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Formation of new organometallic W/Cu/S clusters from reactions of [{(η⁵-C₅Me₅)WS₃}₃Cu₇(MeCN)₉](PF₆)₄ with donor ligands. Crystal structures and optical limiting properties of [(η⁵-C₅Me₅)WS₃Cu₃(Py)₆](PF₆)₂, [(η⁵-C₅Me₅)WS₃Cu₃Br-(PPh₃)₃](PF₆), and [(η⁵-C₅Me₅)WS₃Cu₄(Py)Cl(dppm)₂](PF₆)₂

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Abstract

Treatment of a preformed cluster $\{(\eta^5-C_5Me_5)WS_3\}_3Cu_7(MeCN)_9](PF_6)_4$ (1) in MeCN with excess pyridine afforded a tetranuclear cationic cluster $[(\eta^5-C_5Me_5)WS_3Cu_3(Py)_6](PF_6)_2$ (2). On the other hand, reactions of 1 with excess PPh₃ under the presence of LiBr gave rise to the other tetranuclear cationic cubane-type cluster $[(\eta^5-C_5Me_5)WS_3Cu_3Br(PPh_3)_3](PF_6)$ (3) while analogous reactions of 1 with dppm and LiCl followed by addition of excess pyridine generated an intriguing pentanuclear cationic cluster $[(\eta^5-C_5Me_5)WS_3Cu_4(Py)Cl(dppm)_2](PF_6)_2$ (4). Compounds 2–4 were fully characterized by spectroscopy and X-ray crystallography. The cluster dication of 2 adopts an incomplete WS₃Cu₃ cubane-like structure while the cluster cation of 3 contains a WS₃Cu₃Br cubane-like structure. The structure of the cluster dication of 4 consists of an unique WS₃Cu₄Cl framework in which an open cubane-like WS₃Cu₃Cl fragment and a Cu(dppm)₂ fragment are connected by one Cu–Cl bond and two Cu–dppm–Cu bridges. The optical limiting (OL) properties of the MeCN solutions of 1–4 were investigated with 7-ns laser pulses at 532 nm. © 2005 Elsevier B.V. All rights reserved.

Keywords: Tungsten cluster; Copper cluster; Sulfide cluster; Crystal structures; Optical limiting properties

1. Introduction

In the past decades, synthesis of Mo(W)/Cu/S clusters has attracted much interest due to their interesting

chemistry [1–11], and their potential applications in biological systems [1,4,10] and electro/photonic materials [7,8f,9f,9g,11]. There are several approaches to the construction of the Mo(W)/Cu/S clusters. The first one is the "one-pot" synthesis, in which thiometallates (e.g. $[MO_{4-n}S_n]^{2-}$ (M = Mo, W; n = 1-4) and $[(\eta^5-C_5Me_5)MS_3]^-$ (M = Mo, W)) were mixed with Cu(I) salts or complexes in organic solvents to generate various Mo(W)/Cu/S clusters, which could be found in

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numerous literatures [1–9,11]. The second one is that a preformed small Mo(W)/Cu/S cluster is used to react with certain donor ligands (e.g. S^{2-} , 4,4'-bipy), yielding larger clusters or polymeric clusters [3,5c,6b,9e–9g]. The third one, which is less explored, is involved in the formation of a smaller cluster from the reaction of a preformed cluster with donor ligands (e.g. 2,2'-bipy and PPh₃) [9c,9e,12]. For example, reaction of [NEt₄]₄[WS₄-Cu₅Cl₇] with 2,2'-bipyridine and PPh₃ afforded [WS₄Cu₃Cl(2,2'-bipy)]_n and [WS₄Cu₃(PPh₃)₃Cl], respectively [12a]. In the two reactions, the [WS₄Cu₅] framework was turned into a smaller incomplete WS₄Cu₃ cubane-shaped core.

On the other hand, we have been interested in the preparation and third-order non-linear optical properties of Mo(W)/Cu/S clusters over the last ten years [8b,8c,8e,8f,9]. We once communicated an interesting reaction in which $[{(\eta^2-C_5Me_5)WS_3}_3Cu_7(MeCN)_9]$ - $(PF_6)_4$ (1) reacted with 1,4-pyrazine (1,4-pyz) under the presence of LiCl, forming an interesting 2D polymer $\{[(\eta^5-C_5Me_5)WS_3Cu_3Cl(MeCN)(1,4-pyz)](PF_6)\}_{\infty}$ [9e]. In this reaction, the triple incomplete cubane-like [W₃S₉Cu₇] framework of 1 was turned into a smaller incomplete WS₃Cu₃ cubane-like framework. In addition, we are involved in the construction of Mo(W)/ Cu/S clusters from some preformed clusters with luminescent or optical limiting properties [9g,9f]. As discussed later in this paper, the solution of compound 1 exhibited slightly better optical limiting (OL) performance than that of C_{60} . Therefore, is it possible to use 1 as a useful precursor to make other new W/Cu/S clusters with better OL properties via its reactions with other donor ligands? With the question in mind, we carried out reactions of 1 with Py, PPh₃, and dppm, and some in the presence of LiX (X = Cl, Br). Three novel

smaller W/Cu/S clusters $[(\eta^5-C_5Me_5)WS_3Cu_3(Py)_6]-(PF_6)_2$ (2) $[(\eta^5-C_5Me_5)WS_3Cu_3Br(PPh_3)_3](PF_6)$ (3) and $[(\eta^5-C_5Me_5)WS_3Cu_4(Py)Cl(dppm)_2](PF_6)_2$ (4) were produced in relatively high yields. Furthermore, we have also examined the OL properties of 1–4 in acetonitrile with 7-ns laser pulses at 532 nm. Herein, we report their preparation and structural characterization along with their OL properties in solution.

2. Results and discussion

2.1. Synthesis and spectral characterization

As shown in Scheme 1, reaction of 1 in MeCN with excess Py formed a homogeneous solution, from which $[(\eta^5-C_5Me_5)WS_3Cu_3(Py)_6](PF_6)_2$ (2) was isolated as dark red plates in 65% yield. We once reported that lithium halide sometimes proved to be useful in making cluster-based supramolecular compounds [9e,9f]. For example, reactions of $[PPh_4][(\eta^5-C_5Me_5)WS_3Cu_3(CN)_3]$ with 1,4-pyrazine under the presence of LiCl yielded a supramolecular cube $[{(\eta^5-C_5Me_5)WS_3Cu_3}_8Cl_8(CN)_{12}-$ Li₄] [9f]. In the structure of this compound, each chloride interacts weakly with three copper atoms of the $(\eta^{2}-C_{5}Me_{5})WS_{3}Cu_{3}$ core, which may stabilize the resulting cluster framework. However, analogous reaction of 1 in MeCN with excess Py in the presence of LiCl (4 equiv.) gave rise to the same cluster 2 in 60% yield. When excess PPh₃ was added into the acentonitrile solution of 1, it formed a clear red solution, from which a red product was isolated after a workup. According to its elemental analysis, X-ray fluorescence analysis and IR spectrum, we tentatively assumed its chemical formula to be $[(\eta^5-C_5Me_5)WS_3Cu_3(PPh_3)_3](PF_6)_2$. Interest-



Scheme 1.

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