



# The effect of Ba-site substitution on the magnetic behavior of ordered perovskite $R\text{BaMn}_2\text{O}_6$ ( $R = \text{rare earth}$ )



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## ARTICLE INFO

### Article history:

Received 5 February 2015

Received in revised form

16 April 2015

Accepted 17 April 2015

Available online 25 April 2015

### Keywords:

Perovskite manganite

Chemical substitution

Structural randomness

Magnetic transitions

## ABSTRACT

We have investigated the effect of Ba-site substitution with  $Ae$  ( $Ae = \text{Sr}$  and  $\text{Ca}$ ) on the magnetic properties of  $A$ -site ordered  $R\text{BaMn}_2\text{O}_6$  with  $R = \text{Pr}$ ,  $\text{Nd}$ , and  $\text{Nd}_{0.5}\text{Sm}_{0.5}$ . Ba-site substitution reduces the mismatch between the average ionic radii of  $R$ - and Ba-sites, but introduces structural randomness into Ba-site. A reduction in the  $R/\text{Ba}$  mismatch suppresses the charge/orbital ordered insulating (COOI) state of  $\text{Nd}_{0.5}\text{Sm}_{0.5}\text{BaMn}_2\text{O}_6$  and the  $A$ -type antiferromagnetic (AAF) one of  $\text{PrBaMn}_2\text{O}_6$ , while the Curie temperature of  $\text{PrBaMn}_2\text{O}_6$  is almost insensitive to Ba-site substitution. At the multicritical point where the COOI, AAF, and ferromagnetic metallic (FM) states meet, the effect of the structural randomness is dominant; with increasing the structural randomness, the FM and AAF transition temperatures of  $\text{NdBaMn}_2\text{O}_6$  decrease.

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## 1. Introduction

Perovskite manganites  $R_{1-x}Ae_x\text{MnO}_3$  ( $R = \text{rare earth}$ ,  $Ae = \text{alkaline earth}$ ) have been attracting much interest because of their colossal magnetoresistance (CMR) and potential application in magnetic devices such as magnetic sensor [1,2]. Among many perovskite manganites,  $R\text{BaMn}_2\text{O}_6$  is one of the candidate materials for such magnetic devices, because both the ferromagnetic metallic (FM) and charge/orbital ordered insulating (COOI) phases, which are indispensable for the CMR effect, show up above room temperature [3–7]. The crystal structure and electronic phase diagram of  $R\text{BaMn}_2\text{O}_6$  are shown in Fig. 1.  $R\text{BaMn}_2\text{O}_6$  has the  $A$ -site ordered perovskite structure in which RO and BaO layers are alternately stacked along the  $c$ -axis.  $R/\text{Ba}$  order is caused by the large mismatch between the ionic radii of  $R^{3+}$  and  $\text{Ba}^{2+}$  ( $R/\text{Ba}$  mismatch).  $R\text{BaMn}_2\text{O}_6$  with  $R = \text{Sm–Dy}$ , and  $\text{Y}$  have the COOI ground state, and the COOI transition temperature  $T_{\text{CO}}$  reaches 500 K at  $R = \text{Y}$  [4]. On the other hand,  $R\text{BaMn}_2\text{O}_6$  with  $R = \text{Pr}$  and  $\text{Nd}$  exhibit successive magnetic transitions; the FM transition occurs above room temperature, and then the  $A$ -type antiferromagnetic (AAF) transition does below the Curie temperature  $T_{\text{C}}$ . As seen from the phase diagram, the COOI, FM, and AAF orders coincide with each other around room temperature to form a multicritical point at  $R = \text{Nd}$  [5–7].

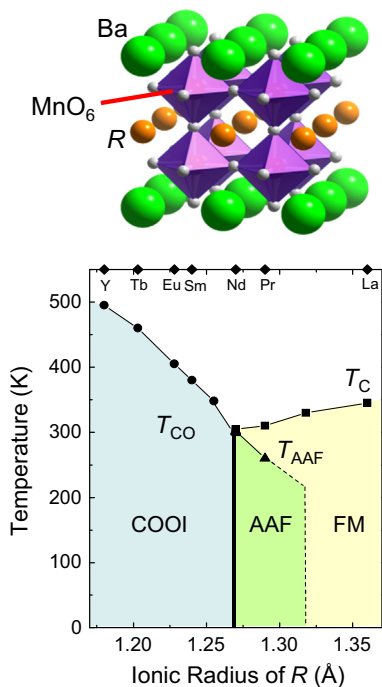
Around the multicritical region, chemical substitution of Ba-site with heterovalent  $R^{3+}$ , which corresponds to electron-doping, drastically changes the magnetic and transport properties of  $R\text{BaMn}_2\text{O}_6$  [8,9]. In the case of  $\text{NdBaMn}_2\text{O}_6$ , partial substitution of  $\text{Ba}^{2+}$  with  $\text{Nd}^{3+}$  creates FM domains in the AAF matrix, and as a result of 5%–substitution, the FM domains are dominant [9]. Nakajima et al. reported that  $R\text{Ba}_{1-x}\text{La}_x\text{Mn}_2\text{O}_6$  ( $R = \text{Sm}_{0.90}\text{La}_{0.10}$ ,  $x = 0.14$ ) exhibits the CMR effect at room temperature [8]. These results indicate that elaborate investigation of the effect of Ba-site substitution might provide crucial keys for realizing new magnetic devices using the CMR effect. To our knowledge, the effect of substitution of Ba-site with isovalent  $Ae$  ( $Ae = \text{Sr}^{2+}$  and  $\text{Ca}^{2+}$ ), which corresponds to band-width control, has not been systematically studied yet. In this study, we have prepared  $R\text{Ba}_{1-x}Ae_x\text{Mn}_2\text{O}_6$  ( $R = \text{Pr}$ ,  $\text{Nd}$ , and  $\text{Nd}_{0.5}\text{Sm}_{0.5}$ ,  $x = 0$ , 0.05, and 0.10), and investigated their magnetic properties.

