



# Pd-catalyzed direct arylation of phenylpyrazole: Synthesis of fipronil derivatives with aryl boronic acids promoted by a stoichiometric amount of NIS

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## ARTICLE INFO

### Article history:

Received 21 December 2011

Received in revised form 7 February 2012

Accepted 9 February 2012

Available online 21 February 2012

### Keywords:

Palladium-catalyzed

Direct arylation

Phenylpyrazole

Fipronil derivatives

Trifluoromethyl group

## ABSTRACT

The palladium-catalyzed direct arylation of phenylpyrazole with aryl boronic acid promoted by a stoichiometric amount of NIS has been reported. Several phenyl pyrazoles, especially for those with trifluoromethyl groups, can participate in the reaction, providing a series of fipronil derivatives of 4-aryl-phenylpyrazole with potential bioactivity in moderate to good yields. All the compounds were characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR and HRMS spectroscopic techniques.

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## 1. Introduction

Phenyl pyrazole is an important class of heterocyclic compounds and has shown a wide range of biological activities [1]. In addition, phenyl pyrazole derivatives possessing fluorine-containing groups, for example, fipronil (**I**, Fig. 1) [5-amino-1-[2,6-dichloro-4-(trifluoromethyl)-phenyl]-4-trifluoromethylsulfinyl-1H-pyrazole-3-carbonitrile] and ethiprole (**II**, Fig. 1), with 4-EtSO replacing with the 4-CF<sub>3</sub>SO, are effective against a host of insect pests of crops including grass hoppers, boll weevils, rice insects, termites, houseflies, fruitflies and thrips [2]. The new analog of desulfinyethiprole (**III**, Fig. 1) was surprisingly found to have high insecticidal activity [3]. So we believe that 4-aryl-1-phenylpyrazole compounds (**IV**, Fig. 1) may have the same high insecticidal activity. However, there are relatively few reports for the synthesis of compounds **IV** [4], which was synthesized mainly via the reaction of 4-iodo-1-phenylpyrazole with arylboronic acid [5]. This method requires to prepare substrates to form the electrophiles, and the overall process is neither atom-economical nor green. Therefore, the protocol for novel procedures for the

synthesis of 4-aryl-1-phenylpyrazoles containing CF<sub>3</sub> group is still a great challenge.

Recently, great interests have been aroused to develop mild methods for the direct functionalization of C–H to construct C–C [6], C–N [7], C–O [8] and C–S [9] bonds. It is well known that Suzuki–Miyaura reaction is one of the most attractive methods to construct C–C bond [10]. With the development of the Suzuki–Miyaura reaction, boronic acids have been developed as powerful reagents owing to their nontoxicity, stability and compatibility with most functional groups [11]. So we envisioned that 4-aryl-1-phenylpyrazoles can be obtained through the metal-catalyzed direct arylation of phenylpyrazoles with aryl boronic acids. And it is very fortunate that the direct arylation of phenylpyrazole occurred smoothly using PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> as catalyst. Here, we describe the novel synthesis of 4-aryl-1-phenylpyrazole derivatives containing CF<sub>3</sub> group. All the compounds were characterized by <sup>1</sup>H NMR, <sup>13</sup>C NMR and HRMS spectroscopic techniques.

## 2. Results and discussion

We choose the reaction of 5-amino-1-[2,6-dichloro-4-trifluoromethylphenyl]-1H-pyrazole-3-carbonitrile (**1a**) with phenyl boronic acid (**2a**) as a model system to determine the optimal reaction conditions (Table 1). Initially, a series of Pd catalysts, including PdCl<sub>2</sub>, Pd(OAc)<sub>2</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub> and PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>, were tested for the reaction using 2 equiv NaHCO<sub>3</sub> as base, 1 equiv NIS as iodine source, C<sub>2</sub>H<sub>5</sub>OH:H<sub>2</sub>O (3:1) as solvent at 80 °C under N<sub>2</sub> (entries 1–4). Among these catalysts screened, PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> showed the

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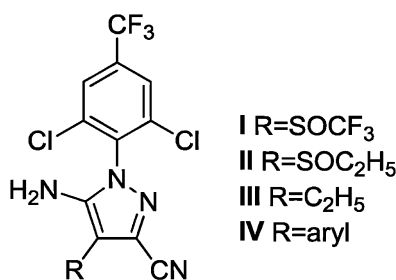


Fig. 1. Fipronil (I) and analogs (II, III, IV).

