

Molecular insights into nitrogenase FeMoco insertion – The role of His 274 and His 451 of MoFe protein α subunit

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Received 28 February 2007; received in revised form 21 March 2007; accepted 23 March 2007

Available online 19 April 2007

This paper is dedicated to the memory of Professor Ed Stiefel

Abstract

The final step of FeMo cofactor (FeMoco) assembly involves the insertion of FeMoco into its binding site in the molybdenum–iron (MoFe) protein of nitrogenase. Here we examine the role of His α 274 and His α 451 of *Azotobacter vinelandii* MoFe protein in this process. Our results from combined metal, activity, EPR, stability and insertion analyses show that mutations of His α 274 and/or His α 451, two of the histidines that belong to a so-called His triad, to small uncharged Ala specifically reduce the accumulation of FeMoco in MoFe protein. This observation indicates that the enrichment of histidines at the His triad is important for FeMoco insertion and that the His triad potentially serves as an intermediate docking point for FeMoco through transitory ligand coordination and/or electrostatic interaction. © 2007 Elsevier Inc. All rights reserved.

Keywords: Nitrogenase; Biosynthesis; FeMoco; MoFe protein

1. Introduction

Nitrogenase catalyzes the nucleotide-dependent reduction of the inert atmospheric dinitrogen to the bioavailable form of ammonia (for recent reviews see Refs. [1–5]). The molybdenum (Mo) nitrogenase of *Azotobacter vinelandii* is a two-component system, comprising the iron (Fe) protein and the molybdenum–iron (MoFe) protein. The Fe protein (also designated Av2) is a homodimer (encoded by *nifH*), with one ATP binding site per subunit and a single [4Fe–4S] cluster bridged between the subunits. The MoFe protein (also designated Av1) is an $\alpha_2\beta_2$ -tetramer (encoded by *nifD* and *nifK*), containing two unique metal clusters per $\alpha\beta$ subunit pair: the [8Fe–7S] P-cluster [6], which is bridged between α - and β -subunits and ligated to six Cys ligands; and the [Mo–7Fe–9S–X–homocitrate]¹ FeMo cofactor (FeMoco) [7], which is located within the α -subunit and

bound to the protein by Cys α 275 at the terminal Fe and His α 442 at the opposite Mo that is further coordinated by an endogenous homocitrate. Nitrogenase catalysis involves complex association/dissociation between the Fe protein and the MoFe protein, during which process electrons, concomitant with ATP hydrolysis by the Fe protein, are sequentially transferred from the [4Fe–4S] cluster in the Fe protein through the P-cluster to the FeMoco in the MoFe protein, where substrate reduction takes place.

The biosynthesis of nitrogenase component proteins and complex metal clusters are controlled by the nitrogen fixation (*nif*) genes [8]. The assembly of MoFe protein occurs in two steps. First, a P-cluster-replete, yet FeMoco-depleted form is synthesized in a process that involves the expression of *nifD* and *nifK*—the structural genes encoding the α - and β -subunits of the MoFe protein, and the formation of P-cluster—likely through fusion of its substructural, [4Fe–4S] cluster units at the target location (*in situ*) in the protein [8–10]. Then, the FeMoco—assembled outside of the MoFe protein (*ex situ*) in a process that involves the participation of *nifB*, *nifE*, *nifN*, *nifH*, *nifV* and *nifQ*, is

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¹ The identity of X is unknown but it is considered to be C, O or N [7].

inserted into its binding site, resulting in the formation of an active, holo form of MoFe protein [8]. The *ex situ* assembly of FeMoco is presumably initiated with the formation of an Fe/S core on NifB, followed by the relocation and rearrangement of the core on the NifEN complex [11–14]. Recently, we identified a Mo-free, NifEN-bound FeMoco precursor that closely resembled the core structure of the mature FeMoco [11,12]. We also showed that, coupled to the hydrolysis of ATP, the Fe protein inserted Mo and homocitrate into the precursor while it was still bound to NifEN, resulting in a fully complemented cluster that could be inserted into the MoFe protein through direct protein–protein interaction [13,14]. These recent developments on *ex situ* FeMoco biogenesis not only clarify certain aspects of the cluster assembly in nitrogenase, they may also provide useful insights into the biosynthetic strategy of complex metal clusters in other systems [15–17].

On the other hand, the mechanism of the final step of FeMoco assembly—namely, how FeMoco is inserted into its target binding site in the MoFe protein, remains largely unknown. Much of the current understanding in this regard comes from the crystallographic studies of a $\Delta nifB$ MoFe protein from a *nifB*-deletion strain of *A. vinelandii* [18]. In agreement with the genetically-based theory that *nifB* is the starting point of FeMoco biosynthesis, the $\Delta nifB$ MoFe protein is FeMoco-deficient, yet it contains normal P-clusters [18]. More importantly, the $\Delta nifB$ MoFe protein can be readily reconstituted, *in vitro*, into an active holo

