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# Mixed-ligand iminopyrrolato-salicylaldiminato group 4 metal complexes: Optimising catalyst structure for ethylene/propylene copolymerisations

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#### **Abstract**

Treatment of  $MCl_3(OC_6H_3-2-'Bu-6-CH=NC_6F_5)(THF)$  (M=Ti, Zr) with a variety of different potassium iminopyrrolate salts  $(K^+[RN=CHC_4H_3N]^-)$ , (R= phenyl, cyclo-hexyl, ethyl) afforded the corresponding titanium and zirconium mixed-ligand complexes  $MCl_2(N-O)(N-N)$ . The molecular structures of  $TiCl_2(OC_6H_3-2-'Bu-6-CH=NC_6F_5)(C_2H_5N=CHC_4H_3N)$  (Ic),  $TiCl_2(OC_6H_3-2-'Bu-6-CH=NC_6F_5)(C_6H_{11}N=CHC_4H_3N)$  (Ic),  $TiCl_2(OC_6H_3-2-'Bu-6-CH=NC_6F_5)(C_6H_{11}N=CHC_4H_3N)$  (Ic) show distorted octahedral geometries with trans- $O^-,N^-/cis$ - $Cl_2$  arrangements. On activation with MAO the titanium (iminopyrrolato)(salicylaldiminato) complexes show excellent activities in ethylene polymerisation and are significantly more effective ethylene/propylene copolymerisation catalysts, both in terms of activity and propene incorporation, than either of the parent complexes. The ethylene-propylene copolymers show ca. 80% 1,2 regioselectivity and at high propylene incorporation tend towards an alternating structure.

Keywords: Titanium; Zirconium; Ethene; Copolymerization; Salicylaldiminate; Iminopyrrolate

#### 1. Introduction

The development of new, more selective and active homogeneous polymerisation catalysts remains a driving force within organometallic chemistry. Of the non-metallocene group 4 metal complexes tested in olefin polymerisations, octahedral metal dichloride systems containing bidentate mono-anionic N,N<sup>-</sup> or N,O<sup>-</sup> ligands have proved to give particularly effective catalysts [1,2]. Bis (salicylaldiminato) complexes of titanium and zirconium of type A (Chart 1) are potent catalysts for ethylene homopolymerisations, but are inefficient copolymerisation catalyst [3–6]. They are sensitive to steric factors; for example, reducing the size of the *ortho*-substituent increases the

fraction of bulky comonomer incorporated, while the activity is significantly decreased [7]. On the other hand, bis(iminopyrrolato) titanium complexes (structure **B**) give lower ethylene polymerisation productivities but achieve higher incorporation of both cyclic and terminal alkenes in the polyethylene backbone [8,9].

These limitations can be overcome with hetero-ligated group 4 metal complexes [10–14] which have been shown to combine these characteristics and are able to produce highly active catalysts capable of high levels of comonomer incorporation. The most successful systems for  $\alpha$ -olefin co-polymerisations are the (iminopyrrolato)(salicylaldiminato)titanium complexes, which have previously been shown to combine high activity with high comonomer incorporation in copolymerisations of ethylene with 1-hexene, cyclopentene and norbornene [10]. We report here the syntheses and structures of two new titanium (N–N)(N–O)

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$$Ar = C_6F_5$$

$$Ar = C_6F_5$$

$$Ar = C_6H_5 (M = Ti)$$

$$1c R = Et (M = Ti)$$

$$2b R = cyclohexyl (M = Zr)$$

Chart 1.

mixed-ligand complexes and that of a zirconium analogue. When activated with methylaluminoxane (MAO) the titanium complexes are shown to be highly effective ethylene–propylene copolymerisation catalysts.

#### 2. Syntheses

The complexes  $TiCl_2(OC_6H_3-2-{}^tBu-6-CH=NC_6F_5)-(R-N=CHC_4H_3N)$  (1a, R=Ph; 1b, R=cyclohexyl; 1c, R=Et) as well as  $ZrCl_2(OC_6H_3-2-{}^tBu-6-CH=NC_6F_5)(Cy-N=CHC_4H_3N)$  (2b) were synthesised following previously published procedures [8,9] by reacting the mono(salicylal-diminato) complexes  $MCl_3(OC_6H_3-2-{}^tBu-6-CH=NC_6F_5)$  (THF) (M=Ti, Zr) in dichloromethane with potassium salts of the corresponding iminopyrroles (Scheme 1). The products were obtained as dark red to orange crystals.

The <sup>1</sup>H NMR and <sup>19</sup>F NMR spectra in CDCl<sub>3</sub> of the complexes were consistent with the presence of a single stereoisomer at room temperature, with the <sup>1</sup>H NMR spectrum showing only two sharp imine peaks, one each for the iminopyrrolato and the salicylaldiminato ligands. The

Scheme 1.

<sup>19</sup>F spectra of all the complexes indicate hindered rotation of C<sub>6</sub>F<sub>5</sub> groups, rendering the *ortho*- and *meta*-fluorines inequivalent.

The structures of **1b**, **1c** and **2b** were determined by X-ray crystallography and are shown in Figs. 1–3, respectively. Complexes **1c** and **2b** have essentially octahedral geometry, with the two anionic functions *trans* to one another while the two chloride ligands are *cis*. These structures are consistent with the most stable calculated geometry for such octahedral complexes bearing two mono-anionic ligands, and is also the geometry observed for most bis(salicylaldiminato), bis(iminopyrrolato) as well as for related mixed-ligand group 4 metal complexes [4,9–11].

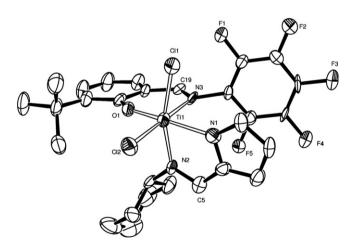


Fig. 1. ORTEP representation of the structure of 1b showing 50% probability ellipsoids. Hydrogen atoms have been omitted for clarity. Selected bond lengths (Å) and angles (°) with estimated standard deviations: Ti(1)-Cl(1) 2.282(3), Ti(1)-Cl(2) 2.282(2), Ti(1)-N(1) 2.072(6), Ti(1)-N(2) 2.119(6), Ti(1)-N(3) 2.242(6), Ti(1)-O(1) 1.820(5); Cl(1)-Ti(1)-Cl(2) 96.84(10), N(1)-Ti(1)-N(2) 76.4(2), N(2)-Ti(1)-N(3) 84.9(2), O(1)-Ti(1)-N(1) 162.7(2).

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