

Part 1: 1,3-Dipolar addition of activated alkyne towards coordinated azido group in ruthenium(II) complexes containing η^5 -cyclichydrocarbons

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Abstract

The indenyl and pentamethylcyclopentadienyl ruthenium(II) complexes $[(\eta^5\text{-L}_3)\text{Ru}(\text{L}_2)\text{Cl}]$ ($\text{L}_3 = \text{C}_9\text{H}_7$, $\text{L}_2 = \text{dppe}$ (**1a**), $\text{L}_2 = \text{dppm}$ (**1b**); $\text{L}_3 = \text{C}_5\text{Me}_5$, $\text{L}_2 = \text{dppe}$ (**2a**); $\text{L}_2 = \text{dppm}$ (**2b**) (where, $\text{dppe} = \text{Ph}_2\text{PCH}_2\text{CH}_2\text{PPh}_2$ and $\text{dppm} = \text{Ph}_2\text{PCH}_2\text{PPh}_2$) reacts with NaN_3 to yield the azido complexes $[(\eta^5\text{-C}_9\text{H}_7)\text{Ru}(\text{L}_2)\text{N}_3]$, $\text{L}_2 = \text{dppe}$ (**3a**), dppm (**3b**) and $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{L}_2)\text{N}_3]$, $\text{L}_2 = \text{dppe}$ (**4a**), dppm (**4b**), respectively. The azido complexes undergo [3 + 2] dipolar cycloaddition reaction with dimethylacetylenedicarboxylate to yield triazole complexes $[(\eta^5\text{-C}_9\text{H}_7)\text{Ru}(\text{L}_2)(\text{N}_3\text{C}_2(\text{CO}_2\text{Me})_2)]$, $\text{L}_2 = \text{dppe}$ (**5a**), dppm (**5b**) and $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{L}_2)(\text{N}_3\text{C}_2(\text{CO}_2\text{Me})_2)]$, $\text{L}_2 = \text{dppe}$ (**6a**), dppm (**6b**), respectively. The complexes were fully characterized on the basis of microanalyses, FT-IR and NMR spectroscopy. The crystal structures of the starting complex (**1a**) and representative complexes **5a**, **5b** and **6a** have been established by single X-ray study.

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1. Introduction

1,3-Dipolar cycloaddition is a common process in organic chemistry. The process involves the reaction between 1,3-dipoles having allyl and dipolarophiles type. Among various 1,3-dipoles, organic azides are particularly known to be important for the synthesizing heterocyclic compounds [1–6]. By analogy, Dori and Ziolo [7a] and Fruhauf [7b] have reported that coordinated azide in metal complexes can also undergo cycloaddition reaction. Thus azido complexes have been reported to react with nitriles [8–14] and isonitriles [9,15–17] to produce

metal–nitrogen and metal–carbon bonded tetrazoles, respectively. Interestingly it has been reported that tetrazole complexes can also be produced by the reaction of coordinated nitrile in metal complex with NaN_3 [18].

Literature surveys, reveals that most of the cycloaddition reactions of the azido complexes have been studied with platinum(II) [11c,16] and palladium(II) metals [14b,19] although a few have been known in compounds of other transition metals, viz., rhodium(I) iridium(I), cobalt(III) and tin(IV) [20]. However, cycloaddition reaction of coordinated azido complexes in half-sandwich systems has not been studied much except a report appeared in cyclopentadienyl ruthenium cases in recent years [21]. In contrast, to the best of our knowledge, such cycloaddition reactions of coordinated azido complexes in indenyl and pentamethylcyclopentadienyl systems have not been studied so far. It has been

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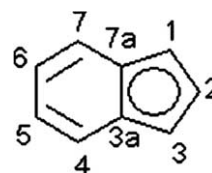
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reported that the chemistry of indenyl and pentamethylcyclopentadienyl ruthenium complexes differ from the analogous cyclopentadienyl ruthenium complex in certain aspects such as higher reactivity and labile nature of the *organic moieties*. The higher reactivity of the Cp* complexes is attributed to the steric and inductive effect whereas in the case of indenyl complexes ring slippage nature from η^5 - to η^3 - and back to η^5 - of the indenyl ligand is the solely responsible for the high reactivity [22]. It is noteworthy that reversible slippage of the metal across the five-membered ring can form a more reactive η^3 -indenyl tautomer that has both benzenoid resonance stabilization and accessible coordination site [23]. Our current study on the reactivity of the indenyl and Cp* ruthenium bis phosphine complexes, viz., $[\text{Cp}^*\text{Ru}(\text{PPh}_3)_2\text{Cl}]$ or $[(\text{ind})\text{Ru}(\text{PPh}_3)_2\text{Cl}]$ (where, ind = indenyl) towards N-base ligands reveals that the stability of the indenyl or Cp* ligands are largely depend on the steric nature of the incoming ligand, such that reaction of $[\text{Cp}^*\text{Ru}(\text{PPh}_3)_2(\text{CH}_3\text{CN})]^+$ or $[(\text{ind})\text{Ru}(\text{PPh}_3)_2(\text{CH}_3\text{CN})]^+$ with satirically demanded multidentate ligand, tetra-2-pyridyl-1,4-pyrazine (tppz), displaced the *organic moiety* and isolated the complex of the type $[\text{Ru}(\text{tppz})(\text{PPh}_3)_2(\text{CH}_3\text{CN})]^+$ [24]. However, the reaction with less steric N-base ligand, the *organic moiety* remain intact to the metal and thus forming complex of the type $[\eta^5\text{-L}_3]\text{Ru}(\text{PPh}_3)(\text{L}_2)]^+$ [25] (where L_3 = indenyl or Cp*). In general, sterically demanded multidentate ligand displaced them from the complexes. The part 1 of this communication, we described the syntheses of azido complexes of indenyl and pentamethylcyclopentadienyl ruthenium(II) and their reactions with dimethylacetylenedicarboxylate to generate triazole complexes (**5a,b** and **6a,b**). The spectral and structural characterization of the complexes has been discussed.