protein by addition of isolated FeMoco, suggesting that it represents a physiologically relevant snapshot of MoFe protein right before the final insertion of FeMoco [18]. In contrast to the wild-type MoFe protein, the $\Delta nifB$ MoFe protein contains a positively charged funnel that could allow the entry and navigation of the negatively charged FeMoco toward its target location [18]. Such a FeMoco insertion funnel is created as a result of major structural rearrangement of the α III domain of the $\Delta nifB$ MoFe protein, which involves the significant repositioning of a number of residues. Some of these residues belong to a stretch of polypeptide ranging from $\alpha 353$ to $\alpha 364$ which, upon repositioning by distances up to 20 Å, forms a loop at the entrance of the funnel. Among them are the positively charged Arg $\alpha 359$ and Arg $\alpha 361$, as well as the His $\alpha 362$, which could serve as the first contact point for FeMoco during the insertion process. Moving into the funnel, the C_{α} of His $\alpha 442$ is relocated by a distance of approximately 5 Å and joins His $\alpha 274$ and His $\alpha 451$ to form a striking His triad (Fig. 1), which could help guide the negatively charged FeMoco to the destined location. The movement of His $\alpha 442$ also leads to a switch in position of this residue with Trp $\alpha 444$, which could help secure FeMoco at its final location. Additionally, Lys $\alpha 426$ is shifted by roughly 5 Å, which could serve as another anchoring point for FeMoco during insertion. These observations have led to the proposed mechanism of FeMoco insertion, which involves the initial docking of FeMoco at the entrance loop of the

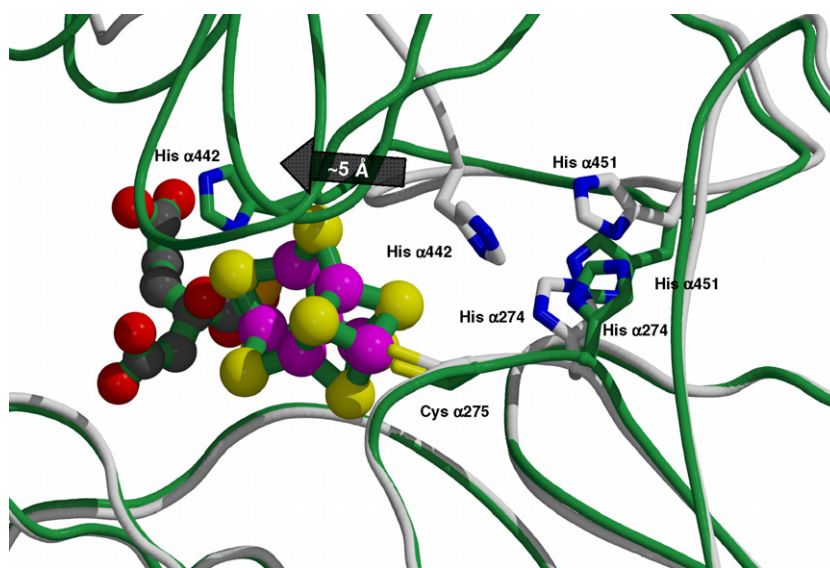


Fig. 1. Overlaid protein environment of Av1^{wild-type} and Av1 ^{$\Delta nifB$} in the vicinity of the FeMoco binding site. Parts of the C_{α} backbones of Av1^{wild-type} and Av1 ^{$\Delta nifB$} are shown and colored as follows: Av1^{wild-type} in green and Av1 ^{$\Delta nifB$} in light grey. FeMoco is shown in ball-and-stick presentation and colored as follows: oxygen in red, carbon in dark grey, molybdenum in orange, sulfur in yellow, iron in magenta. Note that, from this angle, only a small portion of the molybdenum atom is visible, while the central atom X (of unknown origin) is invisible. The side chains of His $\alpha 274$, His $\alpha 451$, His $\alpha 442$ and Cys $\alpha 275$ are shown in line presentation and colored as follows: His $\alpha 274$, His $\alpha 451$, His $\alpha 442$ of Av1^{wild-type} in green and blue, His $\alpha 274$, His $\alpha 451$, His $\alpha 442$ of Av1 ^{$\Delta nifB$} in light grey and blue, Cys $\alpha 275$ of Av1^{wild-type} in green and yellow, and Cys $\alpha 275$ of Av1 ^{$\Delta nifB$} in light grey and yellow. His $\alpha 442$ and Cys $\alpha 275$ are the two protein ligands that are covalently attached to FeMoco in Av1^{wild-type}. In Av1 ^{$\Delta nifB$} , the position of Cys $\alpha 275$ remains largely unchanged, whereas His $\alpha 442$ is re-positioned by a distance of approximately 5 Å and joins His $\alpha 451$ and His $\alpha 274$ in the formation of a striking His triad, which presumably serves to guide the FeMoco to its final location. The distances between the His triad residues, measured by the distances between their respective C2 atoms of the imidazole rings (the carbon atom between the two ring nitrogen atoms) are: (i) 3.62 Å between His $\alpha 451$ and His $\alpha 274$; (ii) 3.60 Å between His $\alpha 274$ and His $\alpha 442$; and (iii) 3.66 Å between His $\alpha 442$ and His $\alpha 451$. Additionally, His $\alpha 442$ can potentially form a hydrogen bond with the nearby Asp $\alpha 234$.

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