

Synthesis of polyacetylenic montiporic acids by means of organosilicon compounds

Vito Fiandanese *, Daniela Bottalico, Giuseppe Marchese, Angela Punzi

Dipartimento di Chimica, Università di Bari, via E. Orabona 4, 70126 Bari, Italy

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Abstract

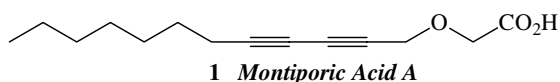
A straightforward synthesis of polyacetylenic montiporic acids A and B has been developed, based upon the selective and sequential substitution of the two trimethylsilyl groups of the readily available 1,4-bis(trimethylsilyl)-1,3-butadiyne.

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1. Introduction

In recent years, stony corals have been objects of investigation by organic chemists as sources of interesting bioactive natural products. In particular, the stony coral *Montipora* sp. is especially rich in acetylenic compounds that have been shown to possess antifungal, antibacterial, ichthyotoxic, and cytotoxic properties [1]. Recently, two new diacetylenic carboxylic acids, montiporic acids A **1** and B **2**, have been isolated [1a,1c] from the eggs of the scleractinian hermaphroditic coral *Montipora digitata* and exhibited interesting biological activities, such as antimicrobial activity against *Escherichia coli* and cytotoxicity against P-388 murine leukemia cells.



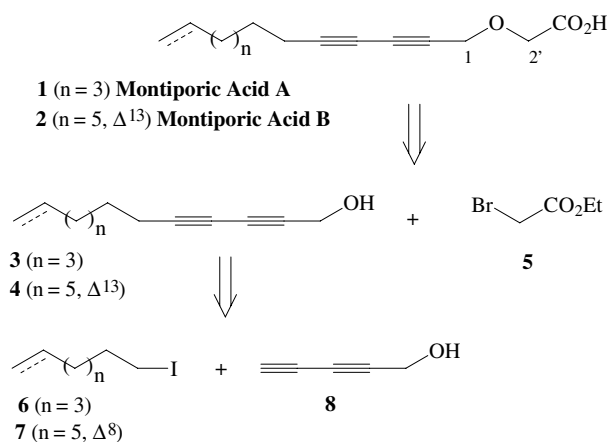
So far, in spite of the interesting properties of these compounds, only one report [2] has described the total synthesis of montiporic acids A and B, employing two different synthetic strategies for the two products.

In connection with our previous studies dealing with the synthesis of stereodefined conjugated polyunsaturated systems [3] and of a series of natural compounds [4], we have recently reported [5] a straightforward and general route to a variety of unsymmetrically substituted conjugated diynes, based upon the selective and sequential substitution of the trimethylsilyl groups of the readily available 1,4-bis(trimethylsilyl)-1,3-butadiyne with alkyl, aryl and vinyl groups. Now, we wish to report a straightforward and expeditious procedure for the total synthesis of both montiporic acids A **1** and B **2** starting from a common intermediate and following the same reaction sequence.

2. Results and discussion

Our overall retrosynthesis is summarized in Scheme 1. Montiporic acids A and B differ in the aliphatic chain linked to the diyne moiety, a *n*-heptyl group for montiporic acid A (**1**, *n* = 3) and a 8-nonenyl group

* Corresponding author. Tel.: +390805442075; fax: +390805442129.
E-mail address: fianda@chimica.uniba.it (V. Fiandanese).



Scheme 1.

for montiporic acid B (**2**, $n = 5, \Delta^{13}$). Thus, the disconnection of the O–C₂' bond leads to the polyunsaturated alcohols **3**, **4** and to the bromoester **5**. A further disconnection of the C₅–C₆ bond leads to the appropriate halides **6**, **7** and to the key intermediate **8**.

Accordingly, the synthesis of both the montiporic acids **1** and **2** can be realized employing the same methodology, starting with a coupling reaction between the diynol **8** [6] and the readily available halides **6** or **7** which leads to the polyunsaturated alcohols **3** [1a,2,7] or **4** [1a,1d,2]. The subsequent functionalization with the halide **5** can directly give both the montiporic acids **1** and **2**.

The synthesis of the diynol **8** was performed as depicted in Scheme 2, in accordance with our recently published procedure for the synthesis of conjugated diynes [5], starting from the same intermediate, the commercially available 1,4-bis(trimethylsilyl)-1,3-butadiene **9**.

The diyne **9** was selectively desilylated with MeLi–LiBr complex affording the lithium salt of the mono-silylated terminal diyne which was coupled with paraformaldehyde to give the mono-silylated diynol **11** [6b,8] in 86% yield. A further desilylation reaction of **11** with K₂CO₃ in MeOH led to compound **8** in 80% yield.

Therefore, the synthesis of montiporic acids A and B was performed as depicted in Scheme 3.

Both montiporic acids A and B were synthesized starting from the same compound **8**. In the case of montiporic acid A, the coupling reaction of the lithium salt

of **8** with 1-iodoheptane **6** led to compound **3** in 62% yield.

The subsequent reaction of compound **3** with ethyl 2-bromoacetate **5** under phase transfer catalysis conditions [9] gave montiporic acid A in 80% yield.

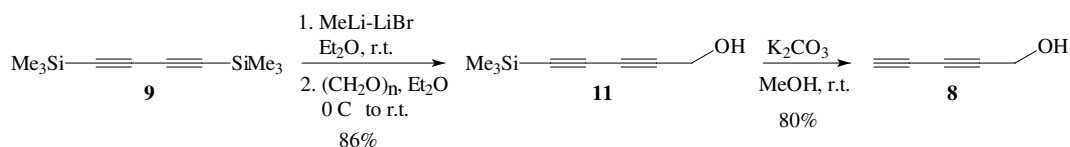
For the synthesis of montiporic acid B, that was obtained in 82% yield, the same reaction sequence was followed employing in the first step a different halide, 9-iodonon-1-ene **7**.

It is noteworthy that also the diacetylene alcohols **3** and **4** are marine metabolites isolated from the stony coral *Montipora* sp. and other species of hermatypic corals [1a,1d]. These compounds possess antifungal, antibacterial, ichthyotoxic properties and exhibited significant cytotoxicity against a small panel of human solid tumor cell lines.

In conclusion, the procedure described here appears to be a useful route to polyacetylenic montiporic acids. A special advantage of our strategy is represented by the possibility of synthesizing both the montiporic acids starting from a common intermediate and employing the same reaction sequence. Moreover, the simplicity of the operations involved, the mild reactions conditions and the ready availability of the silyl derivative employed are additional features making the procedure very promising.

3. Experimental section

THF and Et₂O were dried before use by distillation from Na/benzophenone ketyl under nitrogen. All other solvents were used as obtained. All reactions were carried out under nitrogen in dried glassware. Macherey–Nagel silica gel (60, particle size 0.040–0.063 mm) for column chromatography and Macherey–Nagel aluminum sheets with silica gel 60 F₂₅₄ for TLC were used. GC analysis were performed on a Varian 3900 gas chromatograph equipped with a J&W capillary column (DB-1301, 30 m × 0.25 mm i.d.). GC/mass-spectrometry analysis were performed on a Shimadzu GCMS-QP5000 gas chromatograph-mass spectrometer equipped with a Zebron capillary column (methyl polysiloxane, 30 m × 0.25 mm i.d.). ¹H NMR spectra were recorded in deuteriochloroform on a Bruker AM 500 spectrometer at 500 MHz. ¹³C NMR spectra were recorded in deuteriochloroform on a Bruker AM 500 spectrometer at 125.7 MHz. IR spectra were measured



Scheme 2.

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