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#### Graphical Abstracts/J. Fluorine Chem. 185 (2016) v-x

#### Recycling hydrofluoric acid in the nuclear industry: The OverAzeotreopic Flash process (OVAF)

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is highly stable to potential fluctuations.

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A flow-sheet is proposed to recycle fluorine in the uranium enrichment process.
It avoids handling large azeotropic flows of corrosive hydrofluoric acid.



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## The addition of polyfluoroalkanesulfenic acids to alkenes



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• In the addition of polyfluoroalkanesulfenic acids to various carbon–carbon double bonds has been investigated. • Both Markovnikov adduct and anti-Markovnikov adduct could be obtained with different alkenes. • The excellent regioselectivity was obtained.





### Electrochemical oxidation of graphite in aqueous hydrofluoric acid solution at high current densities

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• Electrochemical oxidation of graphite was performed at high current densities in 47% HF aqueous solution. • Covalent C—F bonding formed above 2.4 V vs Pb/PbF<sub>2</sub>. • Stage 1 type oxygenated fluorine–graphite intercalation compound with an interlayer spacing of 0.55 nm was obtained. • This material showed a high capacity of 550 mAh/g as a cathode of lithium primary battery.



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of new 3,3,4,4-tetrafluoropyrrolidine-2,5-dithione was performed. • The synthesis method of new fluorinated N-thioacyl derivatives of  $\alpha$ -amino acids esters was proposed.

# Theoretical and experimental studies for preparing 1, 1-dibromo-<br/>1,2,2,2-tetrafluoroethane on gas-phase bromination of 1,1,1,2-<br/>tetrafluoroethaneJ. Fluorine Chem., 185 (2016) 91Ruzhu Hu<sup>a</sup>, Chengping Zhang<sup>a</sup>, Feiyao Qing<sup>a</sup>, Hengdao Quan<sup>b</sup> $r \rightarrow f_{abc}$ $r \rightarrow f_{abc}$

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• 1, 1-dibromotetrafluoroethane was first synthesized from gas-phase bromination of HFC-134a comprehensively. • The impact of reaction conditions in gas-phase bromination of HFC-134a has been developed. • The possible reaction pathways were proposed and were verified using the density functional theory. • This process provided a potential method to prepare CF<sub>3</sub>CFBr<sub>2</sub>.





proceeded smoothly. • Co/diamine-catalyst showed the satisfactory activity in this reaction. • The choice of diamine ligand and solvent are very important for excellent yields.

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