

Syntheses, displacement behavior, hetero-atom carbene, and crystal structures of platinum complexes containing the *N,N*-dimethylthiocarbamoyl, $\text{Me}_2\text{NC}=\text{S}$, ligand: Structures of $[\text{Pt}(\text{PPh}_3)(\text{Cl})]_2(\mu, \eta^2\text{-SCNMe}_2)_2$, $[\text{Pt}(\text{PPh}_3)(\eta^2\text{-dppa})\{\eta^1\text{-C}(\text{S})\text{NMe}_2\}][\text{Cl}]$, and $[\text{Pt}(\text{PPh}_3)_2\{\eta^1\text{-C}(\text{SEt})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$

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Abstract

In solution state, the complex $[\text{Pt}(\text{PPh}_3)_2(\eta^1\text{-SCNMe}_2)(\text{Cl})]$ (**1**) shows intermolecular displacement of two triphenylphosphine ligands to form the bridging η^2 -thiocarbamoyl diplatinum complex $[\text{Pt}(\text{PPh}_3)\text{Cl}]_2(\mu, \eta^2\text{-SCNMe}_2)_2$ (**2**). Intramolecular displacement of the chloride product η^2 -thiocarbamoyl complex $[\text{Pt}(\text{PPh}_3)(\eta^2\text{-SCNMe}_2)][\text{Cl}]$ (**3**) was not been detected from the $^{31}\text{P}\{^1\text{H}\}$ NMR experiments. However, the chelating η^2 -thiocarbamoyl complex $[\text{Pt}(\text{PPh}_3)_2(\eta^2\text{-SCNMe}_2)][\text{PF}_6]$ (**4**) can be produced by the reaction of **1** with NH_4PF_6 in acetone at ambient temperature. Treatment of **1** with dppa {bis(diphenylphosphino)amine} in dichloromethane at room temperature results in the formation of the complex $[\text{Pt}(\text{PPh}_3)(\eta^2\text{-dppa})\{\eta^1\text{-C}(\text{S})\text{NMe}_2\}][\text{Cl}]$ (**5**). Treatment of **4** with EtOCS_2K or MeOCS_2K yields Fischer-type carbene-complex $[\text{Pt}(\text{PPh}_3)\{\eta^1\text{-C}(\text{SEt})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$ (**7**) or $[\text{Pt}(\text{PPh}_3)\{\eta^1\text{-C}(\text{SMe})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$ (**8**). The carbene-complexes **7** and **8** are formed via alkyl migration of the alkylidithiocarbonate ligand to the thiocarbamoyl ligand. Complex **1** reacts with EtOCS_2K , resulting in the formation of the η^2 -dithiocarbonate complex $[\text{Pt}(\text{PPh}_3)(\eta^1\text{-SCNMe}_2)(\eta^2\text{-S}_2\text{COEt})]$ (**6**) and the carbene-complex $[\text{Pt}(\text{PPh}_3)\{\eta^1\text{-C}(\text{SEt})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$ (**7**) with a ratio of 3:2 according to the integration of the $^{31}\text{P}\{^1\text{H}\}$ NMR spectra. By continuously stirring mixtures **6** and **7** in the CH_2Cl_2 solution for 4 h, complex **7** was formed as the final product. All of the complexes were identified by spectroscopic methods and complexes **2**, **5**, and **7** were determined by single-crystal X-ray diffraction.

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1. Introduction

Although thiocarbamoyl metal complexes are known in Nb, Ta (VB) [1], Mo, W (VIB) [2,3], Ru, Rh, and Ir (VIIIB) [4,5], the Pt thiocarbamoyl complex has received little attention. These complexes have been prepared from the

reaction of metal carbonylates with *N,N*-dimethylthiocarbamoyl chloride, $\text{Me}_2\text{NC}(=\text{S})\text{Cl}$, via nucleophilic displacement of chloride [6], nucleophilic attack by an amine on an electrophilic thiocarbonyl complex [7], electrophilic attack at a coordinated isothiocyanate [8], reaction of hydrosulfide with haloaminocarbene [9] or isonitrile [10] ligands, cleavage of a dithiocarbamate ligand [2], or C–H activation of thioformamides [11]. Oxidative addition [4] of both chloride and *N,N*-dimethylthiocarbamoyl ligand to the metal complex is most generally used and is relevant to the work described herein.