2. Experimental

All solvents were dried in appropriate drying agents and distilled prior to use [26]. $\text{RuCl}_3 \cdot 3\text{H}_2\text{O}$ (Arora Matthey Limited) and dimethylacetylenedicarboxylate

(Aldrich) were used as received. NMR spectra were recorded on an AMX-400 MHz spectrometer at 400.13 (^1H), 161.97 (^{31}P) or 100.61 MHz (^{13}C) with SiMe_4 or 85% H_3PO_4 as internal references and coupling constants are given in hertz. Infrared spectra were recorded as a KBr pellets on a Perkin–Elmer Model 983 spectrometer. Elemental analyses were carried out at the Regional Sophisticated Instrumentation centre (RSIC) Shillong, using a Perkin–Elmer 2400 CHN/S analyzer. Electronic spectra were recorded on a Hitachi-330 spectrophotometer in dichloromethane at ca. 10^{-4} M. The precursor complexes $[(\eta^5\text{-C}_9\text{H}_7)\text{Ru}(\text{dppe})\text{Cl}]$ (**1a**) and $[(\eta^5\text{-C}_9\text{H}_7)\text{Ru}(\text{dppm})\text{Cl}]$ (**1b**) were prepared following a literature method [27] while the complexes $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{dppe})\text{Cl}]$ (**2a**) and $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{dppm})\text{Cl}]$ (**2b**) were prepared by slight modification of the reported method [28] as described below. $^{13}\text{C}\{^1\text{H}\}$ NMR data for the indenyl complexes are presented in Table 1. The following atom labeling scheme is used for the ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR spectroscopic data of indenyl complexes.



2.1. Synthesis of precursor complex $[(\eta^5\text{-C}_5\text{Me}_5)\text{-Ru}(\text{dppe})\text{Cl}]$ (**2a**)

A two neck round bottom flask was charged with $[(\eta^5\text{-C}_5\text{Me}_5)\text{Ru}(\text{PPh}_3)_2\text{Cl}]$ (100 mg, 0.125 mmol), dppe (60 mg, 0.150 mmol) and 50 ml of toluene was added. The mixture was heated to reflux under nitrogen atmosphere for 15 h and cooled to room temperature. The solvent was removed under reduced pressure and the solid residue was purified by column chromatography over alumina. A bright yellow band was collected when eluted with dichloromethane which on concentration

Table 1
 $^{13}\text{C}\{^1\text{H}\}$ NMR data for the indenyl complexes^{a,b}

Complex	$\eta^5\text{-C}_9\text{H}_7$					Others	
	C-1,3	C-2	C-3a,7a	$\Delta\delta(\text{C-3a,7a})^b$	C-4,5	C-6,7	
3a	65.08	87.71	109.33	−21.37	124.63	125.78	27.86 (m, $\text{P}(\text{CH}_2)_2\text{P}$), 127.66–133.73 (m, PPh_2)
3b	63.09	87.14	109.07	−21.63	123.20	125.65	49.40 (t, $J_{\text{C-P}} = 19.7$, $\text{P}(\text{CH}_2)_2\text{P}$), 128.32–137.69 (m, PPh_2)
5a	66.35	93.60	108.81	−21.89	125.13	127.51	29.42 (m, $\text{P}(\text{CH}_2)_2\text{P}$), 51.21 (s, (OCH_3)), 127.66–133.73 (m, Ph), 138.12 ($\text{C}(\text{CO}_2\text{ Me})$), 167.8 (s, (CO_2))
5b	66.50	89.44	108.76	−21.94	124.98	125.18	49.66 (t, $J_{\text{C-P}} = 21.4$, $\text{P}(\text{CH}_2)_2\text{P}$), 51.11 (s, (OCH_3)), 127.59–134.61 (m, Ph), 138.04 ($\text{C}(\text{CO}_2\text{ Me})$), 168.72 (s, (CO_2))

^a Spectra recorded in CDCl_3 ; δ in ppm and J in Hz. Abbreviations: s, singlet; t, triplet; m, multiplet.

^b $\Delta\delta(\text{C-3a,7a}) = \delta(\text{C-3a,7a})(\eta^5\text{-indenyl complex}) - \delta(\text{C-3a,7a})(\text{sodium indenyl})$, $\delta(\text{C-3a,7a})$ for sodium indenyl, 130.70 ppm.

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