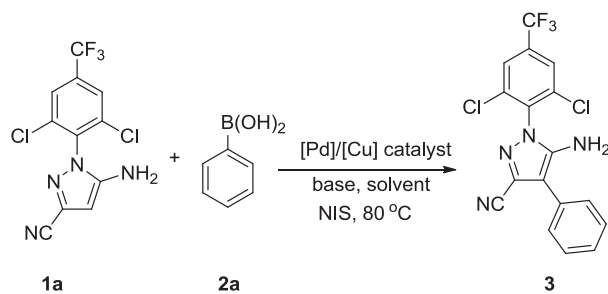
highest efficiency. However, when the reaction was conducted under atmospheric condition, the yield was decreased to 54% (entry 3). Other transition-metal catalyst, Cu(OAc)<sub>2</sub> was also measured and the byproduct of 4-iodo-1-phenylpyrazole was found to be obtained in a high yield (entry 5). Next, NaHCO<sub>3</sub> was replaced with different bases, such as CH<sub>3</sub>ONa, Cs<sub>2</sub>CO<sub>3</sub>, K<sub>2</sub>CO<sub>3</sub>, t-BuOK, Na<sub>2</sub>CO<sub>3</sub> and CH<sub>3</sub>COOK, the results indicated that the identity of base was critical to the success of the direct arylation (entries 6–11). Subsequently, when the amount of NaHCO<sub>3</sub> was decreased to 1.5 equiv, the yield of **3** was reduced to 50% (entry 3). Promoted by the help from NIS, other iodide reagents, such as I<sub>2</sub> or ICl, were investigated in the presence of PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> and they are less effective than NIS (entries 12–13). Based on our previous work [12], we think NIS promoted the generation of 4-iodo-1-phenylpyrazole in situ compared to I<sub>2</sub> and ICl, which was then reacted with phenyl boronic acid to give the arylated product **3**. It is noteworthy the slightly excess of NIS affected the yield slightly,

but only 75% yield of **3** was obtained when decreasing the amount of NIS to 0.9 equiv (entries 14–15). It was interesting that no arylated product but only the by-product 4-iodo-1-phenylpyrazole can be obtained in high yields (entries 16–17) when the mixture solvent of C<sub>2</sub>H<sub>5</sub>OH:H<sub>2</sub>O was replaced with C<sub>2</sub>H<sub>5</sub>OH and dioxane.

With the optimal catalytic system in hand, the scope of boronic acids in the reaction with **1a** was next investigated (Table 2). Gratifyingly, aryl boronic acids with electron-rich and electron-poor groups were successfully converted to the corresponding arylated products in moderate to good yields. Generally speaking, the electron-donating groups, such as methoxy, methyl and ethyl, on the phenyl ring of boronic acids were beneficial for the transformation (entries 2–4), whereas electron-withdrawing groups like chloro, fluoro and trifluoromethyl, decreased the efficiency (entries 7–9). Yet, steric hindrance affected the efficiency. For example, 88% yield of compound **4** without ortho substituent was isolated, while the yield of compounds **7** and **8** derived from ortho-substituted boronic acid were decreased to 69% and 60%, respectively (entries 5–6).

Subsequently, the substrate scope was extended to 5-amino-1-[2-chloro-4-trifluoromethylphenyl]-1-H-pyrazole-3-carbonitrile (**1b**) and 5-amino-1-phenyl-3-methylpyrazole (**1c**). With the strong electron-withdrawing group of trifluoromethyl on the phenyl ring, substrate **1a** and **1b** reacted with aryl boronic acids smoothly to give the arylated products in good yields (entries 1–10). However, when CF<sub>3</sub> on the phenyl ring of substrate **1c** disappeared, as well as CH<sub>3</sub> instead of CN on the pyrazole ring, no more than 62% of arylated products can be obtained (entries 11–18). The relatively poor reactivity may largely attribute to absence

**Table 1**  
Optimization of the arylation conditions.<sup>a</sup>



Entry	Catalyst	Base	Iodide source	Solvent	Yield (%) <sup>b</sup>
1	PdCl <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	25
2	Pd(OAc) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	8
3	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	85 (50 <sup>c</sup> , 54 <sup>d</sup> )
4	Pd(PPh <sub>3</sub> ) <sub>4</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	52
5	Cu(OAc) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	90 <sup>d</sup>
6	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	CH <sub>3</sub> ONa	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	36
7	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	Cs <sub>2</sub> CO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	6
8	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	K <sub>2</sub> CO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	32
9	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	t-BuOK	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	30
10	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	Na <sub>2</sub> CO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	20
11	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	CH <sub>3</sub> COOK	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	19
12	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	I <sub>2</sub>	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	36
13	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	ICl	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	40
14 <sup>e</sup>	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	86
15 <sup>f</sup>	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH:H <sub>2</sub> O	75
16	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	C <sub>2</sub> H <sub>5</sub> OH	96 <sup>g</sup>
17	PdCl <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub>	NaHCO <sub>3</sub>	NIS	Dioxane	93 <sup>g</sup>

<sup>a</sup> Reaction conditions: **1a** (0.2 mmol), **2a** (0.4 mmol), [NIS, I<sub>2</sub> or ICl] (1 equiv), [Pd] or [Cu] (10 mol%), base (0.4 mmol), C<sub>2</sub>H<sub>5</sub>OH:H<sub>2</sub>O (3:1), N<sub>2</sub>, 80 °C for 18 h.

<sup>b</sup> Isolated yield.

<sup>c</sup> NaHCO<sub>3</sub> (0.3 mmol).

<sup>d</sup> Without N<sub>2</sub>.

<sup>e</sup> NIS (1.1 equiv).

<sup>f</sup> NIS (0.9 equiv).

<sup>g</sup> The yield of by-product 4-iodo-1-phenylpyrazole

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