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Our previous report showed that the sulfur atom of the thiocarbamoyl ligand assists the displacement of either the chloride or the triphenylphosphine ligand to form the η^2 -thiocarbamoyl palladium complexes [12]. We report here the formation of two hetero-atom carbene complexes of the platinum via alkyl migration of the alkylthiocarbonate to thiocarbamoyl ligand, as well as the displacement behavior and reactions of $[\text{Pt}(\text{PPh}_3)_2(\eta^1\text{-SCNMe}_2)(\text{Cl})]$ (**1**) with neutral phosphorus ligand. Three X-ray crystal structure analyses have been carried out to provide structural parameters.

2. Results and discussion

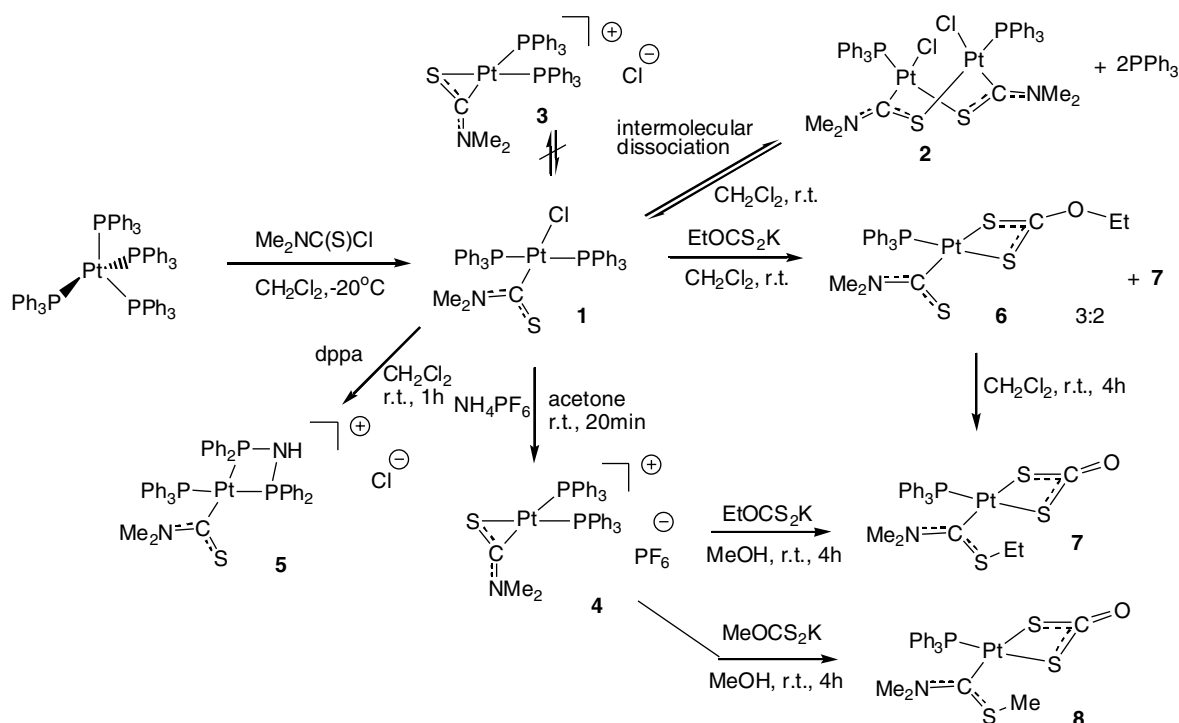
2.1. Syntheses

We have previously reported the reaction of $\text{Pt}(\text{PPh}_3)_4$ with *N,N*-dimethylthiocarbamoyl chloride, $\text{Me}_2\text{NC}(=\text{S})\text{Cl}$, in dichloromethane at -20°C , yielding the air-stable yellow complex $[\text{Pt}(\text{PPh}_3)_2(\eta^1\text{-SCNMe}_2)(\text{Cl})]$ (**1**) [13].