## 2. Experimental

$R\text{Ba}_{1-x}Ae_x\text{Mn}_2\text{O}_6$  in polycrystalline form were prepared by a solid state reaction. We employed  $\text{Pr}_6\text{O}_{11}$ ,  $\text{Nd}_2\text{O}_3$ ,  $\text{Sm}_2\text{O}_3$ ,  $\text{BaCO}_3$ ,  $\text{CaCO}_3$ ,  $\text{SrCO}_3$ , and  $\text{Mn}_3\text{O}_4$  as starting materials. Mixed powders with appropriate molar ratio were pressed into the pellets, and sintered at 1523 K under Ar atmosphere. Then, we annealed these sintered samples under  $\text{O}_2$  flow to obtain oxygen-stoichiometric samples. We could not obtain single phase samples of  $R\text{Ba}_{1-x}\text{Sr}_x\text{Mn}_2\text{O}_6$  with  $x > 0.1$  and  $R\text{Ba}_{1-x}\text{Ca}_x\text{Mn}_2\text{O}_6$  with  $x > 0.05$ . This might be due to site-mixing between  $R^{3+}$  and dopant  $Ae^{2+}$ . Magnetic properties were studied using a Quantum Design

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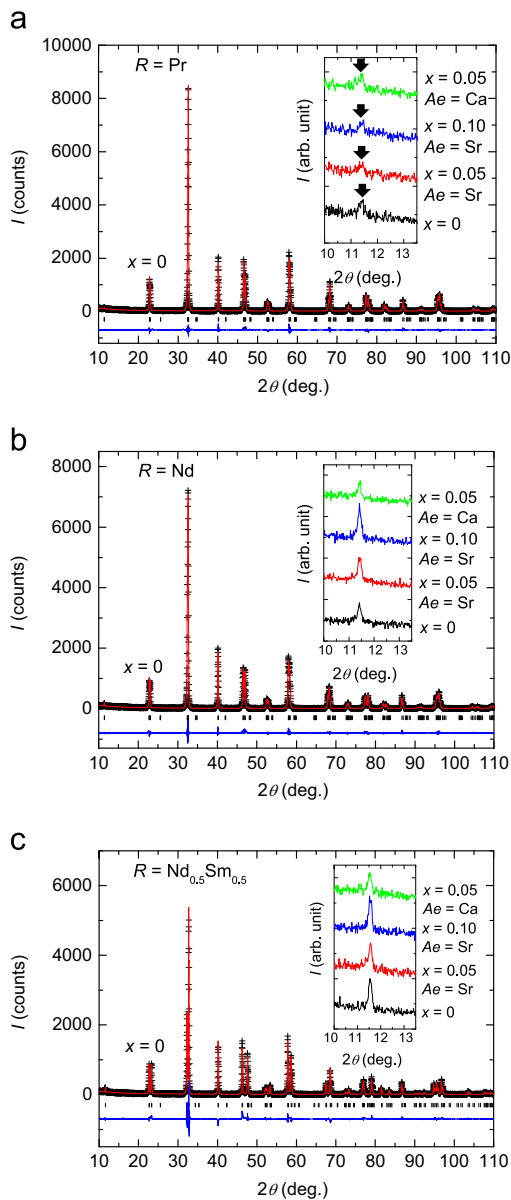
**Fig. 1.** Crystal structure and electronic phase diagram of  $\text{RBaMn}_2\text{O}_6$ .  $T_{\text{CO}}$ ,  $T_{\text{C}}$ , and  $T_{\text{AAF}}$  denote charge/orbital ordered insulating (COOI), ferromagnetic metallic (FM), and A-type antiferromagnetic (AAF) phase transition temperatures, respectively.

magnetic property measurement system (MPMS). Sample characterization was performed by powder X-ray diffraction (XRD) methods using  $\text{CuK}\alpha$  radiation at room temperature.

