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Synthesis and characterization of new chiral palladium β-diimine complexes

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Dedicated to Professor Miha Tiŝler, Professor Emeritus of the University of Ljubljana, on the occasion of his 80th birthday.

Abstract

The synthesis and characterization of a range of chiral β -diimine ligands and their complexes with palladium(II) has been investigated. The introduction of chirality can be easily achieved through a combination of both achiral and chiral building blocks. The absolute configuration of the stereochemical centers has been determined. In addition, representative X-ray structures of both ligands and complexes have been determined.

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1. Introduction

Nickel and palladium complexes containing sterically demanding α -diimine (diazabutadiene) ligands are highly efficient catalysts for olefin polymerizations [1–3], alkyne cyclotrimerizations [4] and, as shown recently, also for Suzuki cross-coupling reactions [5]. On the other hand, few investigations have focused on the chemistry of analogous nickel and palladium β -diimine complexes. Feldman et al. reported the synthesis of a sterically hindered β -diimine ligand bearing no substituents at the C $_{\beta}$ atom and examined its reactions with Ni(II) and Pd(II) catalyst precursors [6]. However, β -diimines lacking substituents at the central carbon atom typically form hydrogen-bridged β -iminoamine tautomers. Accordingly, in the presence of base, e.g., under catalytic conditions, deprotonation occurs giving rise to the formation of β -diketiminate complexes [7] and other products [8] rather than β -diimine complexes. Recently, Woods and co-workers synthesized several β -diimine ligands in which the problematic CH acidity was circumvented by diimine dialkylation [9]. In a subsequent paper, these authors described the synthesis of some palladium β -diimine complexes and provided structural comparisons with the corresponding α -diimine analogues [10]. We recently reported the synthesis and reactivity of a series of sterically demanding non-chiral β -diimine ligands as well as their Ni(II) and Pd(II) complexes, where the central carbon atoms of the ligands were part of five- and six-membered rings in order to ensure that the ligands kept their diimine character even under basic conditions [11].

In the present contribution we extend our synthetic efforts to obtain novel chiral β -diimine ligands, where chirality can be introduced into two different positions of the ligands by using a combination of chiral and achiral building blocks. Condensation of an achiral dialdehyde with chiral amines renders the ligands chiral at the diimine moieties, whereas a chiral dialdehyde and an achiral amine

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leaves the ligand backbone chiral. A combination of these possibilities finally affords ligands which are chiral both at the ligand backbone and at the diimine functionality. Moreover, the synthesis of a series of Pd(II) complexes with our new ligands is described. Representative X-ray structures of ligands and palladium complexes are reported.

2. Results and discussion

2.1. Synthesis of the ligands

Following the methodology established previously [11] the β -diimine ligands were synthesized by condensation of 1,1-cyclopentanedicarbaldehyde and various amines. By using chiral amines diimines bearing chirality at the diimine moiety were obtained (Scheme 1).

To introduce chirality into the cyclic framework, the synthesis of a chiral dialdehyde was necessary (Scheme 2). After asymmetric reduction of 1,4-diphenylbutane-1,4-dione using commercially available (–)-DIP-chloride the resulting diol (S,S)-2 was converted into the dimesylate (S,S)-3 using an optimized procedure [12]. The cyclization step was carried out by alkylation with malonic acid diethylester and sodium hydride in the presence of 15-Crown-5 to yield the enantiomerically pure diethyl malonate (R,R)-4. Traces of (meso)-diester were easily removed by column chromatography.

Reduction with lithium aluminum hydride to the diol (R,R)-5 followed by a Swern oxidation gave the dialdehyde (R,R)-6 in an excellent yield. The enantiomeric excess was estimated by synthesis of the racemic diol (rac)-5 and (meso)-5 and HPLC as well as by ¹⁹F NMR analysis of their Mosher esters (see Supplementary material,





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