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# Graphical Abstracts/J. Fluorine Chem. 166 (2014) v-x

# HBF<sub>4</sub>·OEt<sub>2</sub>: An efficient fluorinated acid catalyst for the one-pot synthesis of secondary and tertiary *N*-homoallylic carbamates

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- $\bullet$   $\mathsf{HBF}_4.\mathsf{OEt}_2$  is used as a catalyst for the first time in this reaction.
- HBF<sub>4</sub>.OEt<sub>2</sub> is used as superior alternative to metal triflates and other catalysts.
- ullet The broad substrates compatibility of  ${\rm HBF_4.OEt_2}$  makes this method more useful. ullet This method is operationally simple and easily scalable. ullet A mechanism is also proposed by showing the catalytic involvement of  ${\rm HBF_4.OEt_2.}$

# Quantum chemical calculation of <sup>19</sup>F NMR chemical shifts of trifluoromethyl diazirine photoproducts and precursors

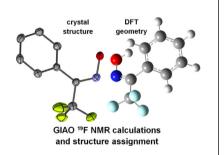
Björn Raimer<sup>a</sup>, Peter G. Jones<sup>b</sup>, Thomas Lindel<sup>a</sup>

<sup>a</sup>Institute of Organic Chemistry, TU Braunschweig, Hagenring 30, 38106 Braunschweig, Germany <sup>b</sup>Institute of Inorganic and Analytical Chemistry, TU Braunschweig, 38106 Braunschweig, Germany

• Quantum chemical prediction of  $^{19}$ F NMR spectra. • E/Z assignment of oximes via  $^{19}$ F NMR spectroscopy. •  $^{19}$ F NMR reference table of  $CF_3$  groups associated with photoaffinity labeling.

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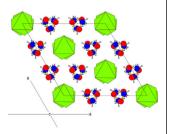
# Synthesis and characterization of new hydroxylammonium fluoromanganates and fluoroscandates

Matjaž Kristla, Brina Dojerab, Nuša Hojnika, Amalija Golobičc

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• New hydroxylammonium fluorometallates with the formulas  $(NH_3OH)_2MnF_4$  and  $(NH_3OH)_3ScF_6$ , were prepared in aqueous HF. •  $(NH_3OH)_2MnF_4$  crystallizes monoclinic,  $P2_1/c$ , the structure consisting of  $[MnF_4^2]n$  anionic layers perpendicular to a axis. •  $(NH_3OH)_3ScF_6$  crystallizes trigonal, R3c, with isolated  $ScF_6$  octahedra, surrounded by hydroxylammonium cations. • Both compounds decompose during thermal analysis in several steps, yielding corresponding fluorides  $(MnF_2$  and  $ScF_3$ ).

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# Surface fluorination effects on TiAl particle oxidation resistance

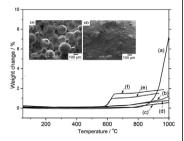
Shigeyuki Kimura<sup>ab</sup>, Fumihiro Nishimura<sup>c</sup>, Jae-Ho Kim<sup>b</sup>, Susumu Yonezawa<sup>bc</sup>, Masayuki Takashima<sup>c</sup>

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cHeadquarters for Innovative Society-Academia Cooperation, University of Fukui, 3-9-1 Bunkyo, Fukui 910-8507, Japan

• Surface fluorination of TiAl alloy particles was conducted at 25–200 °C with F₂ gas. • Below 125 °C, an oxyfluoride layer on TiAl particles was made from the oxide layer. • At more than 150 °C, the fluoride layer on TiAl was made from the oxyfluoride layer. • Fluorinated TiAl oxidation resistance can be 10 times that of untreated material.

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# Studies of the products from the reactions of co-acids containing concentrated HF and dilute HNO<sub>3</sub> with Zircaloy-4

H.F. Guab, L.F. Zhangab, M.Y. Liab, L. Wangc, X.N. Lic, S.Q. Wuab, S. Pengc, B. Gaoc, G.P. Lia

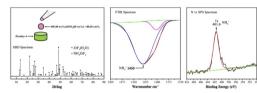
<sup>a</sup>Institute of Metal Research, Chinese Academy of Sciences, 72 Wenhua Road, Shenyang 110016, People's Republic of China

<sup>b</sup>University of Chinese Academy of Sciences, 19 Yuquan Road, Beijing 100049, People's Republic of China cState Nuclear BaoTi Zirconium Industry Company, Baoji 721003, People's Republic of China

• The co-acid containing HF (40 wt.%) and HNO3 (65 wt.%) with four volume ratios, 100:0, 94:6, 89:11 and 80:20, was used. • The products from the reactions of HF-HNO<sub>2</sub>

system with Zircalov-4 were identified by XRD. FTIR and XPS synthetically. • The usage of concentrated HF solution leads to the formation of crystalline ZrF<sub>4</sub>(H<sub>2</sub>O). • The usage of co-acids containing HF and HNO<sub>3</sub> leads to the formation of crystalline ZrF<sub>4</sub>(H<sub>2</sub>O) and NH<sub>4</sub>ZrF<sub>5</sub>.

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## Coupling of trifluoromethyl isocyanide ligands in binuclear iron carbonyl complexes

Guoliang Liab, Lihua Liua, Jing Wanga, Qian-shu Lia, Yaoming Xiec, R. Bruce Kingac

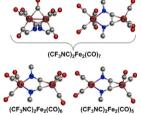
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Department of Chemistry and Center for Computational Chemistry, University of Georgia, Athens, GA 30602, USA

•  $(CF_3NC)_2Fe_2(CO)_n$  (n = 7, 6, 5) and  $(CF_3NC)Fe(CO)_n$  (n = 4, 3) have been studied by DFT. • Coupling of  $CF_3NC$ ligands is predicted to occur in  $(CF_3NC)_2Fe_3(CO)_n$  (n = 6, 5).  $\bullet$   $(CF_3NC)Fe(CO)_4$  has nearly degenerate equatorial and axial trigonal bipyramidal isomers.

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## Regioselective synthesis and through-space <sup>13</sup>C–<sup>19</sup>F spin-spin coupling NMR of new tetracyclic 3-(trifluoromethyl)-spiro(chromen[4,3-c]pyrazole-4,1'-cycloalkanes)

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OMe O NH<sub>2</sub>NHR<sup>1</sup>, EtOH, reflux, 24 h
$$R^{1} = Ph, Me; Z = CH2, O; n = 1 - 3$$

• TFAA in the acylation of spiro[chroman-2,1′-cycloalkan]-4-ones (Kabbe's adducts). • 3-(Trifluoromethyl)-spirochromen[4,3-c]pyrazole-4,1'-cycloalkanes are synthesized. ● Through-space <sup>13</sup>C-<sup>19</sup>F spin-spin coupling NMR is presented and discussed. ● Structures solved by <sup>1</sup>H, <sup>13</sup>C and <sup>19</sup>F NMR (spin-spin coupling), MS and X-ray data. ● <sup>19</sup>F NMR assigned the trifluoromethyl group position at pyrazole derivatives.

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