### 3. Results and discussion

**Fig. 2** shows the XRD patterns of  $\text{RBa}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  with (a)  $R = \text{Pr}$ , (b)  $\text{Nd}$ , and (c)  $\text{Nd}_{0.5}\text{Sm}_{0.5}$ . Structural analysis indicates that all the samples prepared in this study have A-site ordered perovskite structure and are free of impurity phases [10]. The (001) reflection around  $2\theta = 11.5^\circ$ , which arises from the alternate stack of RO and BaO layers, is a piece of evidence for A-site cation order, and this reflection is clearly observed in the XRD patterns of all the samples (the insets of **Fig. 2**). Both  $\text{PrBa}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  and  $\text{NdBa}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  have the orthorhombic structure with a space group of  $Pmmm$  and a unit cell of  $a_p \times a_p \times 2a_p$  at room temperature. Here  $a_p$  denotes a lattice parameter in a simple cubic perovskite setting. On the other hand,  $\text{Nd}_{0.5}\text{Sm}_{0.5}\text{Ba}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$ , which has the COOI state at room temperature, crystallizes in the pseudo-tetragonal structure with a space group of  $P4/mmm$  and a unit cell of  $a_p \times a_p \times 2a_p$ . Morikawa et al. recently reported that the COOI  $\text{RBaMn}_2\text{O}_6$  ( $R = \text{Sm}$ ) has the orthorhombic structure with a space group of  $Pnam$  ( $2\sqrt{2}a_p \times \sqrt{2}a_p \times 4a_p$ ) at room temperature [11]. However, in this study, for simplicity, the XRD profiles of  $\text{Nd}_{0.5}\text{Sm}_{0.5}\text{Ba}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  are indexed as the pseudo-tetragonal structure. The refined lattice parameters are listed in **Table 1**.

The orthorhombic or tetragonal lattice distortion of  $\text{RBaMn}_2\text{O}_6$  is caused by the large  $R/\text{Ba}$  mismatch. Thus, substitution of  $\text{Ba}^{2+}$  with a smaller cation such as  $\text{Ca}^{2+}$  and  $\text{Sr}^{2+}$  is expected to relax the lattice distortion caused by the  $R/\text{Ba}$  mismatch (the ionic radii of  $\text{Ba}^{2+}$ ,  $\text{Sr}^{2+}$ ,  $\text{Ca}^{2+}$ , and  $R^{3+}$  are 1.61 Å, 1.44 Å, 1.34 Å, and 1.18–1.36 Å, respectively). As expected, with increasing Sr-doping level, the lattice distortion, defined as  $(a+b)/c$  in this study, is being relaxed, that is,  $(a+b)/c$  is approaching unity (**Table 1**). Ca-substitution also relaxes the lattice distortion as in the case of Sr-substitution.



**Fig. 2.** Rietveld refined X-ray diffraction profiles of  $\text{RBa}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  ( $\text{Ae} = \text{Sr}$  and  $\text{Ca}$ ) with (a)  $R = \text{Pr}$ , (b)  $\text{Nd}$ , (c)  $\text{Nd}_{0.5}\text{Sm}_{0.5}$  at room temperature.

**Fig. 3(a)** and (b) shows the temperature ( $T$ ) dependence of the zero-field-cooled (ZFC) magnetization ( $M$ ) for  $\text{PrBa}_{1-x}\text{Sr}_x\text{Mn}_2\text{O}_6$  (Pr–Sr) and  $\text{PrBa}_{1-x}\text{Ca}_x\text{Mn}_2\text{O}_6$  (Pr–Ca) at  $H = 1$  kOe. As mentioned above,  $\text{PrBaMn}_2\text{O}_6$  exhibits the successive magnetic transitions; the  $M$  shows an abrupt increase at  $T_{\text{C}} = 305$  K, and then suddenly drops at the AAF transition temperature  $T_{\text{AAF}} = 260$  K with a thermal hysteresis. In this study, the  $T_{\text{C}}$ 's of  $\text{RBa}_{1-x}\text{Ae}_x\text{Mn}_2\text{O}_6$  ( $R = \text{Pr}$  and  $\text{Nd}$ ) are determined from the Curie–Weiss fitting. With increasing Sr-concentration, the  $T_{\text{AAF}}$  is monotonously decreasing from 260 K ( $x = 0$ ) to 254 K ( $x = 0.05$ ) then to 246 K ( $x = 0.10$ ). In the case of Pr–Ca with  $x = 0.05$  (**Fig. 3(b)**), at first glance, the  $T_{\text{AAF}}$  seems to be same as that of Pr–Sr with  $x = 0.10$ . However, as seen from the derivative of the  $M$  ( $dM/dT$ ) (the inset of **Fig. 3(b)**), the abrupt change in the  $M$  of Pr–Ca with  $x = 0.05$  starts at  $T_{\text{AAF}} = 250$  K, which is higher than that of Pr–Sr with  $x = 0.10$  ( $T_{\text{AAF}} = 246$  K) but lower than that of Pr–Sr with  $x = 0.05$  ( $T_{\text{AAF}} = 254$  K). On the other hand, the  $T_{\text{C}}$  is not so susceptible to Ba-site substitution. The  $T_{\text{C}}$ 's of Pr–Sr with  $x = 0.05, 0.10$ , and Pr–Ca with  $x = 0.05$  are 305 K, 304 K, and 303 K, respectively. These values almost coincide with that of the

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