclic ligands with different pendant groups.



Scheme 2. Reactions of ligand L¹ and various Cd²⁺ salts with different counter anions.

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