The dichloromethane solution of **1** undergoes intermolecular displacement of the triphenylphosphine ligand of **1** to form the bridging η^2 -thiocarbamoyl diplatinum complex $[\text{Pt}(\text{PPh}_3\text{Cl})_2(\mu, \eta^2\text{-SCNMe}_2)_2]$ (**2**) but the intramolecular displacement of the chloride ligand of **1** to form the chelating η^2 -thiocarbamoyl complex $[\text{Pt}(\text{PPh}_3)_2(\eta^2\text{-SCNMe}_2)]\text{Cl}$ (**3**) has not been found (Scheme 1). Continuous refluxing of the chloroform solution of these mixtures for 8 h produces complex **2** as the final product with 93% isolated yield (Fig. 1). Complex $[\text{Pt}(\text{PPh}_3)_2(\eta^2\text{-SCNMe}_2)]\text{PF}_6$ (**4**) can be synthesized from the reaction of **1** and ammonium hexafluorophosphate, NH_4PF_6 , in acetone with 92% isolated yield at ambient temperature [14]. The yellow-orange complex **4** is more stable and is less soluble than that of **1**. In the Pd analog, complex $[\text{Pd}(\text{PPh}_3)_2(\eta^1\text{-SCNMe}_2)(\text{Cl})]$, showing the inter- and intramolecular displacement behavior, was shown to form complexes $[\text{Pd}(\text{PPh}_3\text{Cl})_2(\mu, \eta^2\text{-SCNMe}_2)_2]$ and $[\text{Pd}(\text{PPh}_3)_2(\eta^2\text{-SCNMe}_2)]\text{Cl}$ [12] and the displacement of either the phosphine or the chloride ligand occurred in CH_2Cl_2 solution of complex $[\text{Pd}(\text{PPh}_3)_2(\eta^1\text{-CH}_2\text{SCH}_3)\text{Cl}]$ [15] to form monomer complexes $[\text{Pd}(\text{PPh}_3)(\eta^2\text{-CH}_2\text{SCH}_3)\text{Cl}]$ and $[\text{Pd}(\text{PPh}_3)_2(\eta^2\text{-CH}_2\text{SCH}_3)]\text{Cl}$. It is plausible that the sulfur atom of the thiocarbamoyl ligand assists triphenylphosphine displacement of **1** to form **2**.

In the reaction of **1** with *dppa* {bis(diphenylphosphino)amine} in dichloromethane at room temperature, a chloride and triphenylphosphine were displaced, forming complex $[\text{Pt}(\text{PPh}_3)(\eta^2\text{-dppa})\{\eta^1\text{-C}(\text{S})\text{NMe}_2\}]\text{Cl}$ (**5**) with 88% isolated yield.

Nucleophilic displacement of the chloride in **1** with anionic dithiocarbonate ligand, EtOCS_2^- , in dichloromethane at room temperature produces complex $[\text{Pt}(\text{PPh}_3)(\eta^1\text{-SCNMe}_2)(\eta^2\text{-S}_2\text{COEt})]$ (**6**) and carbene-complex $[\text{Pt}(\text{PPh}_3)\{\eta^1\text{-C}(\text{SEt})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$ (**7**) in good yields and $^{31}\text{P}\{^1\text{H}\}$ NMR spectrum indicates the 3:2 ratio of **6** and **7**. The dichloromethane solution of the mixtures **6** and **7** slowly undergo the ethyl-thiocarbamoyl coupling reaction of **6** to give **7** as the final product with 78% isolate yield over 4 h at room temperature. The same procedure was used to prepare the other carbene-complex $[\text{Pt}(\text{PPh}_3)\{\eta^1\text{-C}(\text{SMe})(\text{NMe}_2)\}(\eta^2\text{-S}_2\text{CO})]$ (**8**) with 85% isolated yield. The conversion of complex **6** to carbene complex **7** is proposed to



Scheme 1.

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